

Environmental Effects of Ashfall in Argentina from the 2008 Chaitén Volcanic Eruption

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Abstract

Analyses of air, water and vegetation samples collected in June 2008 offer new insights into the environmental effects of the May 2008 Chaitén eruption on Argentina, which was subject to significant ashfall between 42°S and 46°S.

Results from air filtration in the ash-affected town of Esquel (with samples analysed by gravimetry and scanning electron microscopy) show the total mass of resuspended ash in the air is well-correlated with traffic activity. However, this variation is primarily related to varying amounts of the largest particles, with little variation in the amounts of fine ash particles (i.e., $d < 4 \mu\text{m}$). This result suggests that the hazard associated with resuspended ash remains high even when traffic activity is low and the air is not visibly dusty. We estimate $\text{PM}_{2.5} \sim 200 \mu\text{g m}^{-3}$, $\text{PM}_4 \sim 300 \mu\text{g m}^{-3}$ and $\text{PM}_{10} \sim 1000 \mu\text{g m}^{-3}$; these concentrations far exceed WHO air quality guidelines and likely persisted for several months.

Results from water and vegetation sampling (with samples analysed by inductively coupled plasma mass spectrometry and ion chromatography) indicate that ashfall resulted in significant compositional changes in ephemeral lakes and Coirón grass (*Festuca pallescens*). For B, Cd, Zn, Tl, Cu and Ni, there are strong linear correlations between concentrations and ash thickness (where $>2 \text{ mm}$) in both datasets. These results suggest that the eruption of Chaitén led to significant changes in the concentrations of trace volatile elements within the environment. Analysis of vegetation samples collected in January 2009 indicates that the elevated element concentrations in Coirón grass persisted for < 8 months. These results offer insights into the environmental fate of volatile trace elements emitted during volcanic eruptions.

Keywords

Environment; Ash; Leachate; Chaitén; Patagonia; Eruption; Volcanic; Coirón

31 **1 Introduction**

32 **1.1 The environmental impacts of volcanic eruptions**

33 Explosive volcanic eruptions may lead to significant environmental impacts over a wide range
34 of spatial scales. Within the “proximal” area surrounding the eruption site, the physical impacts
35 may be immediate and devastating, due to blast waves, pyroclastic flows, lahars and volcanic
36 bombs (Annen and Wagner, 2003). The spatial extent of these hazards and impacts is limited by
37 the rapid cooling of volcanic ejecta and gases, and gravitational settling of the largest clasts. In
38 contrast, the more subtle chemical impacts of volcanic eruptions may be experienced over much
39 larger areas, as fine volcanic ash, aerosol and volcanic gas are dispersed widely by the prevailing
40 atmospheric motions.

41 Recent studies have offered new insights into the distal environmental impacts of volcanic
42 eruptions. These impacts range from the truly global scale (e.g., the climatic effects of sulphuric
43 acid aerosols; Robock et al., 2000), to the continental scale (e.g., localized ozone holes due to
44 halogen emissions; Millard et al., 2006), to the regional scale, just beyond the reach of the
45 proximal volcanic phenomena, where environmental impacts are dominated by the fallout of
46 volcanic ash (e.g. Inbar et al., 1995) and, potentially, acid sulphate deposition (Delfosse et al.,
47 2005; Gauci et al., 2008). On the regional scale, the most significant direct impact for human
48 health may be through the inhalation of fine volcanic ash. Particles with aerodynamic diameters
49 $d < 10 \mu\text{m}$ (the “thoracic” fraction) may enter the bronchi causing respiratory aggravation (e.g.,
50 triggering asthma attacks in susceptible people). The finest particles ($d < 4 \mu\text{m}$; the “respirable”
51 fraction) do not impact upon the bronchi but pass into the alveolar region of the lungs
52 potentially leading to respiratory illnesses. The two main concerns at present are crystalline
53 silica (causing silicosis and cancers; Baxter et al., 1999) and Fe-rich ash particles (causing lung
54 inflammation and the in-vivo production of hydroxyl radicals through the Fenton reaction;
55 Horwell et al., 2007). The impacts of fine ash on human populations are not limited to transient
56 ashfall events, but may persist for some months due to the resuspension of ash by winds or
57 vehicle movements (Hobbs et al., 1983; Inbar et al., 1995; Horwell et al., 2003). Further
58 mechanisms for toxicity may be related to compounds on the surface of the ash, which are
59 adsorbed as volcanic gases cool and condense in the eruption column (Witham et al., 2005;
60 Delmelle et al., 2007). The major control on the composition of the adsorbed material is likely
61 the volatility of elements at magmatic conditions; for example, the ratio As/Ca will be higher
62 within adsorbed material compared to the volcanic glass since As is more volatile than Ca (e.g.,
63 Aiuppa et al., 2003; Fulignati et al., 2006). Whilst the potential impacts of these adsorbed (and
64 thus, readily leachable) compounds on human well-being are unclear, the impacts on

65 agriculture are better understood. Animals may take up leachable elements through the
66 ingestion of ash, or by drinking water contaminated with ash. The greatest concerns in this
67 respect are fluoride and sulphur, which are toxic to livestock in high concentrations (e.g., Cronin
68 et al., 1998, 2003). Recent studies have shown that trace elements may also be transferred
69 rapidly from ash into the environment, giving measurably increased concentrations in waters
70 (e.g., Frogner et al., 2001; Jones and Gislason, 2008) and vegetation (e.g., Cronin et al., 1998;
71 Watt et al., 2007; Martin et al., 2009). The effects of trace element deposition on human, animal
72 and plant life are not well understood, although it seems likely that any detrimental impacts
73 would be subtle in comparison to the impacts of fluoride and sulphur, and the most important
74 effect may be the addition of nutrients. In any case, understanding these processes is crucial to
75 understanding of the environmental fate of volcanogenic trace elements.

76 **1.2 This work**

77 Chaitén volcano, situated in the southern Chilean Andes at 42.8 °S, began erupting explosively
78 on 2nd May 2008 (Lara, 2009). Prior to the eruption, the volcano comprised a rhyolitic lava
79 dome within a 2.5 km diameter caldera, both thought to result from the only previous volcanic
80 activity known from Chaitén, dated at 9370 ¹⁴C yr BP (Naranjo and Stern, 2004). The event led
81 to the evacuation of >5000 people from the surrounding area from settlements up to 75 km
82 distant, and estimates of the eruption volume (~0.2 km³ of tephra, Watt et al., 2009) suggest
83 that it was the largest eruption globally since that of Hudson, Chile, in 1991 (Scasso et al., 1994)
84 and the largest explosive rhyolitic eruption since Novarupta, Alaska, in 1912. The most
85 energetic activity took place in the initial days of eruption, culminating on 6th May, although
86 lower-level explosive activity is continuing at the time of writing (April, 2009). Ash was
87 deposited predominantly east of the volcano, reaching the Atlantic coast of Argentina.
88 Preliminary characterization of the ash (Horwell et al., 2008), and our own observations, show
89 that the ash was highly angular, being composed primarily of fragmented glass (obsidian),
90 accompanied by vesicular pumice and rare fragmented crystals. The ash is fine grained with
91 ~5–15 vol% and 10–30 vol% in the respirable and thoracic fractions (respectively; as a
92 cumulative percent within the $d < 1$ mm fraction). The ash had low crystalline silica content (~5
93 wt%) suggesting that the ash is more likely to represent a respiratory hazard in terms of in-
94 vivo hydroxyl formation or other factors than in terms of silicosis. The geography of the area
95 affected by ashfall is characterized by lakes and mountains to the west (>71°W; termed the
96 Patagonian Andes), with wide open plains (<71°W; termed Patagonian Plateau, or Steppe)
97 toward the Atlantic Ocean. Human populations are concentrated in large towns in the west (e.g.,
98 Bariloche and Esquel) and east (e.g., Trelew and Rawson) with smaller, remote settlements
99 scattered throughout the plains. The main industries within the Patagonian Andes region are

100 tourism and agriculture (e.g., cattle and sheep, fruit production), and these are most
101 concentrated around the major settlements.

102 In the first part of this study, we present time-resolved measurements of resuspended ash
103 concentrations in the air close to a busy road in the centre of Esquel (~100 km east of Chaitén;
104 $pop \sim 30,000$; ash thickness at the time of sampling $T \sim 10$ mm) on 4th June 2008
105 (approximately 1 month after first ashfall). Measurements were made using air filtration and
106 gravimetry. Several of the filters were also analysed using scanning electron microscopy (SEM)
107 to determine the grain size distribution of resuspended ash, and to assess whether the size
108 distribution of ash particles varies during the day. In parallel, traffic surveys were performed in
109 order to assess whether resuspended ash concentrations are related to traffic activity (i.e., the
110 number of vehicles passing the sampling site per unit time).

111 In the second part of this study, vegetation (Coirón grass; *Festuca pallescens*) and water (from
112 shallow, ephemeral lakes) samples were collected from across the ashfall-affected region in
113 June 2008 and analysed by inductively coupled plasma mass spectrometry (ICP-MS) for major
114 cations and trace elements, with waters also being analysed by ion chromatography (IC) for
115 major anions (e.g., F, Cl, SO_4^{2-}). These environmental samples were collected during fieldwork
116 that also included the measurement and mapping of the deposit in Argentina (as described in
117 Watt et al., 2009). Ash thickness was measured at all sites, allowing us to investigate the
118 relationships between ash thickness and element concentrations in environmental samples
119 approximately 1 month after first ashfall. An additional suite of Coirón grass samples were
120 collected in January 2009 (when ash thickness had decreased by 50-90% at all sampling sites,
121 with a similar absolute decrease in ash thickness across the region) and analysed similarly.

122 No major ashfall events were encountered during either sampling period (29th May – 11th June
123 2008 and 10th January - 20th January 2009), consistent with the lower activity of Chaitén during
124 these periods.

125 **2 Methodology**

126 **2.1 Field Methods**

127 The air in Esquel was sampled between 08:00 and 19:00 (local time) on 4th June 2008 for
128 particulates using filter packs pumped at rates of ~ 17 L min^{-1} . The filter pack consisted of a
129 particle filter (Millipore, 47 mm, AAWP pore size 0.8 μm) followed by an additional filter
130 (Whatman, 47 mm, type 41) to limit the flow rate. The filter pack was attached to a pole ~ 1.6 m
131 above the ground (i.e., the height of the mouth/nose of an average female adult; Freeman et al.,
132 1995) in the middle of a busy road through the town centre. The pump was powered by a 12V,

133 70 A hr car battery. For each sample a new filter was loaded and the initial flow-rate through
134 the filter pack was measured using a flow-meter. After sampling (typically 35-50 min), the filter
135 was removed, folded in half and placed in two layers of clean zip-lock polyethylene bags. The
136 flow-rate was not generally measured at the end of each sampling run to avoid disturbing
137 particles on the filter. In the one sample (04-01N) where the flow-rate was re-measured, the
138 flow-rate had fallen from 18 L min⁻¹ to 15 L min⁻¹ after 3 hr of sampling. This decrease in flow-
139 rate was mainly due to running down of the battery at the end of the day, rather than
140 accumulation of particles on the filter. The uncertainty on the determination of the flow-rate (F)
141 was therefore limited to the precision of the flow-meter (~10%). The total volume of air (V_A)
142 pumped through the filter pack was estimated from the flow-rate and sampling duration (t). On
143 the day of sampling, the wind speed remained low (< 1 m s⁻¹) throughout most of the day,
144 increasing to ~2-3 m s⁻¹ after 17:30 (local time). No rain, snow or fresh ashfall were observed at
145 any time during the day of sampling. The ash deposit on the ground was dry to the touch and
146 fine grained ($d < 100 \mu\text{m}$). Surveys of traffic movements past the measurement site were also
147 taken throughout the day (except early morning and late evening when light levels were low).
148 Traffic surveys were 7 minutes long and taken at ~30 minute intervals. Traffic was categorized
149 as pedestrians/cyclists, cars/vans or buses/trucks. Also recorded was the number of
150 pedestrians and cyclists wearing dust masks. Extended sampling runs were performed on 5th
151 and 7th June (during the daytime and nighttime periods, respectively). Details of air filtration
152 samples are given in Table 1.

