

1 **Title: Halide homogenization for low energy loss in 2-eV-bandgap perovskites and increased**
2 **efficiency in all-perovskite triple-junction solar cells**

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30

31 **Abstract**

32 Monolithic all-perovskite triple-junction solar cells have the potential to deliver power conversion
33 efficiencies beyond those of state-of-art double-junction tandems and well beyond the detailed-
34 balance limit for single-junctions. Today, however, their performance is limited by large deficits in
35 open-circuit voltage and unfulfilled potential in both short-circuit current density and fill factor in the

36 wide-bandgap perovskite sub-cell. Here we find that halide heterogeneity – present even immediately
37 following materials synthesis – plays, a key role in interfacial non-radiative recombination and
38 collection efficiency losses under prolonged illumination, for Br-rich perovskites. We find that a
39 diammonium halide salt, propane-1,3-diammonium iodide, introduced during film fabrication,
40 improves halide homogenization in Br-rich perovskites, leading to significantly enhanced operating
41 stability and a record open-circuit voltage of 1.44 V in an inverted (p-i-n) device; ~86% of the
42 detailed-balance limit for a bandgap of 1.97 eV. The efficient wide-bandgap sub-cell enables the
43 fabrication of monolithic all-perovskite triple-junction solar cells with an open-circuit voltage of 3.33
44 V and a champion PCE of 25.1% (NREL-certified 23.87%).

45 **Main Text:**

46 Integrating multiple absorber layers into a multijunction solar cell offers the ability to overcome the
47 limitation of carrier thermalization from high-energy photons and transmission of low-energy
48 photons observed in single-junction photovoltaics (PVs)¹. Recent advances in metal halide perovskite
49 semiconductors with bandgap energies from 1.2 to 1.8 eV have enabled their use in tandem solar cells
50 with perovskite, crystalline silicon (c-Si), copper indium gallium selenide (CIGS), and organic
51 photovoltaics (OPV)²⁻⁶. Among these, all-perovskite double-junction (2J) tandem solar cells coupling
52 1.77/1.22 eV absorbers have achieved a steady-state certified power conversion efficiency (PCE) of
53 29%⁷; higher than 26.1% for single-junction perovskite solar cells (PSCs)⁸, indicative of the promise
54 of perovskite PVs.

55 Pairing three perovskite junctions in a monolithic solar cell has the potential to lead to higher
56 efficiencies compared to their 2J counterparts⁹. According to optical and device simulations, all-
57 perovskite triple-junction (3J) solar cells featuring cascaded 2.0, 1.6, and 1.2 eV absorbers have a
58 theoretical efficiency limit of >50% and could target PCE exceeding 36%^{10,11}. However, only a
59 limited number of studies have demonstrated 3J PSCs experimentally, and their realized performance
60 remains lower than that of their single-junction and 2J counterparts¹²⁻¹⁵.

61 Compared to their narrower-bandgap counterparts, inverted (p-i-n) single-junction PSCs of
62 approximately 2 eV bandgap, typically produced by >60% Br contents, suffer from a large open-
63 circuit voltage (V_{oc}) deficit of >700 mV (Fig. 1a,b) and performance loss as a consequence of light-
64 induced halide segregation¹⁶⁻¹⁸. Recent studies have found that the dominant performance-limiting
65 factors are increased band offset with charge transport layers and defect density in the wide-bandgap
66 perovskite absorbers^{19,20}. The latter is particularly evident for perovskites based on a high Br/I ratio,
67 as they are susceptible to bulk and interfacial halide heterogeneity due to the uncontrolled growth of
68 various halide species^{21,22}. In addition to increased trap-mediated energy losses, defect sites can
69 profoundly impact the charge-carrier dynamics of wide-bandgap perovskites when light-induced
70 halide segregation occurs²³⁻²⁸.

71 We sought to improve bulk and interfacial halide homogeneity in approximately 2 eV mixed
72 Br/I perovskites in order to reduce energetic losses and better realize the potential of 3J PSCs.
73 Previously, vacuum deposition has been utilized to produce uniform mixed-halide perovskites due to
74 its superior compositional control²⁹. For solution processing, the introduction of a Lewis acid–base
75 adduct additive has been shown to regulate the crystallization of perovskites^{30,31}; however, few
76 studies have explored the impact of additives on the homogeneity of mixed Br/I perovskites^{21,22}. In
77 pursuing a pathway towards the phase homogenization of high-Br perovskite films with reduced
78 defect densities, we turned to diammonium halide salts, which have been reported to have both a
79 templating effect on perovskite growth as well as a capability to passivate surface defects^{2,32-34}. We

80 show that a dual bulk and interface passivation technique with diammonium halide salts results in
81 significantly suppressed non-radiative recombination within the perovskite bulk and at
82 perovskite/charge-transporting-layer heterojunctions, enabling a record V_{oc} in a p-i-n device, ~86%
83 of the detailed-balance (DB) limit at 1.97 eV. Interestingly, we observe that while the diammonium
84 salt does not suppress light-induced halide segregation, wide-bandgap PSCs retain high operational
85 stability (>28 hours) under prolonged illumination, beyond the stability of previous approximately 2
86 eV PSCs reported in literature^{13,15,17}. We find that the additive promotes halide homogeneity near the
87 buried perovskite/hole-transport-layer (HTL) interface and reduces trap-mediated recombination,
88 collectively providing a pathway for efficient charge-carrier extraction from the halide-segregated
89 regions. Encouraged by the stable and efficient wide-gap perovskite sub-cell, we fabricated all-
90 perovskite 3J solar cells combining 1.97, 1.61, and 1.25 eV absorber layers. Using a front transparent
91 conductive oxide (TCO) layer with high charge mobility and high near-infrared (NIR) transmittance,
92 we demonstrate 3J devices with a V_{oc} of 3.33 V and a PCE of 25.1% (23.87% certified).

93

94 **Wide-bandgap PSC performance and energy loss analysis**

95 We developed a mixed-halide $CS_{0.15}FA_{0.85}Pb(I_{0.4}Br_{0.6})_3$ perovskite composition alloyed using 5%
96 $MAPbCl_3$ (FA = formamidinium, MA = methylammonium) to obtain a bandgap of 1.97 eV with
97 desired surface morphology for multijunction applications (see Methods, Supplementary Figs. 1–
98 3)^{35,36}. To study energetic losses of such wide-bandgap perovskites in a p-i-n configuration, we first
99 measured the quasi-Fermi level splitting (QFLS) by absolute photoluminescence (PL) spectroscopy
100 of a series of perovskite films and half-stacks using $NiO_x/[4-(3,6\text{-dimethyl-9}H\text{-carbazol-9-yl)butyl}]phosphonic\ acid$ (Me-4PACz)^{37,38} as the HTL and [6,6]-phenyl- C_{61} -butyric acid methyl ester
101 (PCBM) as electron transport layer (ETL) (Fig. 1a). Compared to the pristine wide-bandgap
102 perovskite film deposited on glass (neat film), a 55 meV lower QFLS is measured for perovskites
103 deposited on glass/indium tin oxide (ITO)/HTL substrates (w HTL), whereas a 114 meV QFLS loss
104 is obtained for PCBM deposited atop the glass/perovskites (w ETL, Fig. 1c). This suggests that the
105 perovskite/ETL interface is limiting the V_{oc} of wide-bandgap PSCs^{37,39}. Similar to previous studies
106 wherein diammonium cations have been shown to alter the surface potential of perovskites^{2,33,40}, we
107 found that post-treatment of the perovskite with a solution of propane-1,3-diammonium iodide (PDA)
108 salt effectively reduces the QFLS loss at the ETL interface to 45 meV, consistent with the V_{oc}
109 improvement of full devices (Fig. 1d).