153 Shallow ephemeral lakes were selected as a sampling target because: (1) these lakes can be
154 found throughout the sampling region, (2) the lake compositions are initially that of rainwater
155 (which has very low mineral content within the sampling area; Drago and Quiros, 1996), subject
156 to modification by the soil, and by evaporation/dilution, and (3) the lake compositions are
157 expected to show the most significant response to ashfall because of their high surface area /
158 volume ratio ("contamination potential"; Stewart et al., 2006). Samples were collected from the
159 lakes by triple-washing HNO₃-cleaned polypropylene tubes (50 ml capacity) with the sample,
160 followed by a final filling and sealing (gloves were worn at all times to avoid contamination). No
161 in-situ measurements (e.g., temperature, pH or turbidity) were made because of time
162 constraints. The samples were placed in two layers of clean zip-lock polyethylene bags for
163 transportation. Replicate samples were collected at several sites. On return to the UK, water
164 samples were frozen until the time of analysis. Further details relating to water samples are
165 given in Supplementary Table 1 (i.e., precise location, estimate of lake dimensions and ash
166 thickness)

167 Coirón grass was selected as a sampling target because: (1) this species grows abundantly and
 168 extensively within the region (Defossé et al., 1997), and (2) earlier work has shown that grass
 169 species uptake soluble elements rapidly from ash (Rye grass; Cronin et al., 1998). To our
 170 knowledge, Coirón grass has never previously been used as a bio-indicator for trace element
 171 concentrations within the environment so the results of the present study may have broader
 172 implications. The semi-arid vegetation limits agriculture to sheep husbandry in much of the
 173 region, and Coirón grass represent an important forage species (Bertiller and Coronato, 1994;
 174 Paruelo and Golluscio, 1994; Oliva et al., 2005). The grass is also a useful model species because
 175 of its ubiquity across the sampling region. The grass was sampled by collecting ~50 g of
 176 material at each site (typically from ~5 different plants), and placing the sample into a clean zip-
 177 lock polyethylene bag. On return to the UK, vegetation samples were brushed and washed with
 178 distilled deionised water (to remove any ash from the surface of the vegetation), and oven dried.
 179 Further details relating to vegetation samples are given in Supplementary Table 2 (i.e., precise
 180 location and ash thickness)

181 2.2 Analytical Methods

182 2.2.1 Gravimetry of particle filters

183 Particle filters were left for 1 hr in a clean hood to equilibrate with the ambient atmosphere
 184 (22°C, 40% RH). The filters were then weighed (M_{total}) using a balance with a precision of ± 0.05
 185 mg. The mean and standard deviation of the mass of 10 blank filters (also equilibrated for 1 hr
 186 at 22°C, 40% RH) was determined ($M_{\text{blank}} = 76 \pm 0.18$ mg). Thus, we estimate a total uncertainty
 187 of 0.19 mg on the mass of ash ($M = M_{\text{total}} - M_{\text{blank}}$) collected on each filter. These uncertainties are
 188 small in comparison to M (i.e., 0.3 – 5 mg). M can be related to the gravimetric particulate
 189 concentration (m_g ; equivalent to total suspended particles e.g., Monn et al., 1995) of the air by
 190 E1, where V_A is the volume of air sampled. We assume that other particles (e.g., soot, road and
 191 other dusts) make a negligible contribution to m_g . This assumption is reasonable given the
 192 significant amount of ash on the ground in Esquel, and is further justified by the results
 193 presented in Section 3.1.

$$194 \quad \text{E1} \quad m_g = \frac{M}{V_A}$$

195 2.2.2 SEM of particle filters

196 A subset of filter samples (04-01 D, 04-01 E, 04-01 H, 04-01 I and 04-01 M) were also analysed
 197 using scanning electron microscopy. The instrument (FEI XL30 ESEM FEG) operates at low
 198 vacuum (“environmental” SEM) such that pre-coating of the samples was not necessary. For

199 each sample, 10 - 12 areas (with dimensions of $\sim 200 \mu\text{m} \times 300 \mu\text{m}$) were imaged at 500×
 200 magnification (accelerating voltage =10 kV, beam current = 130 pA, spot size = 1.7 nm). The
 201 areas were selected randomly although all were away from the fold in the filter and all
 202 contained at least 50 particles. Particles were well separated on the filter and typically
 203 comprised $< 10 \%$ of the entire field of view. Particles were typically angular shards with low
 204 vesicularity. All imaged particles had diameters of less than $\sim 40 \mu\text{m}$. Particles with diameters of
 205 less than $\sim 4 \mu\text{m}$ could not be reliably discriminated from the filter.

206 An automated sequence was devised using image analysis software (*ImageJ*; Abramoff et al.,
 207 2004) to convert greyscale SEM images into a binary image with black particles on a white
 208 background. The sequence consisted of duplication of the SEM image and despeckle (to remove
 209 high frequency variations due to signal noise and the structure of the filter), followed by manual
 210 thresholding in order to give the best visual agreement between the original and processed
 211 images. The binary images were then analysed by *ImageJ* to calculate the equivalent diameter, d ,
 212 of each particle. Results obtained using the automated sequence were comparable to results
 213 obtained by manually tracing outlines of particles (for samples 04-01D and 04-01H) and all
 214 subsequent samples were analysed using only the automated sequence.

215 Grain size distributions are defined by a number density in $\log(d)$ space (E2A). The class-width
 216 was fixed to $d \log d = 1/15$. To obtain $N(d)$, the differential was divided by the area fraction of
 217 filter analysed (f) and the volume of air sampled (V_A). The value of f (~ 0.001) was calculated by
 218 dividing the total area imaged by the total area of particle deposition ($\sim 6 \times 10^{-4} \text{ m}^2$; estimated
 219 by inspection of the exposed filters). The uncertainties on $N(d)$ are mostly related to V_A so it is
 220 more convenient to use the ratio $N(d)/N(d_0)$ where $d_0 = 4 \mu\text{m}$. Grain size distributions may also
 221 be defined by a volume density (E2B). If all particles are spherical, $V(d)$ simplifies as indicated.
 222 The imaged particulate mass (m_i) can then be calculated from E2C, where ρ is the mass density
 223 of the particles, assumed to be $\sim 2500 \text{ kg m}^{-3}$ (i.e., rhyolitic glass). The uncertainties on m_i are
 224 potentially significant due to uncertainties on V_A , particle shape and mass density (due to
 225 composition and vesicularity). However, only the uncertainties on V_A are independent allowing
 226 m_i to be calculated self-consistently for a series of samples.

227 E2A
$$N(d) = \frac{1}{fV_A} \frac{dN}{d \log d}$$

228 E2B
$$V(d) = \frac{1}{fV_A} \frac{dV}{d \log d} \approx \frac{\pi}{6} d^3 N(d)$$

229 E2C
$$m_i = \rho V_{tot} \approx \rho \int_{-\infty}^{+\infty} V(d) d \log d$$

230 **2.2.3 ICP-MS and IC of water samples**

231 The water samples were thawed in a refrigerator over a period of 48 hr. A 20 ml syringe was
 232 used with a 0.2 μm minisart Sartorius particle filter which was triple washed with ultrapure
 233 Milli-Q+ water (18.2 $\text{m}\Omega$), followed by a single washing with the sample. The syringe was then
 234 used to withdraw 20 ml of sample, which was filtered into two clean centrifuge tubes for ICP-MS
 235 and IC analysis respectively. This procedure was repeated, using a new filter for each sample.

236 The concentrations of major cations and trace elements (in ppb by mass) were determined by
 237 ICP-MS analysis. To each sample, 145 μl of ultra-pure (69%) HNO_3 (VWR Analar grade, quartz
 238 distilled in house) was added prior to analysis. The instrument (Perkin Elmer Elan DRCII) was
 239 calibrated using surface water standard reference materials (USGSP35, USGST167, USGST143,
 240 LGC6019, SPS-SW2; supplied as liquids) and internal standards (10 μl of 1 ppm Rh, Re and In).
 241 Blank values were assessed with ICP-MS analysis of Milli-Q+ stored in the polypropylene tubes
 242 and filtered (as for the samples); these showed negligible concentrations of all reported
 243 elements in comparison to sample concentrations. Uncertainties are calculated as one standard
 244 deviation of the recovery with repeated analysis ($n=6$) of the five standards: As (8%), B (5%),
 245 Ba (3%), Ca (6%), Cd (9%), Co (3%), Cs (1%), Cu (10%), Fe (22%), K (4%), Mg (6%), Mn (3%),
 246 Mo (5%), Na (7%), Ni (3%), Rb (1%), Sr (4%), Tl (6%), Zn (8%). A few elements with certified
 247 concentrations in the reference materials are not reported because of poor recoveries or high
 248 concentrations in the blanks. The concentrations of major anions (in ppb by mass) were
 249 determined by IC analysis. The instrument (Metrohm IC 861) was calibrated using synthetic
 250 standard reference materials (FLUKA/89886). Uncertainties are calculated as one standard
 251 deviation of the recovery with repeated analysis ($n=3$) of the three standards: F^- (1.5%), Cl^-
 252 (1.6%) and SO_4^{2-} (2.0%). The concentration of the reported ions in the blanks was negligible.

253 **2.2.4 ICP-MS analysis of vegetation**

254 The vegetation samples were inspected and ~10-15 healthy green blades from each sample
 255 were passed for further processing. The blades were ground using a rotary mill for 25 min at
 256 250 rpm in sealed agate vessels. The ground material was then sieved at 1 mm and the fine
 257 powder was retained. Approximately 50 mg of each powder was weighed into clean vials and
 258 digested with 3 ml ultra-pure (69%) HNO_3 (VWR Analar grade, quart distilled in house) for 48
 259 hr at 110°C, with 2 ml H_2O_2 (VWR Analar grade) added after the first 24 hr. After 48 hr, the
 260 digests were triple-washed with Milli-Q+ into a centrifuge tube and made up to 12 ml. The

261 digests were then centrifuged at 5000 rpm for 5 min to remove any remnant solids. Using a
262 clean pipette, 5 ml of the digest was transferred to a second centrifuge tube and made up to 14
263 ml with Milli-Q+.

264 The concentrations of elements in each sample (in ppm by mass) were determined by ICP-MS
265 analysis. For June 2008 samples, the instrument (Perkin Elmer Elan DRCII) was calibrated using
266 plant standard reference materials (NIST1573a, NIST1570a, NIST1547, NIST1567a and
267 BCRCRM281; supplied as powders) and internal standards (0.1 ml of 1 ppm Rh, Re and In).
268 Uncertainties were calculated as one standard deviation of the element concentrations with
269 repeated analysis ($n=9$) of the standards. For January 2009 samples, the instrument (Thermo-
270 Finnegan Element 2 HR-ICP-MS) was calibrated using plant standard reference materials (NIST
271 1573a, NIST1547 and BCRCRM281) and internal standard (0.1 ml of 0.5 ppm In). Uncertainties
272 were calculated as one standard deviation of the element concentrations with repeated analysis
273 ($n=4$) of sample A11. Uncertainties for June 2008 and January 2009 samples were, respectively:
274 Al (10%, 16%), As (8%, 24%), B (14%, n. a.), Ba (9%, 17%), Ca (9%, 9%), Cd (13%, 20%), Co
275 (7%, 17%), Cu (9%, 14%), Fe (8%, 16%), K (8%, 12%), Mg (7%, 12%), Mn (8%, 11%), Mo (15%,
276 12%), Na (14%, 13%), Ni (6%, n. a.), Pb (8%, 6%), Rb (8%, 11%), Sr (9%, 9%), Tl (11%, 17%)
277 and Zn (7%, n. a.). A few elements with certified concentrations in the reference materials are
278 not reported because of non-linearity in the calibration or high concentrations in the blanks (i.e.,
279 solutions prepared identically but without the addition of powder). The uncertainties were
280 typically 5-10% larger for the January 2009 samples, perhaps reflecting minor heterogeneities
281 within the powdered samples (i.e., A11).

282 **3 Results and Discussion**

283 **3.1 Airborne particulate matter**

284 Figure 2 shows the variation in gravimetric particulate mass (m_g) throughout the day in Esquel
285 on 4th June 2008, along with uncertainty bars (for one standard deviation; calculated using
286 standard propagation of error formula and E1). Also shown are results from the traffic survey,
287 presented as vehicle or pedestrian movements per minute, with some scaling for clarity. The
288 gravimetric particulate mass peaks at ~10.30 and ~18.30, with the later peak being larger.
289 Traffic survey results show similarly timed peaks indicating that the variation in m_g can be
290 explained (at least partially) by variations in traffic activity. The large peak in m_g at ~18.30
291 reflects both high traffic activity and increased wind speed (i.e., $< 1 \text{ ms}^{-1}$ before 17:30 and ~2-3
292 ms^{-1} after 17:30). The peaks in m_g are also consistent with our field observations of when the air
293 was visibly most dusty. The variations in traffic movements of small (i.e., cars/vans) and large
294 (i.e., buses/trucks) motor vehicles are similar. Hence, it is not possible to assess which type of

295 vehicle had the greatest effect on m_g ; during the course of sampling, all types of motor vehicles
296 were observed to generate visible plumes of dust. These results show that the variations in m_g
297 correspond well with visual observations, such as traffic activity, wind strength and the
298 dustiness of the air. The proportional relationship between m_g and traffic movements was also
299 found on Montserrat at much lower traffic activity (Horwell et al., 2003). The gravimetric
300 particulate mass cannot be calculated for samples taken on 5th and 7th June since V_A is not
301 known; however, the finding that M was $\sim 4\times$ greater over the day-time period (5th June) than
302 the night-time period (7th June) (Table 1) is consistent with our findings that m_g is controlled
303 primarily by traffic activity (however, the wind speed was not monitored during these periods
304 so we cannot exclude greater wind speeds on 5th June). The peaks in m_g also correspond with
305 peaks in pedestrian traffic, meaning that the exposure of humans to resuspended ash was
306 maximised. The usage of dust masks varied throughout the day (0 – 30%; shown on Figure 2)
307 but this variation did not follow m_g (although the data is noisy). At the time of sampling it was
308 notable that the main factor influencing the variation in the usage of dust masks appeared to be
309 the daily routines and attitudes toward mask usage of different groups of people. For example,
310 the usage was particularly poor amongst children; this may relate to the limited availability of
311 small masks, or be because of negative attitudes toward wearing the masks. In contrast, almost
312 all cyclists were observed to be wearing dust masks, potentially reflecting the exposure of
313 cyclists to the more concentrated plumes of ash (i.e., $>m_g$) created behind vehicles. People
314 sweeping ash from the pavements were also observed to be wearing dust masks. These
315 observations perhaps suggest a good awareness of the hazard associated with short exposures
316 to very high concentrations of ash (i.e., $>m_g$), but not of extended exposures to more moderate
317 concentrations (i.e., m_g).

318 Figure 3A shows the scaled number density, $N(d)/N(d_0)$ where $d_0 = 4\ \mu\text{m}$, for the particle filter
319 samples analysed using SEM. In addition to removing any uncertainties on f and V_A , scaling also
320 enables comparison of the size distribution of resuspended ash with that of the “source” ash
321 deposit, which was collected from the ground and analysed by laser diffraction (Watt et al.,
322 2009). Figure 3A shows that the size distributions for resuspended ash are steepened, to
323 varying extents, relative to the size distribution of the ash deposit, with the resuspended ash
324 having significantly fewer of the largest particles. The values of $N(d_0 = 4\ \mu\text{m})$ are approximately
325 comparable between samples (Table 2) indicating that the concentrations of the finest particles
326 do not vary significantly throughout the day, despite large variations elsewhere in the particle
327 size spectrum. By extrapolation, we may further predict that the scaled number density for the
328 resuspended ash samples and the ash deposit are approximately equivalent for $d < 4\ \mu\text{m}$. The
329 striking similarity between pairs of samples taken close in time adds confidence in the size

330 distributions. Figure 3B shows the scaled volume density, $V(d)/V(d_0)$ where $d_0 = 4 \mu\text{m}$, for the
331 particle filter samples analysed using SEM. All samples except 04-01M show that volume is
332 concentrated within the $d < 10 \mu\text{m}$ range. For 04-01M, V_{tot} may be underestimated due to the
333 presence of large particles which were not imaged. However, $V(d)$ for the ash deposit has a
334 maximum at $\sim 75 \mu\text{m}$ (Watt et al., 2009) so the potential for further increase in $V(d)$ above ~ 40
335 μm for 04-01M is limited.

336 These results suggest that variation in m_g (Figure 2) is mostly due to variation in the shape of
337 the size distribution, rather than a uniform scaling of $N(d)$ (which would be the simplest
338 explanation for why m_g is proportional to the number of traffic movements). In rural settings,
339 the main process by which small particles are resuspended is through saltation bombardment
340 (Shao et al., 1993; Wang et al., 2008). In this process, large particles ($d > 100 \mu\text{m}$) are
341 temporarily lofted by wind-driven aerodynamics. The maximum height reached by the saltating
342 particles is on the order of $\sim 0.01\text{-}0.10 \text{ m}$ (Wang et al., 2006). When the saltating particles return
343 to the ground, the impact leads to an efficient transfer of kinetic energy to small particles ($d < 20$
344 μm) which are then resuspended. In this setting, the direct re-suspension of small particles is a
345 minor process due to cohesive forces between particles (although these are disrupted by
346 saltation bombardment). However, the situation in urban environments (e.g., Esquel) may well
347 be different as an important factor affecting m_g (at least, when the wind speed is low) appears to
348 be the number of traffic movements. The most straightforward mechanism that explains our
349 results is direct resuspension of ash particles by the transfer of kinetic energy from the wheels
350 of the vehicle to particles on the ground (leading to disruption of intra-particle cohesive forces).
351 The size distribution of suspended particles is steepened over time due to differences in the
352 settling velocities between particles of different sizes; the settling velocities of ash particles with
353 $d = 1, 10$ and $100 \mu\text{m}$ are $\sim 100 \mu\text{m s}^{-1}$, 10 mm s^{-1} and 1 m s^{-1} (respectively; Hinds et al., 1999).
354 Hence, an increased number of resuspension events (e.g., increased traffic activity) would
355 inhibit steepening by sustaining the concentration of the largest particles. The relative
356 magnitudes of the settling velocities are such that the concentrations of large particles will
357 respond very quickly to changes in traffic movements, whilst the concentrations of small
358 particles are not so variable. Hence, $N(d_0 = 4 \mu\text{m})$ would not be expected to correlate well with
359 traffic movements at the time of sampling since $N(d_0 = 4 \mu\text{m})$ is also affected by traffic
360 movements over the preceding period of several hours (e.g., the relatively high value for 04-01
361 H may reflect carry-over from the morning peak in traffic activity).

362 Results from experimental and theoretical studies suggest that $N(d)$ would decrease with height
363 as a power-law or exponential function (e.g., Wang et al., 2008 and references therein) and that
364 the rate of decrease may be greater for large d . Close to the ground (but above the saltation

365 layer), the size distribution may be enriched in the largest particles relative to the ash deposit
 366 because the smallest particles are more significantly distributed through the vertical column
 367 (Horwell et al., 2003). Although the variation in $N(d)$ with height was not investigated in the
 368 present study, we note that the calculated parameters (e.g., m_g , $N(d)$, $V(d)$, V_{tot}) are valid only for
 369 ~ 1.6 m above the ground, and may be significantly greater at lower heights (e.g., at 0.7 m; the
 370 height of the mouth/nose of average 2 yr old). This factor (along with our observations of mask
 371 usage) suggests that the distribution of small masks suitable for children should be a priority for
 372 hazard mitigation.

373 Figure 4 shows a comparison of the gravimetric particle mass, m_g , and the imaged particle mass,
 374 m_i (calculated using E2B-C; Table 1). The uncertainty on m_i was estimated as 10% based on
 375 uncertainties on V_A . As discussed, additional uncertainties on m_i are not independent for a series
 376 of samples. The gravimetric and imaged particle mass are shown to be proportional, verifying
 377 our methodology. Furthermore, the gradient has a value (i.e., 1.5) close to unity indicating that
 378 over- or under-estimation of m_i was minor (e.g., if we had assumed particles were cubic rather
 379 than spherical, the gradient would be 0.75). Our comparison reveals a non-zero intercept of \sim
 380 $410 \mu\text{g m}^{-3}$, which we interpret as the mass of particles with $d < 4 \mu\text{m}$ since particles in this size
 381 range were not quantified through imaging. Note that if m_i for 04-01 M was indeed
 382 underestimated the effect would be to increase the intercept on Figure 4. The linear relationship
 383 between m_i and m_g indicates that the $< 4 \mu\text{m}$ fraction makes a similar contribution to m_g in all
 384 samples (within uncertainty), consistent with earlier results for $N(d_0 = 4 \mu\text{m})$. In order to
 385 estimate the aerodynamic diameters (d_A) of particles, we assume that all particles are spherical
 386 with mass density $\rho = 2500 \text{ kg m}^{-3}$, giving d_A equal to $\sim 1.6 \times$ the equivalent diameter, d (i.e., $d_A = d$
 387 $\sqrt{(\rho/1000)}$; Hinds, 1999). Thus, $\text{PM}_{2.5}$, PM_4 and PM_{10} correspond to the mass of particles with
 388 equivalent diameters of $< 1.6 \mu\text{m}$, $< 2.5 \mu\text{m}$ and $< 6.3 \mu\text{m}$, respectively. Integration of the scaled
 389 volume density for the ash deposit suggests that of the total volume in the $< 4 \mu\text{m}$ fraction (i.e.,
 390 $\sim 410 \mu\text{g m}^{-3}$), $\sim 70\%$ lies in particles with $d < 2.5 \mu\text{m}$, and $\sim 40\%$ in particles with $d < 1.6 \mu\text{m}$.
 391 Integration of $V(d)$ for the resuspended ash suggests that of the total volume in the $> 4 \mu\text{m}$
 392 fraction, a proportion corresponding to $\sim 500 \mu\text{g m}^{-3}$ (for all of the analysed samples) lies in
 393 particles with $4 \mu\text{m} < d < 6.3 \mu\text{m}$. If we make the approximation that the scaled volume density
 394 for resuspended ash and the ash deposit are equivalent for $d < 4 \mu\text{m}$, then we estimate $\text{PM}_{2.5} \sim$
 395 $200 \mu\text{g m}^{-3}$, $\text{PM}_4 \sim 300 \mu\text{g m}^{-3}$ and $\text{PM}_{10} \sim 1000 \mu\text{g m}^{-3}$, with little variation throughout the day
 396 (at least, within uncertainty and averaged over ~ 1 hr long sampling periods). These results
 397 suggest what whilst m_g can be estimated by visual observations, $\text{PM}_{2.5}$, PM_4 and PM_{10} (which are
 398 better proxies for respiratory hazard than m_g) are high even when the traffic activity and wind
 399 speed are relatively low. Our estimates of PM_{10} and $\text{PM}_{2.5}$ significantly exceed WHO air quality

400 guidelines (24 hr mean; PM_{10} : $50 \mu\text{g m}^{-3}$ and $PM_{2.5}$: $25 \mu\text{g m}^{-3}$) indicating that concentrations of
401 resuspended ash in Esquel reached levels which may have been detrimental to human health.
402 Our estimate of PM_{10} is comparable to estimates of PM_{10} in other human settlements affected by
403 volcanic ashfall (e.g., on Montserrat; Baxter et al., 1999). We observed that traces of the ash
404 remained in Esquel in January 2009, although the air was no longer visibly dusty by this time.
405 Hence, we suggest that the duration of exposure to exceedance levels of PM_{10} and $PM_{2.5}$ is likely
406 to have been on the order of several months (after the first ashfall in May 2008).

407 A preliminary study of Chaitén ash (Horwell et al., 2008) found low crystalline silica content,
408 suggesting a reduced health hazard relative to ash from other eruptions (e.g., Soufriere Hills,
409 Montserrat). However, these earlier results must be taken in the context of the high
410 concentrations of PM_{10} , PM_4 and $PM_{2.5}$ reported here, and also the possibility of further
411 unknown mechanisms for toxicity (either related to the specific chemical and mineralogical
412 composition of the ash, or to $N(d)$ more broadly). Whilst no relationship was found between
413 dust mask usage and m_g , it is intuitive to suggest that amongst the people who wear the masks
414 only occasionally, they are more likely to use them when traffic activity and wind speed are
415 high, and the air is visibly dustier. Our results suggest an improved mitigation strategy for this
416 group of people would be to make decisions based primarily upon the expected duration of
417 exposure, rather than attempt hazard assessment based upon visual observations as it is the
418 finer and thus less visible dust that presents the greatest health hazard.