111 Interestingly, we found that the QFLS loss at the HTL (55 meV) is eliminated entirely upon
112 addition of 1 mol% PDA into the perovskite precursor. This provides a lossless HTL interface as has
113 previously been reported for perovskites of narrower bandgaps^{2,37,41}. The top ETL interface shows a
114 significantly reduced QFLS loss of 68 and 26 meV before and after surface treatment with PDA,

115 respectively. Leveraging these insights, we fabricated bulk and surface-passivated wide-bandgap
116 PSCs using PDA, with a p-i-n structure of ITO/NiO_x/Me-4PACz/perovskite/PCBM/bathocuproine
117 (BCP)/Ag. The PSCs achieved a V_{oc} of 1.44 V, reaching ~86% of the DB limit (Fig. 1b) at a 1.97 eV
118 bandgap, as determined from the external quantum efficiency (EQE) spectrum (Supplementary Fig.
119 4); the highest reported for such solar cells (Fig. 1b)¹⁶. Combining a high fill factor (FF) of 0.83 and
120 short-circuit current density (J_{sc}) of 12.8 mA cm⁻², the champion device has a PCE of up to 15.3%
121 with negligible hysteresis and stable steady-state power output (Fig. 1d and Supplementary Fig. 5).

122 In addition to the performance enhancement, the PDA-modified wide-bandgap PSCs exhibit
123 significantly improved operating stability. Under maximum point power point (MPP) tracking at 1
124 sun illumination, the control device (passivated by PDA only at the surface) dropped to below 90%
125 of its initial performance after 0.5 h and ceased to be operational within 5 h, whereas the PCE of the
126 PDA-modified device retained over 90% after 20 h (Fig. 1f). To our knowledge, this is the highest
127 reported operating stability among mixed Br/I PSCs with a bandgap of approximately 2 eV, including
128 their all-inorganic analogs^{13,15,16}. We note that these hybrid perovskites also display superior ambient
129 (ISOS D-1, Supplementary Fig. 6) and thermal stability (65 °C in N₂, Supplementary Fig. 7)¹⁷. For
130 comparison, we also fabricated solar cells adding 1 mol% n-propylammonium iodide (PA), the
131 monoammonium counterpart of PDA, in the perovskite and found an insignificant change in device
132 stability. This suggests a synergistic effect of the two -NH₃⁺ groups of PDA molecules in improving
133 the performance of the wide-bandgap perovskites.

134

135 **Photoinduced changes in wide-bandgap PSCs**

136 To quantify the impact of halide segregation on the performance metrics under operating working
137 conditions, we collected PL spectra in situ during current density-voltage ($J-V$) measurements on
138 wide-bandgap PSCs (see Methods and Supplementary Figs. 8–9)²⁷.

139 In all cases, the initial PL peak attributed to the mixed-halide phase (~1.9 eV) diminished after
140 ~10³ s of illumination and was followed by a rise of a lower energy peak at ~1.6 eV (Figs. 2a–f). This
141 has been attributed to the formation of lower bandgap I-rich perovskite domains into which charge
142 carriers funnel and recombine radiatively⁴². Coincidentally, the J_{sc} and FF of control and PA-added
143 devices also diminish under continuous operation, leading to a severe loss of their initial PCEs
144 between ~10³ and 10⁴ s (Fig. 2g and Supplementary Fig. 10). We consider that funneled charge
145 carriers accumulate and recombine rapidly in the I-rich domains, rendering significant charge
146 collection losses in the segregated device. This is supported by a significant increase in the rise and
147 fall time during transient-photocurrent measurements, and dramatic shortening of the PL lifetime in
148 the phase-segregated control and PA-modified samples (Fig. 2h and Supplementary Fig. 11)²⁴.
149 Notably, the relative loss in V_{oc} after halide segregation is much smaller, in agreement with previous

150 reports^{20,27}. When we employ Mott–Schottky analysis (Supplementary Fig. 12), its account of loss in
151 internal electric field is also minimal. We offer that light-induced halide segregation will mostly not
152 affect charge selectivity and that interfacial carrier transport is little affected⁴³.

153 In contrast, the current extraction and carrier lifetime of PDA-modified perovskites are
154 significantly less impacted by halide segregation. Although the PL intensity is enhanced by several
155 orders of magnitude upon prolonged illumination (up to 10^5 s), PDA-modified devices show much
156 slower degradation of photovoltaic performance, retaining ~75% of their initial PCE after ~28 h (10^5
157 s). Based on the above findings we infer that PDA additive promotes current extraction from the I-
158 rich low-bandgap domains and improves the operating stability of wide-bandgap PSCs.

159

160 **Formation kinetics of Br-rich wide-bandgap perovskites**

161 To gain mechanistic insights into the role of PDA additive, we carried out detailed microstructural
162 and compositional analysis for Br-rich wide-bandgap perovskites. The grazing-incidence wide-angle
163 X-ray scattering (GIWAXS) patterns suggest that no low-dimensional perovskite or secondary phases
164 are formed upon adding PDA into the perovskite films (Fig. 3a,b). The control and PDA-added
165 perovskite films exhibit isotropic orientation of crystallites with a q -space value of 1.02 \AA^{-1} for the
166 (100) Br-rich perovskite peak⁴⁴. In addition, analysis of the XRD patterns and UV-vis spectra
167 suggests that PDA does not significantly change the cubic lattice constant or bandgap of perovskite
168 films (Fig. 3c,d and Supplementary Fig. 13). This implies that the PDA additive is not mixed into the
169 perovskite crystal lattice^{32,45}. Top-view scanning electron microscopy (SEM) images suggest that
170 adding PDA slightly reduces the grain size of perovskite films (Supplementary Fig. 14).
171 Photoinduced force microscopy (PiFM) and surface X-ray photoelectron spectroscopy (XPS) confirm
172 that PDA additives are retained in the perovskite film (Supplementary Figs. 15–16). Compared to the
173 control film, we also found that Pb 4*f*, I 3*d*, and Br 3*d* core-level peaks of PDA-added films are shifted
174 toward lower binding energies (Supplementary Fig. 17). We infer that PDA possibly forms chemical
175 bonds with the lead halide moieties near the perovskite grain boundaries, leading to a change in the
176 electrostatic interaction between the Pb^{2+} and the halogen anions⁴⁰.

177 To study the impact of PDA on film formation kinetics, we performed in situ UV-vis
178 absorption measurements during spin-coating and thermal annealing of wide-bandgap perovskite
179 films (see Methods and Supplementary Fig. 18). Figs. 3e–f show that directly after dripping the
180 antisolvent (at 39 s), an absorption onset near the 2.3 eV region appears and starts to redshift before
181 the end of the process at ~45 s. This corresponds to the removal of excess solvent, and growth of
182 perovskite nanocrystals with increased bandgap due to quantum confinement^{21,46}. We found that for
183 the as-cast control film, the absorption onset approaches ~2.15 eV, while for the as-cast 1 mol% PDA-
184 added film, the absorption remains relatively low, with an onset above 2.20 eV (Fig. 3g). When the

185 PDA concentration is increased to 20 mol%, the absorption from any perovskite phase is suppressed
186 without annealing (Fig. 3g and Supplementary Fig. 19). This confirms the capacity of PDA to interact
187 with the perovskite precursors and slow the formation of perovskites during spin-coating.
188 Furthermore, whereas annealing the control film at 100 °C immediately gives rise to the characteristic
189 absorption spectra of mixed-halide wide-bandgap perovskites, PDA-added films exhibit delayed
190 crystallization (Supplementary Figs. 20–21). This corresponds to an increased (100) peak intensity
191 for up to 10 mol% PDA-added perovskite films; we propose that the delayed growth of perovskite
192 produces more uniform film formation and higher crystallinity (Fig. 3c). We thus speculate that PDA
193 enables controlled growth among halide species and facilitates homogeneous transformation to
194 intermixed perovskite phases.