419 3.2 Waters

420 Previous studies (e.g., Pedrozo et al., 1993; Diaz et al., 2007) have shown that lake waters vary
421 between highly dilute and slightly acidic in the Patagonian Andes (e.g., mean Ca ~ 5 mg/L, Mg \sim
422 1 mg/L, Na ~ 2 mg/L, pH ~ 6 -7; Diaz et al., 2007) to highly saline and slightly alkaline on the
423 Patagonian Plateau (e.g., mean Ca ~ 50 mg/L, Mg ~ 30 mg/L, Na ~ 320 mg/L, pH ~ 7 -8; Diaz et
424 al., 2007). This difference is primarily due to decreasing orographic rainfall away from the
425 Andes; from ~ 2500 mm/yr at the Argentina/Chile border to ~ 150 mm/yr on the Patagonian
426 Plateau. This gradient in rainfall runs parallel to the gradient in ash thickness (Figure 1). To
427 assess the environmental effects of ashfall, we must account for differences in dilution between
428 the lakes. In order for an element to be used as a proxy for lake dilution: (1) the amount of
429 element transferred from the soil to the lake must not vary significantly over the sampling area
430 or depend on lake depth (i.e., the lake must not be saturated), (2) the element must not be
431 supplied in significant amounts by the ash, and (3) the concentration of the element in rainfall
432 must be low. A suitable candidate is Ca since this element is soluble in acidic waters but
433 involatile at magmatic conditions (Aiuppa et al., 2003). Whilst it is known that soil compositions

434 trend comparably to lake compositions and become more saline on the Plateau (Diaz et al.,
435 2007), this spatial variability has not been studied in detail. Thus, we simply interpret R^2 (i.e.,
436 the square of the correlation coefficient) of X/Ca against ash thickness to be the fraction of
437 variability which can be explained by the ash thickness, with the remainder of the variability
438 being due to other unexplained factors.

439 Figure 5A shows correlation coefficients (R) between element concentrations (normalised to
440 Ca) in waters (X/Ca) and ash thickness at each site (T/mm). Correlations are calculated for the
441 entire dataset ($n=11$) and for $T > 2$ mm ($n=9$). Replicates taken in the field are treated as
442 independent samples. Results for element concentrations (ppb by mass) are tabulated in
443 Supplementary Table 1. Rb, Tl, K, As, Cs, Mn, Zn, Cl, Cd, Zn, Cl, Fe, B, Cu and Ni show strong
444 positive correlations ($R^2 > 0.75$) in samples with $T > 2$ mm. The ordering of elements is not
445 significant due to differences in the analytical uncertainties. The strength of these correlations
446 implies that the ambient soil and lake compositions (at least in the Patagonian Andes, where $T >$
447 2 mm) were a minor contribution (i.e., $\sim 1-R^2$) to the variability in X/Ca for a large number of
448 elements. This result is consistent with all of the sampled lakes (at $T > 2$ mm) being slightly
449 acidic prior to ashfall, ensuring that many elements adsorbed on the ash were soluble in the
450 receiving waters. In more alkaline lakes, a greater than linear relationship between X/Ca and T
451 might be expected as the solubility of elements on the ash would also increase with T due to
452 acidification of the waters (e.g., Stewart et al., 2006). The weak negative correlations (e.g., Mg)
453 are spurious, e.g., Diaz et al. (2007) show that Mg/Ca in lakes vary from 0.2 in the Patagonian
454 Andes to 0.6 in the Patagonian Plateau (consistent with our results). In spite of the uncertainties
455 on ambient element concentrations, the strong positive correlations between X/Ca for certain
456 elements (Rb, Tl, K, As, Cs, Mn, Zn, Cl, Cd, Zn, Cl, Fe, B, Cu and Ni) and ash thickness indicate that
457 these correlations are real, and reflect a causal relationship between ash deposition and water
458 composition. Correlations between raw element concentrations and ash thickness are also
459 strong and positive, though weaker than when the normalisation is applied. These results are
460 consistent with those shown for deeper, permanent lakes (where $T \sim 50$ cm) in Chile following
461 the eruption of Hudson in 1991 (Weaire and Manly, 1996). The samples of Weaire and Manly
462 (1996) were taken ~ 18 months after the onset of the eruption. It therefore seems likely that the
463 presumed compositional changes in deeper, permanent lakes in the Andean regions affected by
464 Chaitén's ashfall may persist for at least several months.

465 We have assumed that the correlations between X/Ca and T are due to the addition of elements
466 into the environment by ashfall. An additional possibility is that ashfall resulted in acidification
467 of the soils, leading to the increased solubility of certain elements (e.g., As, Cd, Cu, Pb) and the
468 release of elements from the soil into the lakes. We suggest that this is not the general case for

469 the samples considered here because: (1) most of the lakes were sampled from the Patagonian
470 Andes region where the soils are already quite acidic (Diaz et al., 2007), and (2) the linear
471 correlations between X/Ca and T would require that the solubility must be proportional to $[H^+]$
472 rather than powers thereof, which is the more general case. Future studies of the response of
473 soils to acidification may shed further light on this uncertainty (in particular, whether a
474 sufficient enough amount of these elements can be released from the soil in order to explain the
475 observations presented here).

476 Accepting for the moment that the measurements reflect the addition of elements into the
477 environment, we may estimate the composition of soluble material on the surface of the ash. For
478 two elements (i.e., Y and Z) which both show a strong positive correlation between normalised
479 element concentrations and ash thickness, the gradient of (Y/Ca) against (Z/Ca) corresponds to
480 Y/Z of the soluble material on the surface of the ash (Table 3; also Figure 5B for $Y = K, Tl, As$ and
481 $Z = Rb$). Furthermore, normalisation to Y/Z in the bulk ash (Watt et al., 2009) yields the
482 following enrichment factors (where $Z = Rb$): As (14), Zn (10), K (2.4), Cu (1.5), Rb (1), Mn
483 (0.28) and Fe (0.019). Note that as these enrichment factors are calculated relative to the bulk
484 ash, differences between elements are smaller than if calculated relative to the lavas (or glass).
485 In any case, these estimates are generally comparable with typical enrichment factors in
486 volcanic plumes (e.g., Zoller et al., 1993; Taran et al., 1995; Aiuppa et al., 2003; Fulignati et al.,
487 2006), with greater enrichments found for the chalcophile elements (e.g., As, Zn) than the
488 lithophile elements (e.g., Mn, Fe). Of course, it is unlikely that the precise order of enrichments
489 would be identical between volcanoes since enrichments will depend on factors such as the
490 oxygen fugacity and magma temperature (which determine volatility). These results add further
491 weight to our earlier conclusion that the variability in element concentrations in ephemeral
492 lakes was due to the deposition of volcanic ash.

493 3.3 Vegetation

494 The mechanisms of element uptake in vegetation are complex (Kabata-Pendias, 1999),
495 occurring through the roots, stem and leaves. Element concentrations in vegetation are an
496 integration of air, soil and rainwater compositions over an unknown period of time. However,
497 the precise details of uptake need not be fully known to use element concentrations in
498 vegetation as an indicator of environmental conditions (e.g., Rossini-Oliva and Valdés, 2003;
499 Kapusta et al., 2006; Martin et al., 2009 and others). In the present context, it is anticipated that
500 elements are dissolved from the ash into a solution, allowing transfer of elements to Coirón
501 grass.

502 Figure 6A shows correlation coefficients (R) between the concentrations of elements in Coirón
503 grass sampled in June 2008 and ash thickness at each site (T/mm). Correlations are calculated
504 for the entire dataset ($n=40$), for $T > 2$ mm ($n=10$) and for $T > 3$ mm ($n=7$). Figure 6B shows
505 element concentrations against ash thickness for B, Pb and Cd. Results for other elements are
506 tabulated in Supplementary Table 3A-B. Figure 6A shows that Coirón grass is an excellent bio-
507 indicator for a range of elements, with strong positive correlations ($R^2 > 0.7$) for B, Pb and Cd and
508 reasonable correlations ($R^2 > 0.5$) for Zn, Tl, Cu, Ni and Co. With the exception of Co, all these
509 elements were also found to correlate with ash thickness in lake waters. As $R \sim 1$ for B, Pb and
510 Cd, the relationship between element concentrations in vegetation and ash thickness is
511 approximately linear (at least for $2 \text{ mm} < T < 11 \text{ mm}$). The result that R is comparable for $T > 2$
512 mm and $T > 3$ mm confirms that the application of an arbitrary cut-off ($T = 2$ mm) does not
513 influence R significantly. The hypothesis that B, Pb and Cd vary in Coirón grass due to factors
514 unrelated to ash thickness is not supported by the data because of the strength of correlation,
515 and because the increase in B in the westernmost samples (i.e., at increased T) runs counter to
516 the ambient environmental trend (i.e., decreasing element concentrations in soils to the west;
517 Diaz et al., 2007). The correlations weaken if an increased range of ash thickness is considered
518 (as found for lake compositions), because the area from which samples are drawn becomes
519 more extensive and more heterogeneous in terms of other factors which may affect the
520 composition of vegetation (e.g., salinity of the soil). The correlations between ash thickness and
521 element concentrations in vegetation are generally weaker than those found for lakes. This
522 result reflects the increased analytical uncertainties (in the analysis of vegetation; Section 2)
523 and potentially non-linear relationships between element concentrations in vegetation and the
524 environment due to biological fractionation (e.g., Kabata-Pendias, 1999). Figure 6B indicates
525 that the element concentrations for $T > 2$ mm are within the normal range produced by other
526 sources of variability. It therefore seems unlikely that the increased element concentrations
527 would have any negative impacts on Coirón grass physiology, unless related to the short
528 timescales over which the element concentrations are established. However, this vegetation was
529 sampled during the Austral winter (outside the growth season) so the uptake of elements from
530 the ash may have been even more significant if ashfall occurred during the summer months. The
531 increase in element concentrations below $T \sim 2$ mm (Figure 6B) is consistent with increasing
532 salinity of the soils on the Plateau (Diaz et al., 2007).

533 Figure 6C shows correlation coefficients (R) between the concentrations of elements in Coirón
534 grass sampled in January 2009 and ash thickness (T/mm) as measured at the same location in
535 June 2008. The range of "original" ash thicknesses from which the January 2009 samples were
536 collected was $2.5 < T/\text{mm} < 6$, with thicknesses in January 2009 being $0.2 < T/\text{mm} < 3$. To allow

537 for fairest comparison between the two data sets, R for the June 2008 samples was re-calculated
538 for the range $2 < T/\text{mm} < 6$ and shown on Figure 6C. Quite clearly, the correlations are much
539 weaker in January 2009, indicating that elevated element concentrations in Coirón grass
540 persisted for <8 months. The mechanisms by which elements are transferred from vegetation to
541 the environment are poorly constrained though it seems likely that shedding and replacement
542 (in the next growth season) and leaching by rainfall may be important processes. In any case,
543 these results add weight to our conclusion that the variability in element concentrations in June
544 2008 was due to ashfall, rather than background variability.

545 An earlier study showed that As, Cd, Cu, Mo, Tl, Zn, Rb, Cs and K became rapidly incorporated
546 into sweet chestnut leaves (*Castanea sativa*; Martin et al., 2009) following a minor eruption of
547 Mt Etna, Sicily during the growth season. In the present study, we offer evidence for the rapid
548 incorporation of B, Pb, Cd, Zn, Tl, Cu and Ni into Coirón grass following the eruption of Chaitén.
549 Despite differences in length scales (i.e., < 30 km from the summit of Mt Etna but > 100 km from
550 Chaitén) and processes of deposition (i.e., heavy ashfall from Chaitén but light ashfall and plume
551 fumigation at Mt Etna), significant changes in the concentrations of Cd, Zn, Tl and Cu in
552 vegetation were found in both studies. These elements correspond well to the elements that are
553 volatile at magmatic conditions (Aiuppa et al., 2003; Fulignati et al., 2006). No evidence for
554 increases in the concentration of non-volatile elements in vegetation due to volcanic eruptions
555 was found in either study. However, increases in the concentration of the non-volatile elements
556 may result if the soil is particularly nutrient poor; for example, Cronin et al., (1998) show uptake
557 of Ca and Mg from ash into rye-grass grown in silica sand. Together, these studies suggest that
558 volcanic eruptions typically lead to an increase in the concentration of trace volatile elements in
559 vegetation over wide areas. Whilst the potential physiological effects of these compositional
560 changes are not yet known, this result may offer new insights into the environmental fate of
561 volcanogenic trace elements.

562 As discussed previously, the use of correlation coefficients in this study is limited in that the
563 complexity of the dataset is reduced to a single explanatory variable. Additionally, the amount of
564 an element deposited (per unit area) is dependent upon the total surface area (A_{tot}) of the ash
565 deposited, rather than the ash thickness. Whilst the isopleth map of Watt et al., (2009) indicates
566 that the size distribution of ash was stable toward the south-east (i.e., $A_{\text{tot}} \propto T$), there was an
567 increase in modal grain size toward the secondary maximum (Figure 1) in the north-east (from
568 where a few of the January 2009 samples were collected). However, allowing for this coarsening
569 does not improve the strength of the correlations in January 2009 and so does not affect our
570 conclusions.