195 We performed XPS depth profile measurements to investigate the phase distribution of wide-
196 bandgap perovskite films (Supplementary Fig. 22 and Fig. 3i). Interestingly, the control and PA-
197 added films show the formation of a more Br-rich perovskite phase at the HTL interface. In
198 comparison, the PDA-modified perovskite film is much more homogeneous, showing even
199 distributions of halides (I, Br, and Cl) during sputtering. The results align with spatially resolved
200 luminescence imaging, wherein a much narrower QFLS distribution and higher mean QFLS values
201 are obtained for the PDA-added films toward the bottom interface (Supplementary Fig. 23). SEM
202 cathodoluminescence (SEM-CL) hyperspectral imaging suggests that PDA-added films show less
203 beam-induced ion migration and improved homogeneity at the microscopic scale (Supplementary Fig.
204 24). For mixed-halide perovskite precursors, density functional theory (DFT) calculations suggest
205 that the formation of Br-based perovskite species is more thermodynamically favored than I-species,
206 which results in large phase heterogeneity during a fast crystallization process (Supplementary Fig.
207 25). In comparison, PDA cations bind strongly to the interfaces of adjacent perovskite fragments,
208 allowing more relaxation time for different halide ions to exchange and thereby the formation of
209 uniform mixed-halide perovskite phases (Fig. 3h and Supplementary Note 1).

210 We discuss here the idea that, in control devices, heterogeneity generated prior to halide
211 segregation occurs could induce a higher defect density towards the bottom interface, contributing to
212 the non-radiative recombination losses at the perovskite/HTL interface. Chemical heterogeneity is
213 correlated with lattice distortions and defective sites near the grain boundaries^{47,48}, further
214 aggravating carrier trapping processes and shortening charge-carrier lifetimes as light-induced halide
215 segregation now occurs in a more defective region (Supplementary Fig. 11)^{25,27}. By contrast, PDA
216 improves the halide homogeneity of perovskite films and eliminates detrimental interfacial losses in
217 the presence of an HTL. Under illumination, PDA could therefore provide effective passivation
218 around grain boundaries, reducing trap-mediated recombination and enabling more efficient current
219 extraction of funneled charge carriers from the device. We note that further suppressing the light-

220 induced halide segregation by introducing lattice compression and distortion can increase device
221 stability in Br-rich perovskite solar cells^{15,49}. Further investigations of the compositional space of Br-
222 rich hybrid organic-inorganic perovskite systems are thus of interest (Supplementary Fig. 26).

223

224 **Monolithic all-perovskite 3J solar cell performance**

225 Monolithic all-perovskite 3J solar cells comprising 1.97 eV wide-bandgap, 1.61 eV mid-bandgap
226 $\text{Cs}_{0.05}\text{FA}_{0.9}\text{MA}_{0.05}\text{Pb}(\text{I}_{0.85}\text{Br}_{0.15})_3$, and 1.25 eV narrow-bandgap $\text{Cs}_{0.05}\text{FA}_{0.7}\text{MA}_{0.25}\text{Pb}_{0.5}\text{Sn}_{0.5}\text{I}_3$
227 absorbers were produced based on a device structure shown in Fig. 4a. A transparent hydrogenated
228 indium oxide (IOH) front contact was used to reduce optical losses due to parasitic absorption in the
229 NIR range⁵⁰ (Supplementary Fig. 27). The 1.97 and 1.61 eV sub-cells are connected through the
230 PCBM/polyethyleneimine (PEI)/ SnO_x /ITO/ NiO_x /Me-4PACz recombination junction. Between the
231 1.61 and 1.25 eV sub-cells, a C_{60} / SnO_x /Au/poly(3,4-ethylene dioxythiophene):polystyrene
232 (PEDOT:PSS) recombination junction was used^{2,6}. We note that PCBM in the 1.97 eV sub-cell yields
233 a higher V_{oc} than C_{60} due to reduced interfacial recombination losses (Supplementary Fig. 28)⁵¹. Here,
234 the SnO_x layer was deposited by a temporal atomic layer deposition (ALD) process, and the PEI layer
235 facilitates the nucleation of SnO_x atop the PCBM (Supplementary Fig. 29)⁵². In contrast to
236 PEDOT:PSS, Me-4PACz provides a lossless HTL interface with 1.61 eV perovskites³⁷. To ensure
237 better coverage of Me-4PACz and thereby a shunt-free recombination junction, we used sputter-
238 coated ITO and solution-processed NiO_x layers^{38,50}. Enabled by the high performance of PDA-
239 modified wide-bandgap sub-cells, the champion 3J device exhibits a high PCE of 25.1% (0.049 cm^2),
240 with a V_{oc} of 3.33 V, a J_{sc} of 9.7 mA cm^{-2} , and FF of 0.78 (Fig. 4b). This high V_{oc} represents a 130-
241 mV loss compared to the summed V_{oc} (3.47 V) of the single-junction PSCs. The 3J cells display stable
242 operating performance, retaining 80% of their initial PCE after 200 hours of continuous MPP tracking
243 under AM1.5G 1-sun illumination (Fig. 4g). Additionally, the 3J cell retained above 86% of its initial
244 efficiency after more than 168 hours of storage at 65°C in N_2 (Supplementary Fig. 30). We also
245 observed negligible hysteresis in the 3J devices (Supplementary Fig. 31). A statistical summary
246 demonstrates good reproducibility of the 3J cells (Supplementary Fig. 32). For PDA-added 3J devices,
247 a record efficiency of 23.87% was certified by NREL PV calibration laboratory using the asymptotic
248 maximum power scan protocol⁵³. The result outperforms the control (wide-bandgap perovskite
249 passivated by PDA only at the surface) 3J device (21.49%), where lower cell metrics were recorded
250 during the measurement (Fig. 4c and Supplementary Figs. 33–34).

251 Exploiting the full efficiency potential of all-perovskite 3J solar cells will require further
252 reduction of optical losses and improved matching of the photo-generated current from all three sub-
253 cells. Notably, the EQE-integrated J_{sc} was 12.6, 9.0, and 10.0 mA cm^{-2} for the wide-, mid-, and
254 narrow-bandgap sub-cells, respectively, suggesting a large current mismatch of 3.6 mA cm^{-2} , with

255 the middle sub-cell limiting the J_{sc} of the 3J device (Fig. 4d). To explore the efficiency potential of
256 all-perovskite 3J, we performed optical modeling using the transfer matrix (TM) method (Fig. 4e and
257 Supplementary Figs. 35–36) and reconstructed the J – V characteristics of the triple cell by scaling the
258 J – V curves of reference single-junction devices to the J_{sc} from simulated EQE, and adding their
259 voltages at each current value (considering a 130 mV V_{oc} loss)⁵⁴. Fine-tuning the thickness of
260 perovskite absorbers and implementing more optically transparent layers (such as ITO and
261 NiO_x/SAM ^{50,55}) should reduce the current mismatch and parasitic absorption losses of the narrow-
262 bandgap sub-cell (Supplementary Figs. 37–39). This enables a J_{sc} of $\sim 11 \text{ mA cm}^{-2}$ in the 3J device
263 and improves the PCE to 29.7%, which surpasses existing state-of-the-art all-perovskite 2J solar cells⁷.
264 Here, we highlight the importance of enhancing optical responses to advance the PCE of 3J devices,
265 and note that this J_{sc} is above previous estimates from simulation work¹¹ wherein impractical
266 perovskite layer thicknesses were used. We note, however, that simply increasing the bandgap by
267 adding more Br or Cl in the 1.97 eV perovskite could result in larger surface roughness and lower
268 device performance (Supplementary Figs. 2–3 and 40–41). Further device engineering is required in
269 order to retain high performance and stability after reducing the thickness or increasing the bandgap
270 of the wide-bandgap perovskite absorbers.