571 4 Conclusions

572 The 2008 eruption of Chaitén had significant environmental effects over a wide geographical
573 area. In this study, we combine results from air, water and vegetation sampling to offer insights
574 into the environmental effects of ashfall in Argentina (between latitudes of 42°S and 46°S).

575 Results from air filtration suggest that in the urban environment the total mass of resuspended
576 volcanic ash (i.e., m_g) is well-correlated with traffic activity. However, it appears that increases
577 in the total mass are primarily related to increased amounts of the largest ash particles, rather
578 than increased amounts of the fine ash particles (i.e., $d < 4 \mu\text{m}$). We suggest that this observation
579 reflects differences in the gravitational settling velocities of large and small particles, such that
580 high concentrations of the largest particles are only found when the frequency of resuspension
581 events is large (i.e., during periods of heavy traffic). The corollary of this result is that the human
582 hazard associated with the inhalation of fine ash particles does not vary throughout the day.
583 Hence, the most important factor determining the hazard to individuals is the duration of their
584 exposure. Our estimates of $\text{PM}_{2.5}$ ($200 \mu\text{g m}^{-3}$) and PM_{10} ($1000 \mu\text{g m}^{-3}$) significantly exceed WHO
585 air quality guidelines. The duration of exposure to exceedance levels of $\text{PM}_{2.5}$ and PM_{10} was
586 likely to have been on the order of several months.

587 Results from water and vegetation sampling indicate that ashfall resulted in significant
588 compositional changes in ephemeral lakes and Coirón grass. Several elements (i.e., B, Cd, Zn, Tl,
589 Cu and Ni) show strong positive linear correlations between concentrations and ash thickness
590 (for $T > 2 \text{ mm}$) in both sets of results. These results suggest that the eruption of Chaitén led to
591 significant changes in the concentrations of trace volatile elements in the environment over a
592 wide area. No evidence was found for increases in the concentration of non-volatile elements in
593 either vegetation or water. Analysis of vegetation samples collected in January 2009 indicates
594 that these elevated element concentrations in Coirón grass persisted for < 8 months. These
595 results offer insights into the environmental fate of volatile trace elements emitted during
596 volcanic eruptions.

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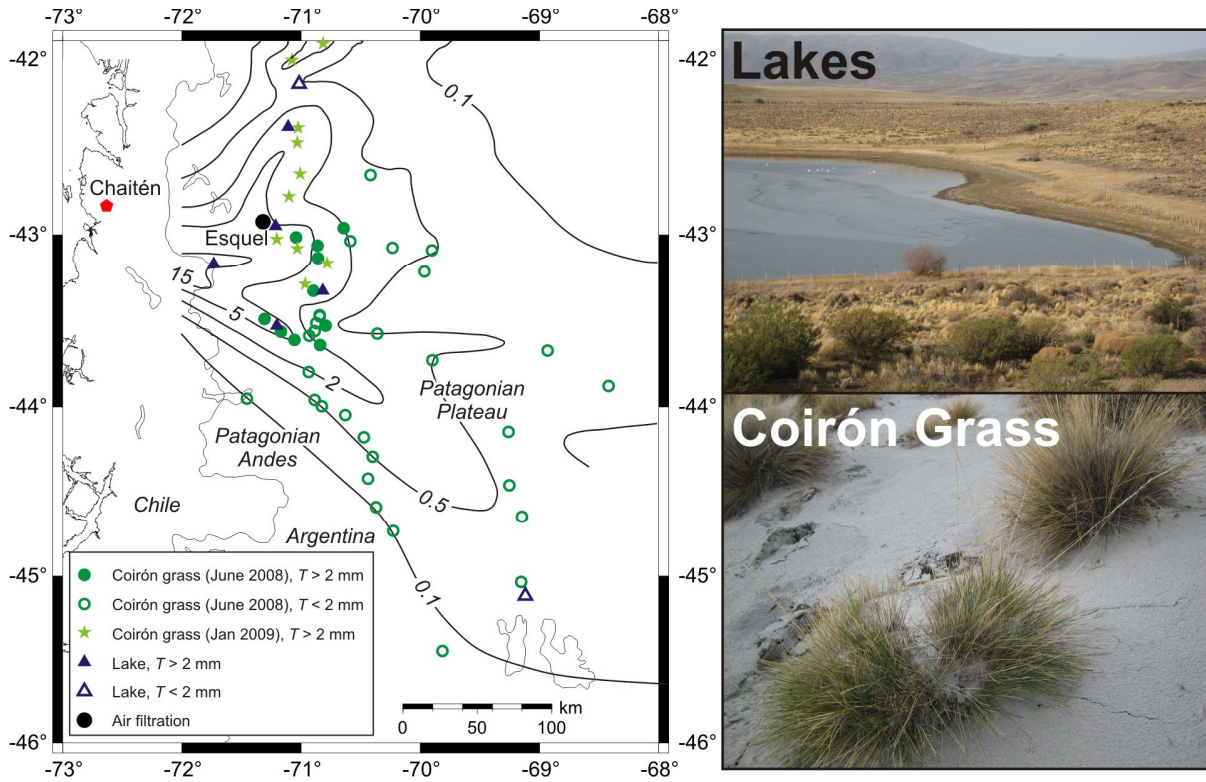
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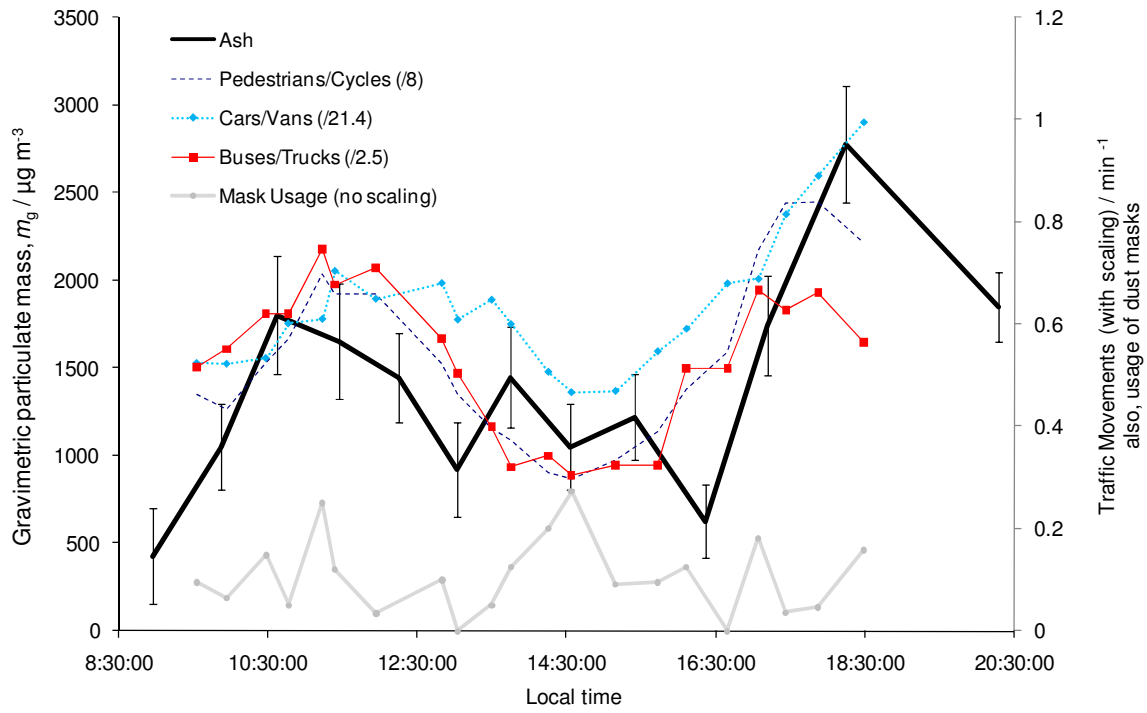
607 Figure 1 Map of the sampling area with approximate sampling locations. Locations are
 608 discriminated according to sample type (air filtration, lake or Coirón grass) and
 609 ash thickness (Watt et al., 2009).



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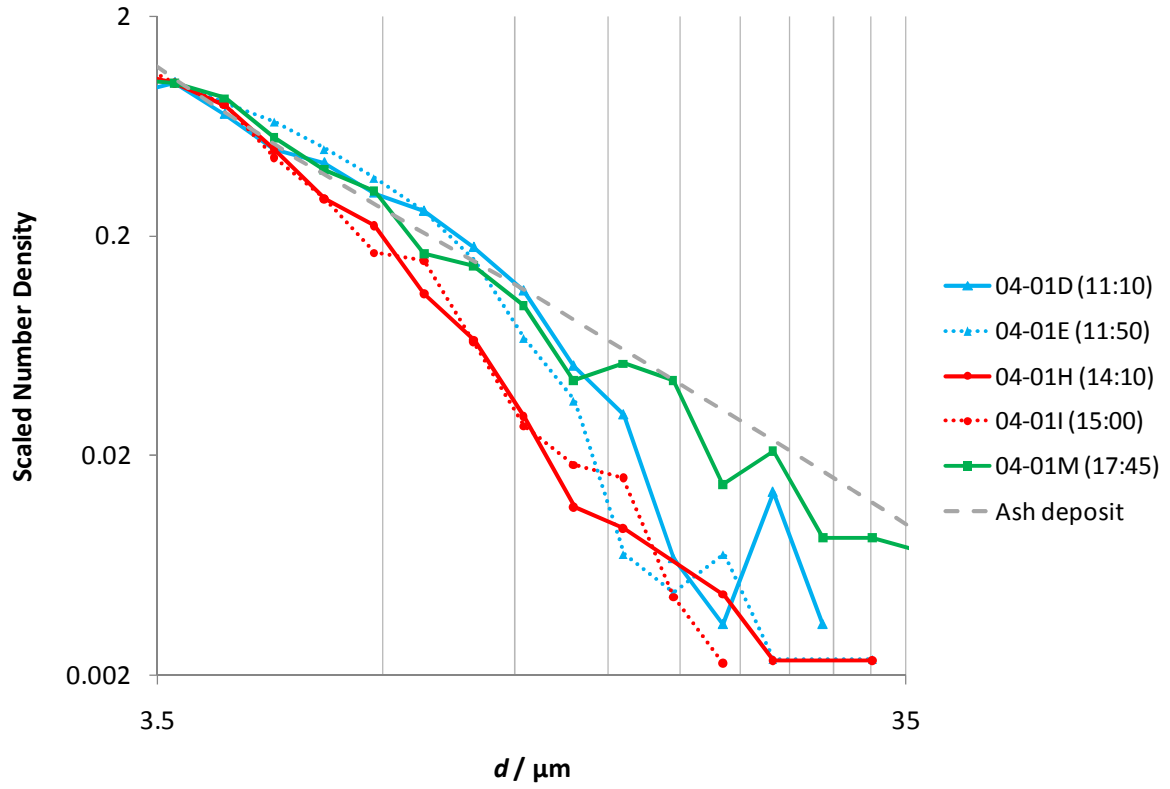
612 Figure 2 Gravimetric particulate mass (m_g) throughout the day. Also shown are traffic
 613 survey results (with scaling) with dust mask usage on the same axis.