271 Furthermore, we anticipate that the V_{oc} of each sub-cell equaling $\sim 95\%$ of its DB limit is
272 feasible through optimizations of the perovskite absorber and the limiting perovskite/charge-
273 transporting layer interfaces. As such, a V_{oc} of 3.71 V can be expected in the 3J device (considering
274 a 100-mV loss). Ultimately, combining a J_{sc} of 11.0 mA cm^{-2} and an FF of 0.83 should enable a PCE
275 of 33.7%, exceeding the DB limit for single-junction solar cells and simulations^{10,11}. Overall, our
276 experimental results and calculations demonstrate the immense potential of all-perovskite 3J devices.

277

278 **Conclusions**

279 We reported a strategy to enhance the performance and operating stability of Br-rich wide-bandgap
280 PSCs. Incorporating the diammonium PDA molecule into Br-rich perovskites has enabled controlled
281 film growth from solution, and improved halide homogeneity. Benefiting from the passivation effect
282 of PDA near grain boundaries, Br-rich wide-bandgap PSCs retain fast charge collection processes
283 and significantly improved stability under prolonged illumination. The PDA-added perovskites have
284 effectively suppressed energetic losses at the perovskite/charge-transporting layer interfaces, leading
285 to a record V_{oc} of 1.44 V at a bandgap of 1.97 eV. Combining 1.97, 1.61, and 1.25 eV perovskite sub-
286 cells, we fabricated record monolithic all-perovskite triple-junction solar cells with an in-lab PCE of
287 25.1% (certified 23.87%) with a high V_{oc} of 3.33 V. Guided by optical simulation, we have
288 demonstrated a route for all-perovskite 3J devices to surpass state-of-the-art all-perovskite tandem
289 technologies and further to achieving a PCE approaching the DB limit.

290

291 **Methods**

292 **Materials.** All materials were purchased from suppliers and used without purification unless stated otherwise.
293 Patterned glass substrates ($3 \times 3 \text{ cm}^2$) with ITO were purchased from Naranjo Substrates. CsI (99.999%), PbI₂
294 (99.99%), PbBr₂ (>98.0%), PbCl₂ (>99.0%), and Me-4PACz (>99.0%) were purchased from TCI Chemicals.
295 Organic halides, including FAI, MAI, MABr, MACl, PDAI₂ ($\geq 98\%$), PAI, and 4-fluoro-phenethylammonium
296 bromide (4F-PEABr), were purchased from Greatcell Solar Materials. SnI₂ (99.99%), SnF₂ (99%), glycine
297 hydrochloride (99%), guanidine thiocyanate (GuaSCN, 99%), PEI solution ($M_n \sim 60,000 \text{ g mol}^{-1}$, $M_w \sim 750,000$
298 g mol^{-1} , 50 wt% in H₂O), ethane-1,2-diammonium iodide (EDA₂I, 98%), and all anhydrous solvents were
299 purchased from Sigma-Aldrich. PCBM (99%) was purchased from Nano-C. PEDOT:PSS (AI 4083) was
300 purchased from Heraeus Clevis. C₆₀ and BCP were purchased from Xi'an Polymer Light Technology.

301

302 **Perovskite precursor solutions.** Wide-bandgap perovskite precursor solutions were prepared by dissolving 1
303 M CsI (0.15), PbI₂ (0.1), PbBr₂ (0.9), FAI (0.85), PbCl₂ (0.053), MACl (0.053) in *N,N*-dimethylformamide
304 (DMF, anhydrous 99.8%) and dimethyl sulfoxide (DMSO, anhydrous 99.9%) at a volume ratio of 4:1. PAI or
305 PDAI₂ (1–20 mol%) was added to the precursor solution. Mid-bandgap perovskite precursor solutions were
306 prepared by dissolving 1.5 M CsI (0.05), PbI₂ (0.8), PbBr₂ (0.2), FAI (0.9), MABr (0.05) in DMF and DMSO
307 at a volume ratio of 4:1. Narrow-bandgap perovskite precursor solutions were prepared by dissolving 1.8 M
308 CsI (0.05), PbI₂ (0.5), SnI₂ (0.5), SnF₂ (0.05), FAI (0.7), MAI (0.25), GuaSCN (4 mg mL⁻¹), 4F-PEABr (2 mg
309 mL⁻¹), glycine hydrochloride (4 mg mL⁻¹), and Sn powder (5 mg mL⁻¹) in DMF and DMSO at a volume ratio
310 of 3:1. All the solutions were stirred at 60 °C for 1 h and filtered by 0.22 μm PTFE syringe filter before use.

311

312 **Device fabrication.** Glass/ITO or Glass/IOH substrates were sequentially cleaned with acetone, soapy water,
313 deionized water, 2-propanol, and 30 min of UV-ozone before use. For both wide-bandgap and mid-bandgap
314 single-junction perovskite solar cells, NiO_x nanoparticles (prepared following a previously reported method³⁸,
315 5 mg mL⁻¹ at a 3:1 volume ratio of H₂O and 2-propanol) were spin-coated on the substrates at 3000 rpm for
316 30 s (1000 rpm s⁻¹ acceleration) in the air without any post-treatment. The substrates were then transferred
317 into a N₂ glovebox, where Me-4PACz solution (0.3 mg mL⁻¹ in ethanol) was spin-coated at 3000 rpm for 30
318 s (1000 rpm s⁻¹ acceleration) and annealed at 100 °C for 10 min. After cooling, wide-bandgap perovskite
319 precursor solution was dropped onto the substrate and spin-coated at 1000 rpm for 10 s (500 rpm s⁻¹
320 acceleration) and 5000 rpm for 30 s. 150 μL anisole was dropped onto the substrate at 5 s before the end. The
321 perovskite precursor film was immediately annealed at 100 °C for 10 min. After cooling, the PDAI₂ solution
322 (0.5 mg mL⁻¹ in 2-propanol/CB at a volume ratio of 2:1) was dynamically spin-coated on the perovskite film
323 at 4000 rpm for 30 s and annealed at 100 °C for 5 min. Afterward, for wide-bandgap perovskites, PCBM (15
324 mg mL⁻¹ in chlorobenzene, CB) was spin-coated at 1000 rpm for 30 s (1000 rpm s⁻¹ acceleration) while 20
325 nm C₆₀ was evaporated onto the mid-bandgap perovskites, before depositing 8 nm BCP and 140 nm Ag on top
326 by thermal evaporation.

327 For Pb-Sn narrow-bandgap perovskite solar cells, PEDOT:PSS (diluted in IPA at a volume ratio of 1:2) was
328 spin-coated on the substrate at 4000 rpm for 30 s and annealed at 120 °C for 10 min in air. The substrates were
329 then transferred to a N₂ glovebox, on which perovskite precursor solution was spin-coated at 1000 rpm for 10
330 s (200 rpm s⁻¹) and 3800 rpm for 45 s (1000 rpm s⁻¹). 300 μL CB was dropped at 18 s before the end. The
331 perovskite precursor film was immediately annealed at 100 °C for 10 min. EDAl₂ (0.5 mg mL⁻¹ dissolved in
332 2-propanol:toluene at a volume ratio of 4:3) at 4000 rpm for 25 s (1300 rpm s⁻¹) and annealed at 100 °C for 5
333 min. Afterward, 20 nm of C₆₀, 8 nm of BCP, and 140 nm of Ag were deposited sequentially by thermal
334 evaporation.

335 For all-perovskite 3J solar cells: The single-junction perovskite absorbers were prepared as previously
336 described. Between wide- and mid-bandgap perovskite sub-cells, after depositing PCBM, PEI solution (0.025
337 wt% diluted in 2-propanol) was spin-coated at 4000 rpm for 30 s (2000 rpm s⁻¹ acceleration) without any post-
338 treatment. 20 nm of ALD-SnO_x was then deposited on top and followed by sputtering of 20 nm ITO. Between
339 mid- and narrow-bandgap perovskite sub-cells, 20 nm of ALD-SnO_x was deposited on C₆₀ and followed by 1
340 nm of Au by thermal evaporation before spin-coating PEDOT:PSS in the air. Shadow masks were used to
341 ensure that only active areas were covered with ITO and Au to avoid shunt losses. Instead of BCP, 20 nm of
342 ALD-SnO_x was deposited atop narrow-bandgap perovskite and C₆₀.

343 **Device characterizations.** *J–V* measurements were performed in a N₂ atmosphere at room temperature, using
344 a solar simulator (Newport, Class A) with a light intensity of 100 mW cm⁻² (calibrated by a reference solar
345 cell from Newport) and a Keithley 2400 source meter. The voltage was scanned from the *V*_{oc} to the *J*_{sc} (reverse)
346 and from the *J*_{sc} to the *V*_{oc} (forward) directions with a scanning rate of 100 mV s⁻¹ (voltage step of 10 mV).
347 The active area was determined by the aperture mask (0.049 cm²). A spectral mismatch of 1.0 was used for all
348 measurements. For stability measurements, solar cells were operated at the maximum power point and
349 measured the output current, using G2V Sunbrick with a simulated AM1.5G 1-sun illumination. EQE
350 measurements were performed in ambient air (Enlitech). For 3J solar cells, LED bias light (735+940 nm for
351 1.97 eV wide sub-cell, 470+940 nm for 1.61 eV mid sub-cell, and 470+735 nm for 1.25 eV narrow sub-cell)
352 and bias voltage equivalent to the summed *V*_{oc} of optically biased sub-cells were applied to measure the EQE
353 response of each sub-cell⁵⁴.

354 For in situ PL and *J–V* measurements²⁷, wide-bandgap PSCs were sealed in a nitrogen-filled box with a quartz
355 window and connected to a Keithley 2400 source meter. A green LED (Thorlabs M530L3, 530 nm), filtered
356 by a 600 nm short pass filter, was used to illuminate the device at a 1-sun equivalent light intensity such that
357 the solar cell initially generates a photocurrent equal to the *J*_{sc} under simulated AM1.5G illumination. The PL
358 light, filtered by a 590 nm long pass filter, was focused on an optical fiber and recorded using a spectrometer
359 (Avantes Avaspec-2048x14). Firstly, the solar cell was held at the *V*_{oc}, and a PL spectrum was recorded with
360 an integration time of 100 ms from a spot in the active area of the solar cell. This is followed by a reverse
361 voltage sweep from +1.5 to -0.1 V and a forward voltage sweep from -0.1 to +1.5 V at a scan rate of 120 mV

362 s⁻¹ (voltage step of 16 mV). Afterward, the solar cell was held at a voltage bias close to the maximum power
363 point under illumination for a certain delay before starting the next PL and *J*–*V* measurement.

364 The Mott–Schottky analysis was performed by measuring the capacitance–voltage (*C*–*V*) characteristics of
365 solar cells using an electrochemical station (Solartron SI1260 impedance/gain-phase analyzer) in an N₂ box
366 under a dark condition. The capacitance was measured by sweeping the voltage (DC) from –0.5 V to 3.0 V.
367 The frequency was kept at 10 kHz with a perturbation voltage of 10 mV.

368 **Film characterizations.** Absolute PL measurements were performed using an integrating sphere (Avantes,
369 AvaSphere-30-REFL) installed in a N₂ glovebox, where a 455 nm LED (Thorlabs M455F3-455 fiber coupled
370 LED source) was used to excite the perovskite film (at ~1 sun equivalent intensity) through an optical fiber
371 fitted with a 550 nm short pass filter. The spectra were measured through an optical fiber connected to a
372 calibrated AvaSpec-HERO spectrometer (Avantes) using a 550 nm long pass filter. XRD patterns were
373 recorded by a Bruker D2 Phaser diffractometer equipped with Cu K α radiation ($\lambda = 1.5405 \text{ \AA}$). GIWAXS
374 measurements were performed at the BXDS (LE-wiggler) beamline of the Canadian Light Source (CLS) using
375 X-rays of a wavelength of $\lambda = 0.84 \text{ \AA}$, at a grazing incidence angle of 1.0°. Ex situ UV-vis absorption spectra
376 were acquired by PerkinElmer Lambda 900 UV-vis-NIR spectrophotometer. Surface and depth profile XPS
377 measurements were measured using a Thermo Scientific K-Alpha with a 180° double-focusing hemispherical
378 analyzer and a 128-channel detector. A monochromatic Al K α radiation (1486.6 eV) was used, and the X-ray
379 spot size was 400 μm . For the sputtering experiments, the perovskite/HTL sample was removed layer-by-layer
380 using argon ion etching operated at mid-current and ion energy of 3000 eV. A snapshot mode was used for
381 each element with the number of frames 5 \times 1 s.

382 In situ UV-vis absorption measurements were performed in a N₂ glovebox during spin-coating and thermal
383 annealing⁵⁶. The backside of the glass substrates was painted with white paint and dried before UV-ozone
384 treatment. The substrate was placed on a spin coater or a hot plate and illuminated by a white light from a
385 halogen lamp. A fiber optical cable was placed at an off-specular angle which collects the light scattered from
386 the paint and transmitted through the as-cast film. The fiber was connected to a spectrometer and the absorption
387 spectra can be calculated using $A(\lambda) = -\log_{10}\left(\frac{I_m(\lambda)-I_{m,dark}(\lambda)}{I_{m,blank}(\lambda)-I_{m,dark}(\lambda)}\right)$, where I_m , $I_{m,dark}$, and $I_{m,blank}$ represent
388 wavelength-dependent photon counts of the sample during measurement, the dark reference before turning on
389 the lamp, and a blank reference for an empty substrate with white paint, respectively.

390 CL hyperspectral maps were acquired in an Attolight Allalin 4027 Chronos in continuous wave mode, at room
391 temperature and under a high vacuum. All secondary electron (SE) images and CL maps were acquired at 3
392 kV acceleration voltage, approximately 63 pA beam current, and 128 \times 128 pixels resolution in a 3.14 μm
393 field of view. CL intensities over specific energy ranges were integrated to generate energy bandpass-filtered
394 images.

395 The luminescence imaging and QFLS mapping were performed by a previously reported method⁵⁷. A 440 nm
396 LED (Thorlabs, M450LP1) was used to illuminate perovskite films deposited on glass substrates at either the
397 front or the back surfaces. The photon flux was calibrated at ~1 sun intensity for wide-bandgap perovskites
398 (Multicomp PRO MP710086 bench power supply). An imaging camera (Andor Zyla 4.2 sCMOS sensor)
399 equipped with a Kowa LM50XC lens (50 mm lens with up to an F2.0 aperture) to acquire the PL images.

400 Photoinduced force microscopy was collected with a Molecular Vista instrument coupled to a QCL laser. The
401 images and spectra were collected using the sideband mode with a PPP-NCHAu gold coated AFM Probe from
402 Nanosensors.