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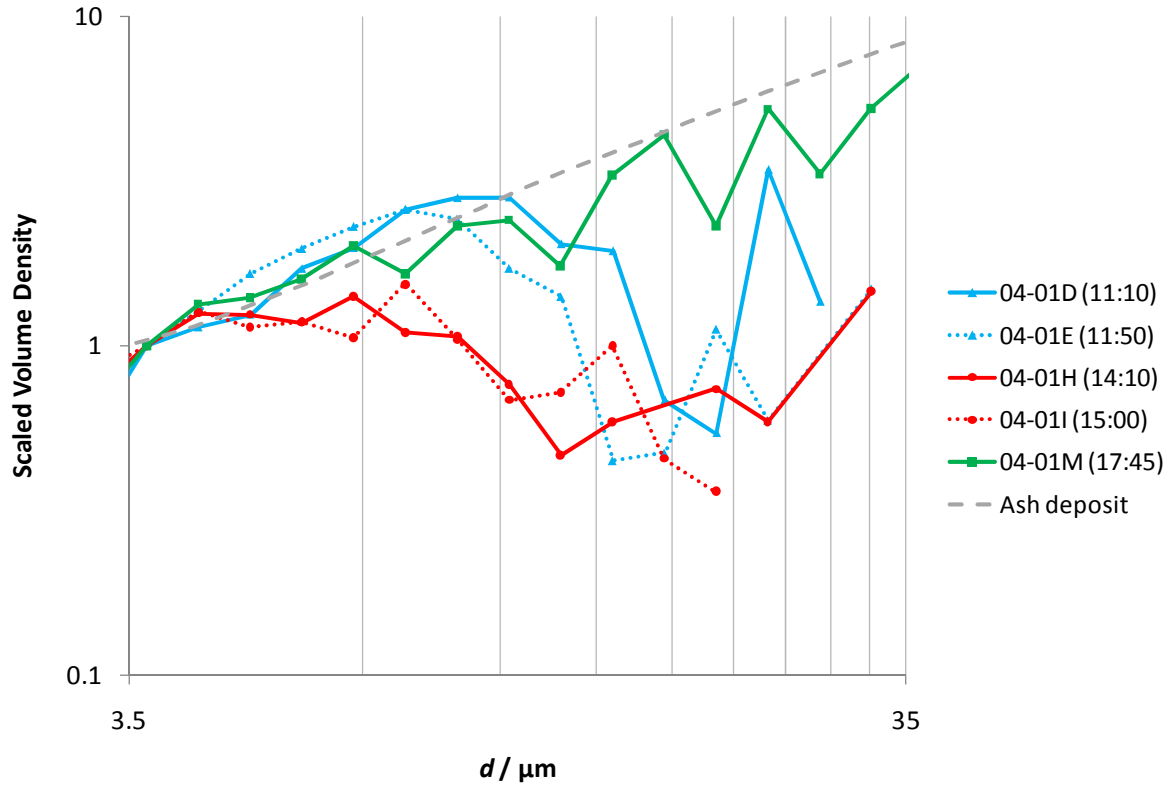
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616 Figure 3A Scaled number density, $N(d)/N(d_0)$ where $d_0 = 4 \mu\text{m}$, for air filtration
617 samples in Esquel on 4th June. Also shown is the scaled number density for
618 the ash deposit, analysed by laser diffraction (Watt et al., 2009).



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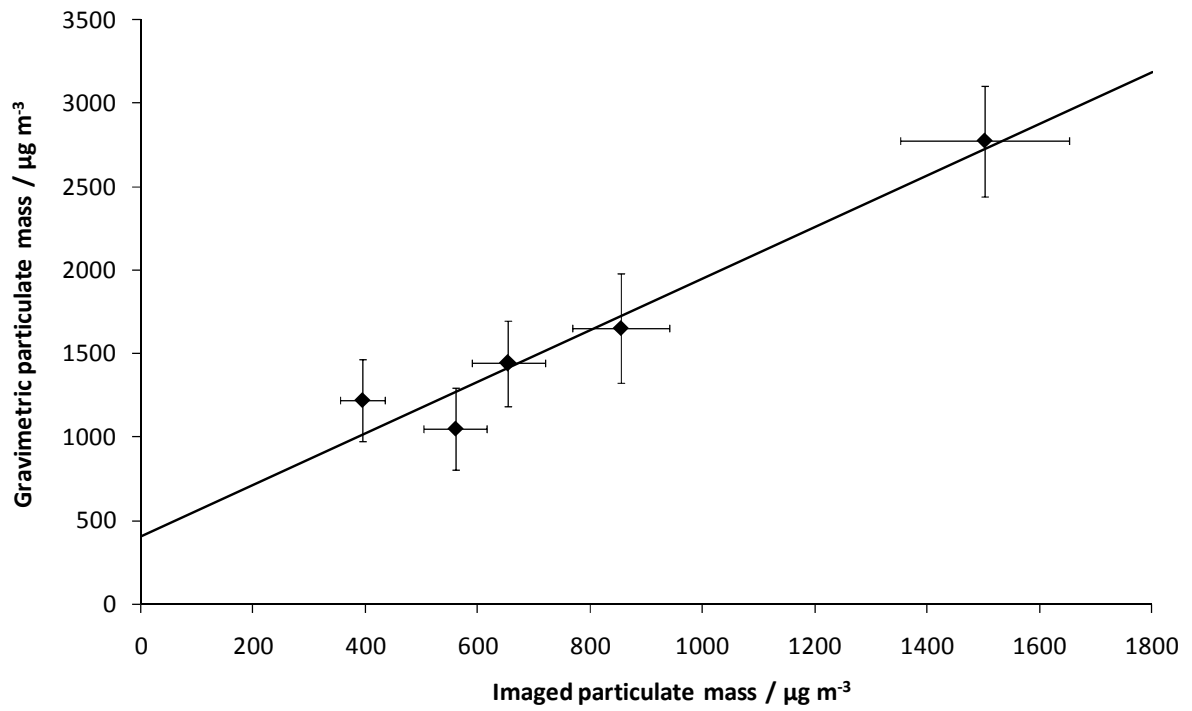
620 Figure 3B Scaled volume density, $V(d)/V(d_0)$ where $d_0 = 4 \mu\text{m}$, for air filtration
621 samples in Esquel on 4th June. Also shown is the scaled number density for
622 the ash deposit, analysed by laser diffraction (Watt et al., 2009).



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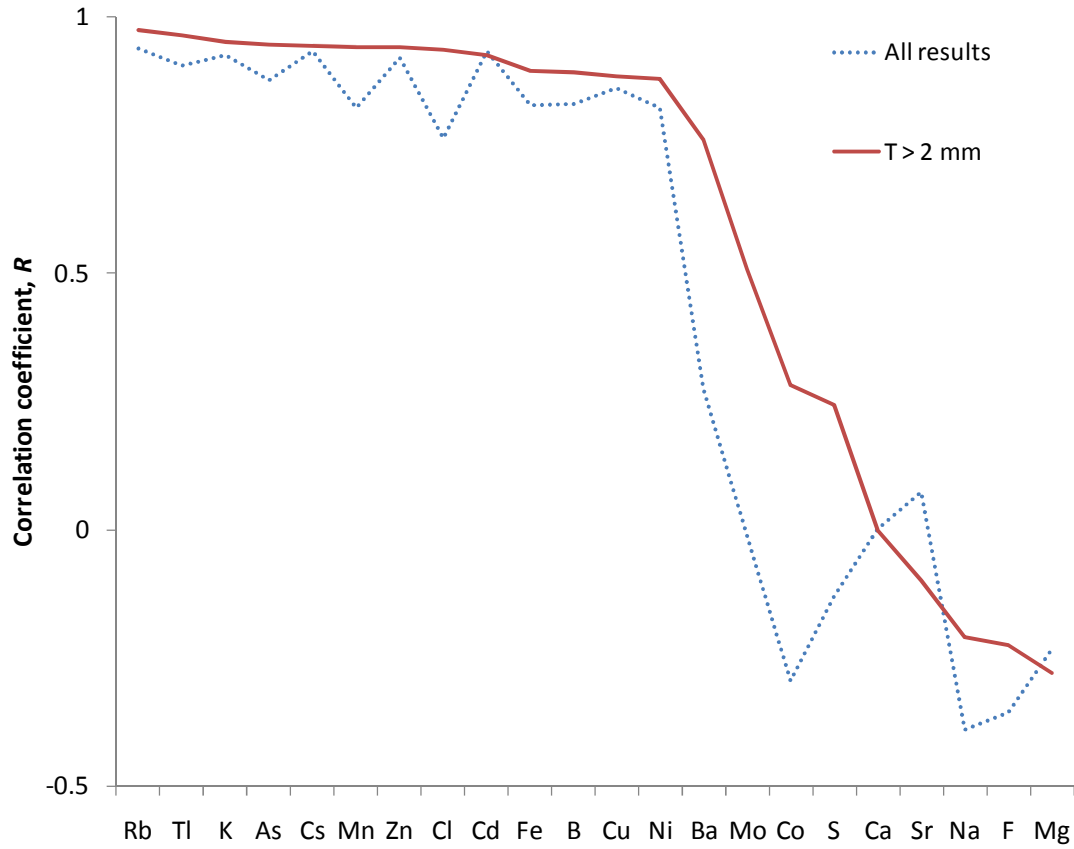
625 Figure 4 A comparison of gravimetric (m_g) and imaged (m_i) particulate mass. The
626 intercept is interpreted as the mass of particles with $d < 4 \mu\text{m}$.



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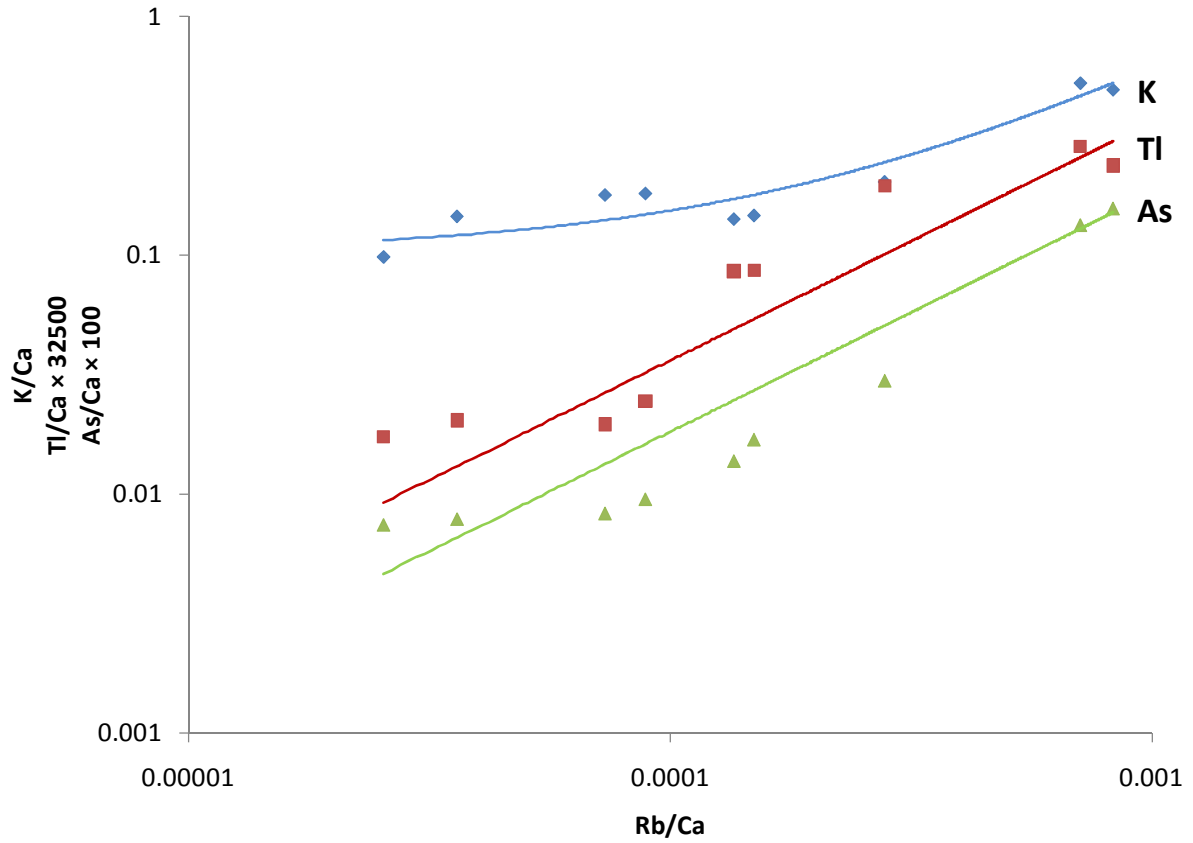
629 Figure 5A Correlation coefficients between concentrations of elements normalised to
 630 Ca (i.e., X/Ca) in ephemeral lakes and ash thickness at each location ($T /$
 631 mm).