403 **DFT calculations.** Ab initio molecular dynamic simulations (AIMD) were performed with CP2K code⁵⁸. We
404 used the mixed Gaussian and plane-wave method (QUICKSTEP formalism) and the Perdew-Burke-Ernzerhof
405 exchange-correlation functional (PBE) within the generalized gradient approximation (GGA)^{59,60}. The van der
406 Waals interactions through the DFT-D3 scheme⁶¹ were included to account for dispersion interactions. The
407 Gaussian basis set employed for all atoms is a double- ζ polarized basis set with polarization functions in
408 conjunction with the Goedecker-Teter-Hutter (GTH) pseudopotentials^{62,63}. To investigate the possible
409 ammonium cation-perovskite interactions, we constructed PDA-perovskite interface, PDA-perovskite surface,
410 and PA-perovskite surface models, respectively. As the FA cation is the dominant A cation in experimentally
411 fabricated perovskite films, we thus only, to reduce the complexity, consider the interaction modes of PDA
412 and PA with FAPbX₃ (X = I or Br). A constant-pressure and constant-temperature (NPT) ensemble was used
413 for the interface model whereas for surface model we used a constant-volume and constant-temperature (NVT)
414 ensemble. The temperature was controlled with a Nose-Hoover thermostat⁶⁴ at room temperature (300 K).
415 Each simulation was run for 10 ps to ensure equilibrium and the time step was set to 1.0 fs. The plane-wave
416 cutoff for the expansion of the electron density was set to 300 Ry. The convergence criteria on forces were set
417 at 4.5×10^{-4} Har/Bohr (0.02 eV/Å). We observe in the AIMD simulations that the halide anions tend to escape
418 from the perovskite surface in the PDA-perovskite surface model. The PDA-perovskite surface model is thus
419 not considered in further calculations.

420 The binding energies (E_b) of ammonium cations with the perovskite interface or surface are defined as $E_b =$
421 $E_{\text{mol/pvsk}} - E_{\text{pvsk}} - E_{\text{mol}}$, where $E_{\text{mol/pvsk}}$, E_{pvsk} , and E_{mol} are the total energies of the ammonium cation-perovskite
422 crystal system, the perovskite system after removing an ammonium halide compound (PDAX₂/PAX, X = I or
423 Br), and the ammonium-halide compound, respectively. Here, the ammonium cation-perovskite systems are
424 taken from the snapshots of the AIMD equilibrium simulations. The atomic positions of these systems are
425 optimized further by CP2K code at the ground state.

426 To calculate the formation energy of the perovskite cluster, we constructed the cubic supercell of $20 \times 20 \times 20$
427 Å with a cluster in the center of the cell. Atomic positions of the cluster are optimized by the Vienna ab initio
428 simulation package (VASP)⁶⁵. The formation energies (E_f) of perovskite cluster with respect to the
429 corresponding binary species is defined as $E_f = E_c(\text{AX:PbX}_2) - E_c(\text{AX}) - E_c(\text{PbX}_2)$, where $E_c(\text{AX:PbX}_2)$,

430 $E_c(\text{AX}_2)$, and $E_c(\text{PbX}_2)$ are the total energies of perovskite cluster, AX (A = FA, Cs; X = I, Br) cluster, and
431 BX_2 cluster, respectively.

432 **Optical simulations.** Optical modeling was performed with GenPro4 program⁶⁶. The optical constants of
433 wavelength-dependent refractive indices and extinction coefficient of all layers were measured individually
434 by depositing corresponding materials on silicon and characterized with a J.A. Wollam ellipsometer.

435
436 **Data availability**

437 All data generated or analyzed during this study are included in the published article and its
438 Supplementary Information and Source Data files. Further data that support the findings of this study
439 are available from the corresponding authors on reasonable request.

440

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464

465 **Author contributions**

466 J.W., L.Z., D.Z., M.W., R.J., and E.S. planned the research and analyzed the results. J.W. and L.Z. fabricated
467 the triple-junction cells. J.W., L.Z., H.C., A.M., and C.L. prepared the triple cells for certifications. J.W.
468 optimized the wide-bandgap sub-cells and coordinated all the characterization of materials and devices. H.C.
469 optimized the mid-bandgap sub-cell. A.M. and C.L. optimized the narrow-bandgap sub-cell. W.R. performed
470 the absolute PL and QFLS analysis. K.D. developed and helped measure the in situ PL/ $J-V$ characteristics and
471 wrote the LabVIEW code. A.C. performed the XPS and in situ absorption measurements. L.Z., N.S. and L.B.
472 fabricated and characterized perovskite thin films for stability. Z.C. performed DFT calculations. K.H.
473 performed PiFM and analyzed the data. A.D. performed luminescence mapping and analysis. S.H. helped
474 analyze the XRD data and crystallization dynamics. H.S. facilitated and supervised the luminescence imaging
475 experiments. G.K. and R.A.O. carried out the SEM-CL measurements and analysis. R.O. performed the
476 transient photocurrent measurements. S.T. and L.G. performed the GIWAX measurements. D.Z. performed
477 optical simulations and provided the IOH substrates for triple-junction cells. Z.W. and B.C. provided valuable
478 suggestions for optimizing the wide-bandgap sub-cell and 3J device configurations. J.W. wrote the first
479 manuscript, and all authors commented on it. E.S. and R.J. supervised the project.

480

481 **Competing interests**

482 H.S. is cofounder and CSO of Oxford PV Ltd. All other authors declare no competing interests.

483

484 **Additional information**

485 **Supplementary Information** accompanies this paper.

486

487 **Fig. 1 | Photovoltaic performance of 1.97 eV wide-bandgap PSCs.** **a**, Inverted (p-i-n) device
488 structure. **b**, Bandgap-dependent V_{oc} for p-i-n single-junction PSCs with a PCE above 5%. The cell
489 data was acquired from an open-access perovskite database¹⁶. In the plot, the DB limit (solid line),
490 90% of the DB limit (dashed line), the highest V_{oc} of this work (solid star), and reported cell data
491 (open brackets) are shown. **c**, QFLS of control and 1 mol% PDA-added perovskite films deposited
492 on glass (neat), with PCBM on top (w ETL), and on ITO/NiO_x/Me-4PACz (w HTL). The perovskite
493 samples (unpassivated) are further post-treated with a PDA solution on the top surface (passivated).
494 **d**, $J-V$ characteristics (reverse scans) of single-junction 1.97 eV PSCs without (control, green) and
495 with 1 mol% PDA addition (purple) in the perovskite bulk. The top surface of unpassivated
496 perovskites (dashed lines) was treated with PDA (passivated, solid lines) on top. **e**, EQE spectrum of
497 the best-performing PSC. The J_{sc} (12.7 mA cm⁻²) was obtained by integrating with the AM1.5G
498 spectrum. **f**, Stabilized MPP tracking of encapsulated control, PA-, and PDA-modified PSCs at room
499 temperature in the air under simulated AM1.5G 1-sun illumination condition. The initial performance
500 of each device is displayed in Supplementary Fig. 5. Note that all the perovskite surfaces were
501 passivated.

502 **Fig. 2 | Photostability and cell performance of 1.97 eV wide-bandgap PSCs during in situ PL**
503 **and $J-V$ measurements.** Time-resolved PL intensity maps under a 530 nm LED illumination at one-
504 sun equivalent intensity for **a**, the control (~2.8 h), **b**, 1 mol% PA- (~16 h), and **c**, 1 mol% PDA-
505 modified (~28 h) PSCs measured at the V_{oc} , and selected PL spectra at different time intervals for **d**,
506 the control, **e**, PA-, and **f**, PDA-modified devices. **g**, Evolution of the photovoltaic parameters (V_{oc} ,
507 J_{sc} , FF, and PCE) under reverse scans and the PL intensities of 1.6–1.7 eV (I-segregated) phases for

508 1.97 eV PSCs under illumination. **h**, The rise (left) and fall (right) time (μs) extracted from transient-
509 photocurrent measurements for the control, PA-, and PDA-modified PSCs.

510 **Fig. 3 | Structure, composition, and formation kinetics of Br-rich wide-bandgap perovskite**
511 **films.** GIWAX patterns of **a**, the control and **b**, 1 mol% PDA-added perovskite films. **c**, XRD patterns
512 and **d**, cubic lattice parameters and bandgap at varying PDA concentrations (0–20 mol%). To exclude
513 the influence of halide on the lattice constant and optical bandgap, stoichiometric PDAI_2 , PDABr_2 ,
514 and PDACl_2 were used. Time-resolved absorption intensity maps during spin-coating of **e**, control,
515 and **f**, 1 mol% PDA-added perovskite precursor solutions. The antisolvent anisole was dropped onto
516 the spinning substrate at ~ 39 s. **g**, The absorption spectra of perovskite precursor films after the spin-
517 coating and after 50 s of thermal annealing at 100°C . **h**, Schematics of PDA cation and mixed Br/I
518 perovskite precursors during the crystallization process. **i**, The ratio of atomic concentrations between
519 halide (Br, I, and Cl) and Pb atoms obtained from XPS depth profiles for the control (solid lines) and
520 PDA-added (dashed lines) perovskite films. The dashed vertical line is a guide to show the interface
521 with the HTL.

522 **Fig. 4 | Photovoltaic performance and simulations of monolithic all-perovskite triple-junction**
523 **solar cells.** **a**, Device configuration of optimized 3J cell stack. **b**, J - V characteristics of the
524 representing 1.97, 1.61, and 1.25 eV, as well as the champion 3J solar cells. **c**, NREL-certified J - V
525 characteristics of 3J devices using a control (green) and PDA-modified (orange) 1.97 eV sub-cells.
526 **d**, EQE spectra of 1.97 (12.6 mA cm^{-2}), 1.61 (9.0 mA cm^{-2}), and 1.25 (10.0 mA cm^{-2}) eV sub-cells
527 in a 3J solar cell. **e**, EQE spectra of 1.97 (10.9 and 11.0 mA cm^{-2}), 1.61 (10.8 and 11.0 mA cm^{-2}),
528 and 1.25 (10.2 and 11.0 mA cm^{-2}) eV sub-cells after thickness optimizations to approach a current-
529 matched condition (dashed lines) and after using optically transparent recombination junction (solid
530 lines). A total EQE spectrum (gray dotted line) represents the summed EQE of all three sub-cells. In
531 both Fig. 4d and e, the J_{sc} was obtained by integrating with the AM1.5G spectrum. **f**, Experimental
532 and reconstructed J - V characteristics of the 3J device from the simulated EQE at different optical
533 conditions and assuming a V_{oc} of 95% of the detailed-balance limit was achieved in each sub-cell. **g**,
534 MPP tracking stability of an encapsulated 3J device in the air at room temperature under simulated
535 AM1.5G 1-sun illumination. The initial performance is displayed in Supplementary Fig. 30. T_{80} is
536 the time for device stability to reduce to 80% of its initial PCE.