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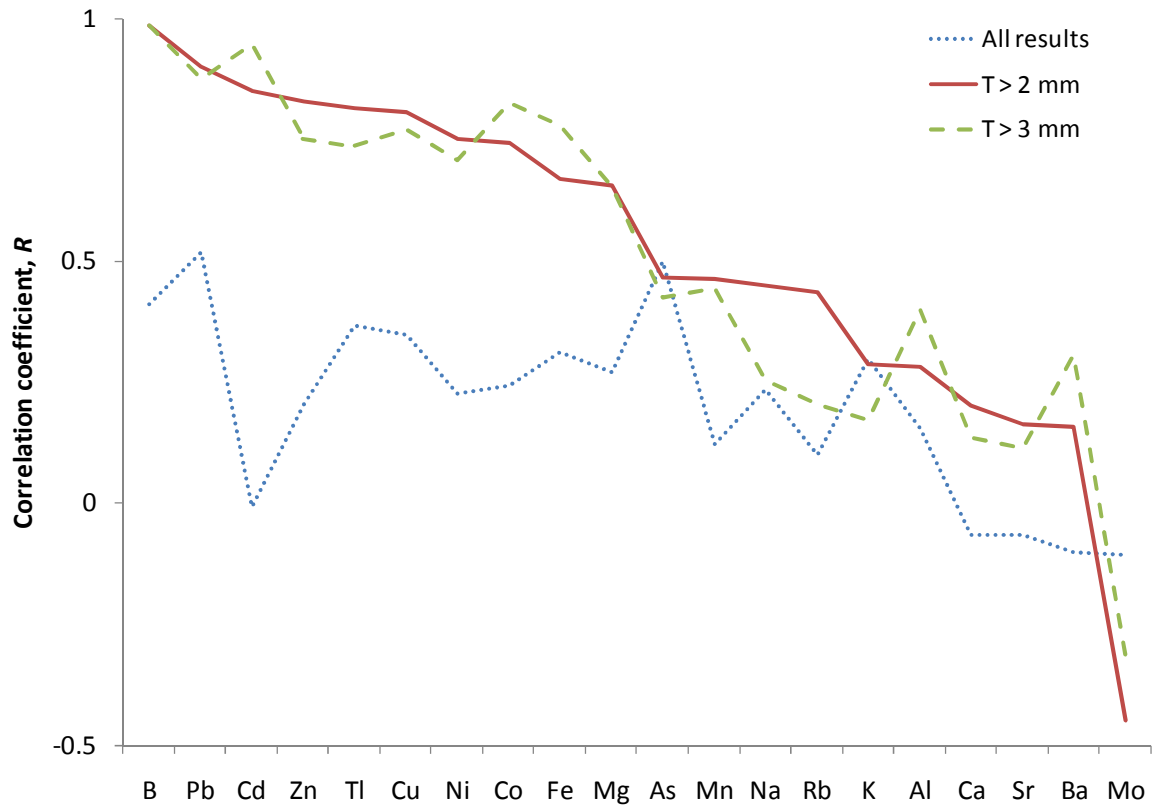
634 Figure 5B Log-log plot of (Y/Ca) where Y =K, Tl and As against (Rb/Ca). The gradient of the
635 linear fit (i.e., $(Y/Ca)=m(Rb/Ca)+c$) corresponds to Y/Z of the soluble material
636 available from the surface of the ash. Scaling is applied to As/Ca and Tl/Ca for
637 clarity. The graph for K shows a non-zero intercept ($c \neq 0$).



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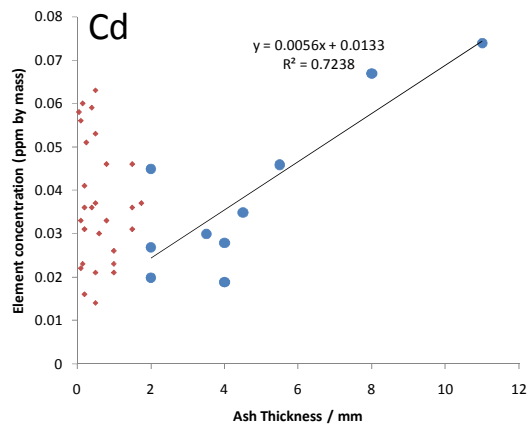
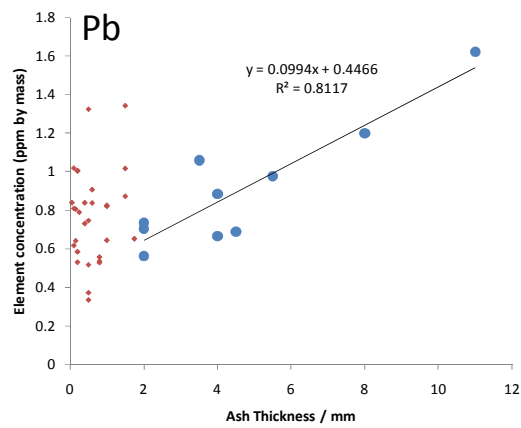
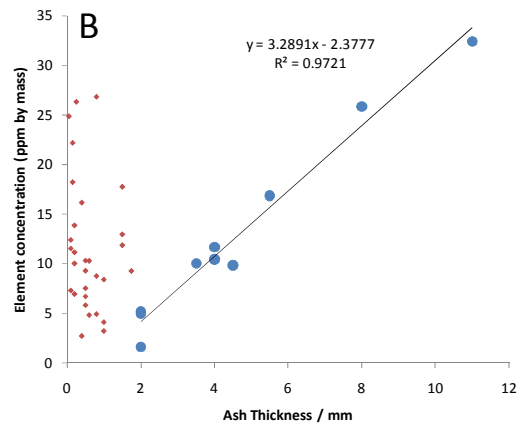
640 Figure 6A Correlation coefficients (R) between concentrations of elements
641 in Coirón grass and ash thickness at each location (T / mm).



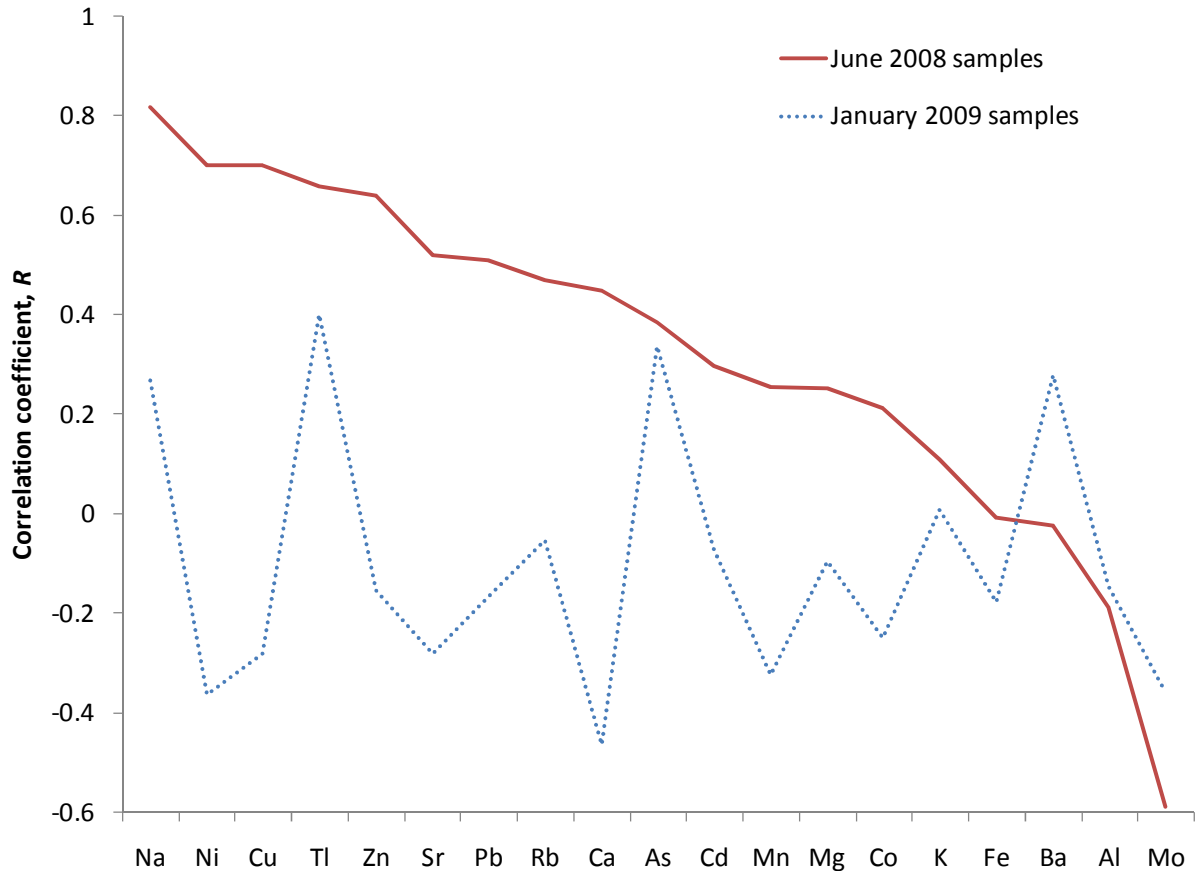
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644 Figure 6B Concentrations of B, Pb and Cd in Coirón grass (ppm by mass) against ash
645 thickness at each location (T / mm). Samples where $T > 2$ mm are highlighted.



650 Figure 6C A comparison of correlation coefficients (R) between concentrations of elements
 651 in Coirón grass and ash thickness at each location (T / mm) between June
 652 2008($2 < T/\text{mm} < 6$) and January 2009 samples ($2.5 < T/\text{mm} < 6$).



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655 Table 1 Sample details for filter packs.

Sample	Date	Sampling Times (local time)	Duration (t / min)	Flow-rate (F / L min ⁻¹)	Volume (V_A / L)	Mass (M / mg)
04-01A	04/06/08	08:40-09:15	35	20	700	0.3
04-01B	04/06/08	09:30-10:15	45	19	855	0.9
04-01C	04/06/08	10:20-10:55	35	19	665	1.2
04-01D	04/06/08	11:10-11:45	35	19	665	1.1
04-01E	04/06/08	11:50-12:40	50	18	900	1.3
04-01F	04/06/08	12:42-13:22	40	19	760	0.7
04-01G	04/06/08	13:25-14:05	40	19	760	1.1
04-01H	04/06/08	14:10-14:55	45	19	855	0.9
04-01I	04/06/08	15:00-15:50	50	18	900	1.1
04-01J	04/06/08	15:55-16:48	53	18	954	0.6
04-01K	04/06/08	16:50-17:33	43	20	860	1.5
04-01M	04/06/08	17:45-18:43	58	18	1044	2.9
04-01N	04/06/08	18:47-12:47	180	15	2700	5
05-27A	05/06/08	09:00 -	-	-	-	12.5
07-01A	07/06/08	23:00 -	-	-	-	3.2

Samples taken on 5th and 7th June were left running unattended from the time indicated until the battery was exhausted. For these samples, V_A is unknown but is presumed common to both.

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657

658 Table 2 $N(d_o = 4 \mu\text{m})$ for samples analysed using SEM.

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Sample	$N(d_o = 4 \mu\text{m})$
04-01 D	7.5×10^6
04-01 E	7.1×10^6
04-01 H	9.4×10^6
04-01 I	7.3×10^6
04-01 M	6.9×10^6

661 Table 3 Estimates for element concentrations relative to Rb within the
662 soluble material on the surface of ash particles.

Element	Y/Rb
As	1.9
B	3.5
Cd	0.0012
Cl	780
Cs	0.013
Cu	0.14
Fe	2.2
K	510
Mn	1.2
Ni	0.27
Rb	1
Tl	0.00098
Zn	3.3

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665 Supplementary Table 1 Sample details and data for lakes.