537

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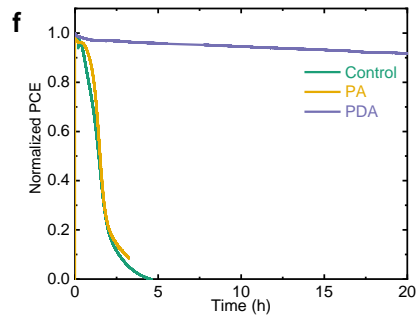
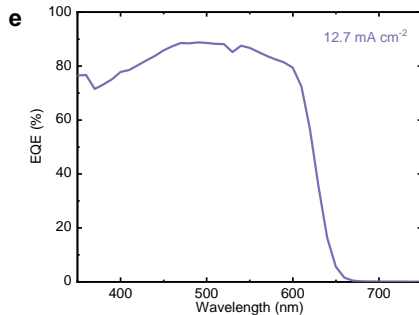
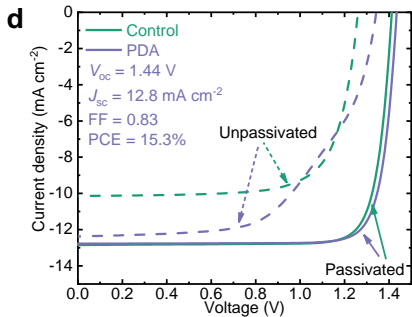
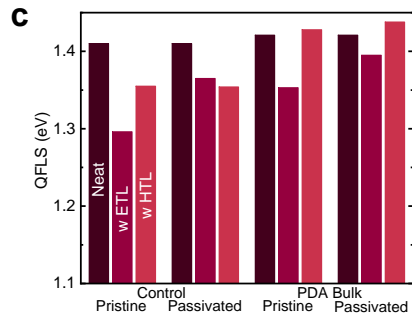
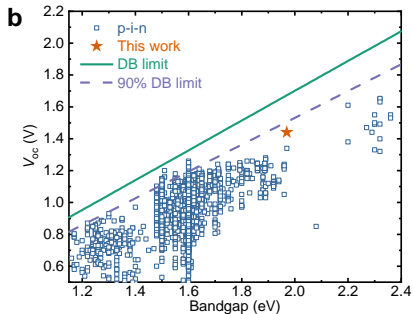
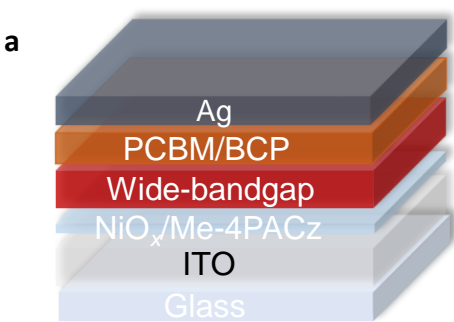
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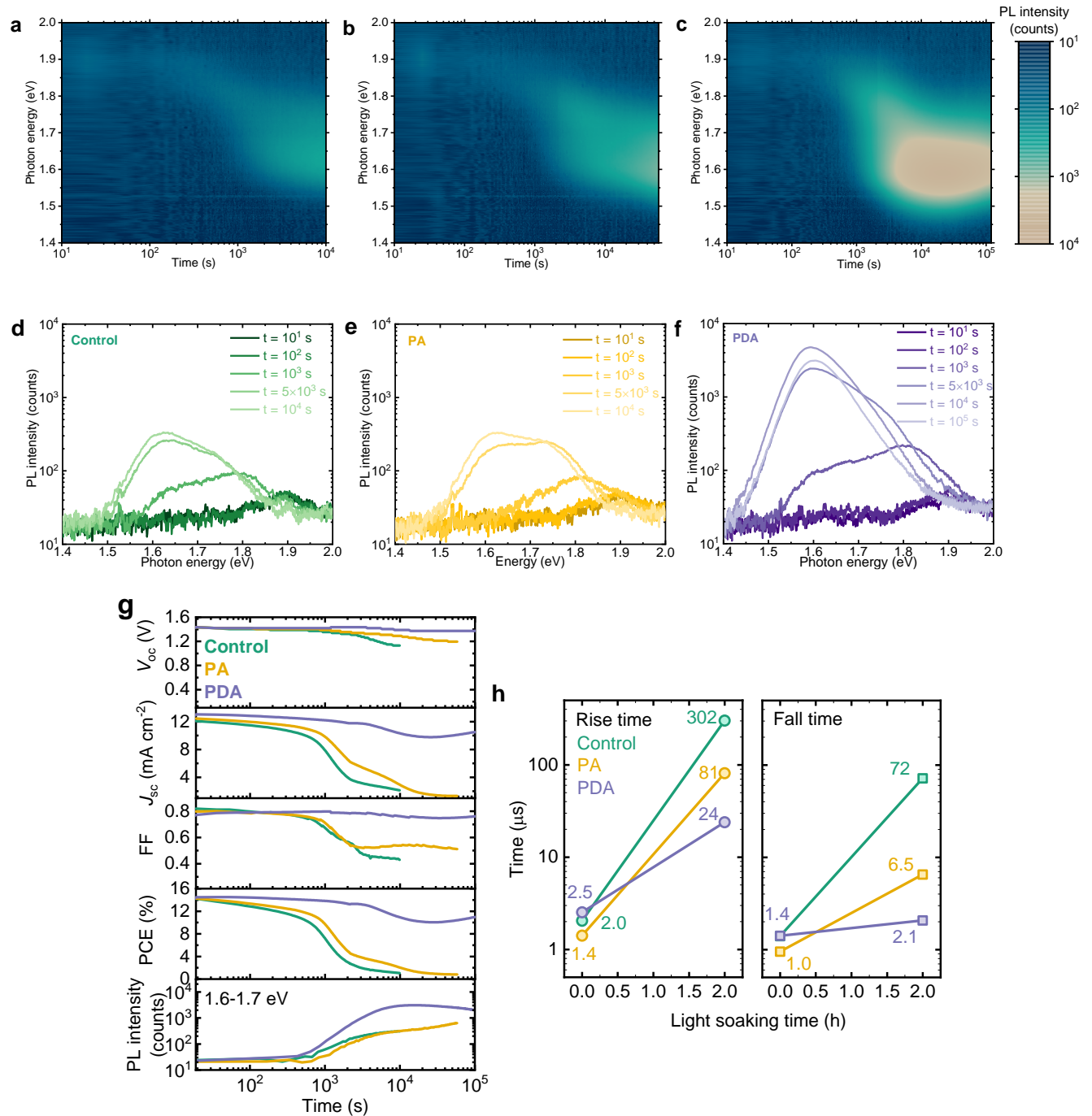
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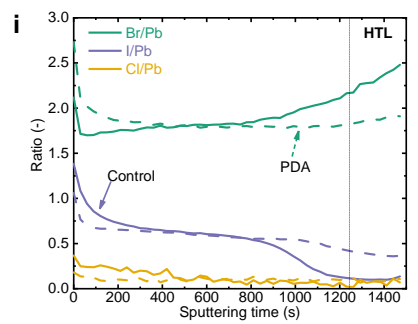
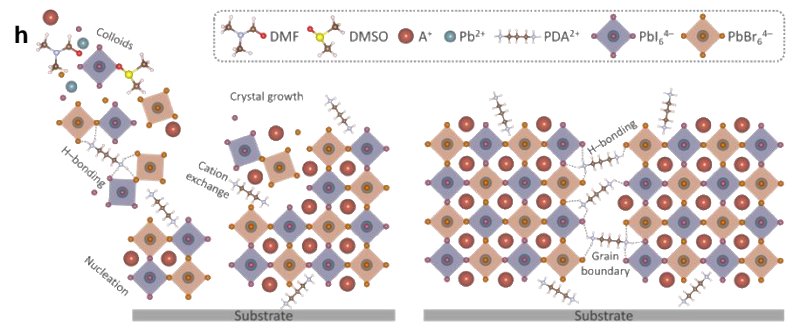
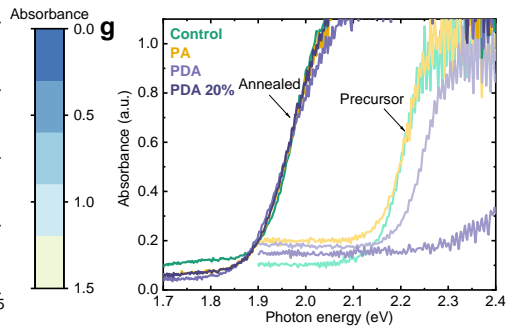
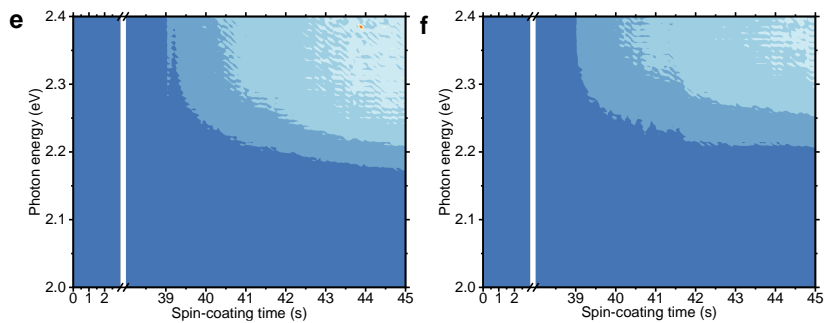
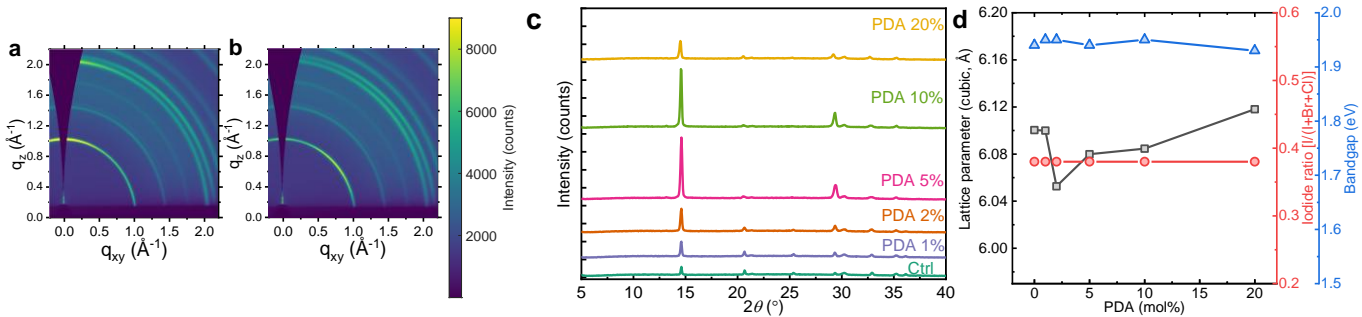
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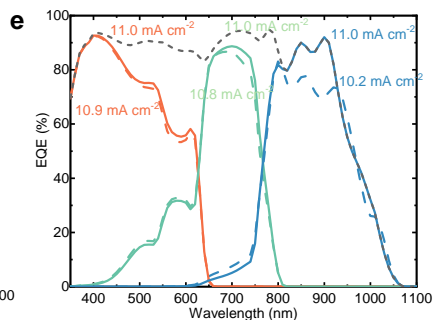
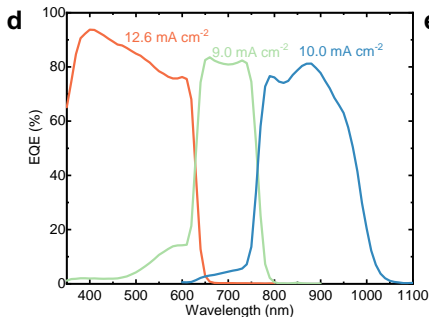
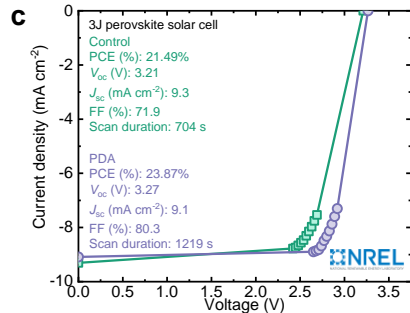
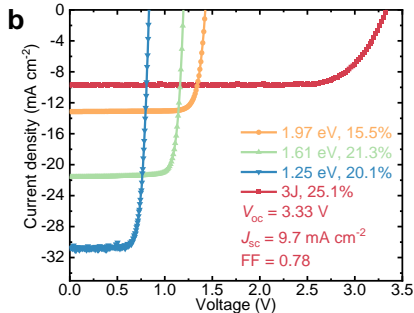
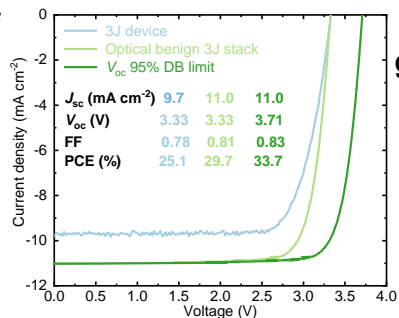
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