Sample Property	05-05	05-18	30-05	31-06	07-02	07-13	02-09
Latitude	-43.168	-43.528	-42.383	-43.325	-42.949	-42.138	-45.118
Longitude	-71.735	-71.204	-71.109	-70.896	-71.217	-71.018	-69.12
Ash thickness (<i>T</i> /mm)	15	9.5	3.5	3.5	2	0.4	0.1
Number of samples	2	1	2	2	2	1	1
Morphometry (m)	60×20×2	50×5×0.75	250×250×5	100×20×0.3	50×40×2	100×20×1	100×50×3
As	11	3	1.5	1.3	3.3	12	3.2
B	24	6.1	7.1	15	18	12	19
Ba	5.5	8.5	5.5	3.3	3.1	21	16
Ca	7400	10000	9400	15000	43000	32000	13000
Cd	0.0065	0.006	0.002	0	0	0	0
Co	0.044	0.045	0.055	0.084	0.19	0.41	0.1
Cs	0.079	0.041	0.017	0.048	0.019	0.063	0.009
Cu	0.89	0.47	0.27	0.07	0.29	0.73	0.21
Fe	23	18	14	28	60	66	19
K	3800	2000	1400	2700	5300	6100	970
Mg	1800	2600	2500	11000	9300	27000	420
Mn	6.6	3.4	1.8	1	1.9	8.9	4.8
Mo	0.33	0.55	0.48	0.45	0.56	2.8	0.77
Na	3700	4800	5200	19000	15000	36000	14000
Ni	1.9	1.1	0.74	0.6	1.2	3.1	0.77
Rb	5.7	2.8	1.4	1.2	1.3	6	0.43
Sr	26	37	40	88	97	140	20
Tl	0.006	0.006	0.0025	0.001	0.0025	0.01	0.002
Zn	19	7.1	1.9	1.7	2.4	8.3	0.27
Cl ⁻	5200	2300	2300	2400	3700	10000	5300
F ⁻	100	100	130	380	480	1200	200
SO ₄ ²⁻	3600	3000	2600	600	20000	15000	16000

Ash thickness (in mm) was measured as described in Watt et al., (2009). Lake morphometry was estimated and given as length × width × depth (in m). Element concentrations (determined by ICP-MS or IC) are given in units of ppb by mass.

667 Supplementary Table 2 Sample details for vegetation.

Sample	Collected	Latitude	Longitude	Ash Thickness (<i>T</i>) / mm
05-16	06/08	-43.488	-71.308	11
05-19	06/08	-43.561	-71.170	8
31-04	06/08	-43.015	-71.043	5.5
04-02	06/08	-43.063	-70.863	4.5
31-05	06/08	-43.134	-70.860	4
05-21	06/08	-43.609	-71.056	4
31-06	06/08	-43.325	-70.896	3.5
31-07	06/08	-43.528	-70.795	2
31-08	06/08	-43.640	-70.840	2
04-13	06/08	-42.960	-70.642	2
31-09	06/08	-43.797	-70.936	1.75
04-04	06/08	-43.035	-70.588	1.5
31-12	06/08	-44.046	-70.630	1.5
05-23	06/08	-43.585	-70.931	1.5
03-09	06/08	-43.467	-70.845	1
05-24	06/08	-43.559	-70.886	1
31-13	06/08	-44.177	-70.473	1
04-16	06/08	-42.656	-70.419	0.8
05-25	06/08	-43.513	-70.871	0.8
05-26	06/08	-43.474	-70.846	0.8
04-06	06/08	-43.075	-70.235	0.6
04-11	06/08	-43.089	-69.902	0.6
01-04	06/08	-43.727	-69.898	0.5
03-08	06/08	-43.573	-70.361	0.5
31-10	06/08	-43.959	-70.887	0.5
31-11	06/08	-43.994	-70.827	0.5
31-14	06/08	-44.292	-70.401	0.5
04-09	06/08	-43.214	-69.965	0.4
06-19	06/08	-41.968	-71.273	0.4
02-05	06/08	-44.646	-69.148	0.25
02-04	06/08	-44.459	-69.255	0.2
01-08	06/08	-43.672	-68.932	0.2
31-15	06/08	-44.419	-70.440	0.2
02-02	06/08	-44.146	-69.262	0.15
02-08	06/08	-45.035	-69.153	0.15
03-05	06/08	-43.950	-71.456	0.1
02-16	06/08	-45.448	-69.815	0.1
31-16	06/08	-44.589	-70.371	0.1
31-17	06/08	-44.731	-70.227	0.05
01-11	06/08	-43.877	-68.421	0.2
A-05	01/09	-42.004	-71.130	2.5 (0.2)
A-07	01/09	-41.841	-70.893	3 (0.2)
A-09	01/09	-42.384	-71.111	3 (0.5)
A-10	01/09	-42.467	-71.097	4 (1)
A-11	01/09	-42.659	-71.080	4 (2)
A-12	01/09	-42.767	-71.113	4 (1.5)
B-01	01/09	-42.969	-71.178	6 (3)
B-02	01/09	-43.050	-70.967	6(3)
B-03	01/09	-43.182	-70.845	4 (1)
B-04	01/09	-43.338	-70.886	3 (1)

For January 2009 samples, the ash thickness is given as the "original" ash thickness measured in June 2008 (Watt et al., 2009) followed by the remeasured ash thickness in parentheses.

669 Supplementary Table 3A Compositions of vegetation samples (ppm by mass)
 670 determined by ICP-MS (for elements Al – K).

Sample	Al	As	B	Ba	Ca	Cd	Co	Cu	Fe	K
31-07	2100	0.21	1.6	7.8	2100	0.045	0.26	4.8	840	4900
31-10	520	0.055	5.9	3	940	0.014	0.081	1.8	220	980
05-26	720	0.11	27	7.4	6400	0.033	0.11	4.2	280	2700
03-09	3600	0.35	4.2	11	2500	0.023	0.38	4.3	1400	3400
02-08	3700	0.25	18	23	4500	0.023	0.32	3.8	1200	730
31-08	2200	0.2	5.2	12	1500	0.02	0.22	4.2	840	1700
03-05	1900	0.29	12	36	3400	0.022	0.2	7.2	760	2100
04-04	3600	0.27	12	13	3400	0.046	0.37	8.5	1500	1800
31-05	800	0.1	10	3.3	3300	0.028	0.11	9.2	300	7600
31-14	660	0.081	10	2.9	2300	0.021	0.089	6.3	280	6000
31-12	2500	0.26	13	12	3300	0.036	0.39	11	1200	2600
02-04	1200	0.12	7	17	2500	0.031	0.19	3.3	500	2200
31-09	930	0.062	9.3	8.4	1400	0.037	0.14	8.5	330	6900
04-13	3500	0.28	5	11	2900	0.027	0.34	4.3	1300	2600
01-08	1600	0.11	11	11	2000	0.036	0.27	13	660	3400
31-13	1900	0.16	3.3	7.1	1000	0.026	0.38	14	700	1800
05-23	1500	0.15	18	8.8	1900	0.031	0.25	12	650	2400
01-11	6900	0.32	10	37	5200	0.041	0.97	15	2800	3900
31-04	2500	0.51	17	13	4200	0.046	0.36	10	1200	3700
31-15	360	0.073	14	2.4	1900	0.016	0.51	6.6	170	5500
02-02	1900	0.12	22	17	2800	0.06	0.25	7.6	720	1500
04-06	3000	0.22	10	9.9	1800	0.31	0.37	4.4	1200	3200
31-11	6300	0.33	9.3	16	3100	0.063	0.53	8.7	2400	3600
31-16	1000	0.16	7.3	6.2	2200	0.056	0.12	5.4	450	2100
31-17	1100	0.14	25	6	1600	0.058	0.14	7.5	470	2100
31-06	1400	0.13	10	5.3	1100	0.03	0.13	5.7	530	4900
04-09	2400	0.2	16	9.3	2500	0.036	0.3	7.8	880	3700
04-16	2000	0.19	5	4.9	3300	0.046	0.25	5.6	780	2800
03-08	320	0.066	7.6	1.6	1700	0.053	0.059	2.8	130	2200
05-25	1100	0.18	8.8	5.9	3300	0.32	0.14	3.2	460	3800
02-05	790	0.059	26	3.6	2400	0.051	0.19	9.8	430	3200
05-16	2000	0.27	32	7	2000	0.074	0.53	19	1700	6100
05-21	2400	0.32	12	10	1700	0.019	0.27	4	930	2300
05-24	1600	0.14	8.4	9.9	1500	0.021	0.27	6.1	680	1700
04-11	1800	0.37	4.9	21	3900	0.03	0.2	6.9	760	5100
01-04	3000	0.39	6.7	8.7	3800	0.037	0.47	6.4	1600	2300
05-19	5800	1.1	26	18	3200	0.067	0.58	5.7	2400	3000
06-19	2900	0.29	2.8	12	1900	0.059	0.32	5.9	1200	3900
02-16	3100	0.19	12	13	3600	0.033	0.35	5	1400	3500
04-02	2500	0.16	9.9	8.6	2000	0.035	0.38	5.5	970	1700
A-05	590	0.19	-	5.2	4600	0.02	0.31	7	660	5800
A-07	550	0.67	-	4.2	3300	0.019	0.31	5.3	770	3900
A-09	890	0.47	-	11	10000	0.027	0.39	7.9	890	6600
A-10	270	0.17	-	6.3	3900	0.1	0.14	5	280	4900
A-11	180	0.13	-	4	2800	0.015	0.098	3.8	200	4200
A-12	320	0.22	-	7.7	3400	0.034	0.18	5.1	320	3900
B-01	980	1.1	-	12	3900	0.028	0.42	8.2	1100	8100
B-02	230	0.33	-	6.8	3000	0.012	0.14	3	290	3000
B-03	280	0.3	-	2.4	3600	0.018	0.16	3.8	340	4700
B-04	830	0.5	-	7.7	8500	0.04	0.43	7.3	990	5400

672 Supplementary Table 3B Compositions of vegetation samples (ppm by mass)
 673 determined by ICP-MS (for elements Mg – Zn)

Sample	Mg	Mn	Mo	Na	Ni	Pb	Rb	Sr	Tl	Zn
31-07	630	40	1.8	180	0.85	0.56	0.9	11	0.01	11
31-10	250	13	0.57	60	0.43	0.37	0.29	6.3	0.008	10
05-26	610	110	1.8	160	0.62	0.54	0.57	26	0.01	11
03-09	720	59	0.53	380	1.1	0.82	2.3	15	0.014	11
02-08	470	43	7.8	730	0.92	0.81	1.7	16	0.013	5.6
31-08	390	22	0.84	220	0.64	0.7	0.94	10	0.012	9.3
03-05	880	21	0.18	320	0.81	1	2.1	26	0.014	13
04-04	490	38	1.8	310	1.5	1	1.6	9.5	0.018	9.9
31-05	880	35	0.62	240	1.1	0.88	2.4	6	0.017	22
31-14	710	22	0.29	95	0.76	0.52	0.88	7.2	0.009	9.6
31-12	550	33	1.3	450	1.7	0.87	1.3	26	0.019	18
02-04	470	18	0.32	170	0.85	0.59	0.95	10	0.012	21
31-09	900	12	3.2	180	1.1	0.65	1.5	10	0.011	11
04-13	640	35	2.8	210	0.81	0.73	1.5	14	0.013	9
01-08	420	36	1.7	160	0.87	1	1.7	13	0.015	16
31-13	320	20	0.77	230	0.6	0.83	0.92	5.9	0.011	11
05-23	430	19	0.57	260	1	1.3	1.1	10	0.016	15
01-11	1200	90	1.5	580	3.1	1	3.1	56	0.022	25
31-04	670	34	0.61	520	1.6	0.98	1.7	37	0.02	18
31-15	340	18	0.45	81	0.69	0.53	0.63	6	0.01	12
02-02	400	30	1.6	230	0.97	0.64	1.2	18	0.015	15
04-06	540	38	0.81	230	2.9	0.91	1.4	9.9	0.013	14
31-11	810	50	1.6	210	1	1.3	2.3	18	0.021	17
31-16	450	19	0.85	120	0.65	0.62	0.58	13	0.011	8.6
31-17	310	22	0.47	170	1	0.84	0.77	10	0.015	11
31-06	530	11	2	270	0.83	1.1	1.3	5	0.014	13
04-09	550	30	1.1	250	1.4	0.73	1.5	14	0.018	47
04-16	680	36	1.1	280	1	0.53	1	14	0.011	8.5
03-08	270	12	0.91	69	0.49	0.34	0.61	6.2	0.008	5.8
05-25	590	82	1.6	180	0.54	0.56	0.67	13	0.009	7.3
02-05	580	26	0.78	190	2.9	0.79	0.98	11	0.013	39
05-16	890	45	0.85	230	2.7	1.6	1.4	9.6	0.021	32
05-21	390	30	1.4	230	0.53	0.66	1.2	11	0.011	11
05-24	420	23	0.89	190	1.5	0.65	0.79	9.1	0.011	9.9
04-11	1100	21	0.78	540	0.99	0.84	5.9	26	0.015	16
01-04	790	34	2.4	570	1.1	0.75	1.8	34	0.019	13
05-19	860	57	1.2	780	0.8	1.2	2.8	17	0.019	16
06-19	400	41	3.1	130	0.63	0.84	2.8	9.5	0.011	13
02-16	770	43	2.7	540	0.76	0.81	1.3	19	0.016	13
04-02	600	63	1.1	320	1.6	0.69	1.3	11	0.012	12
A-05	1400	45	1.3	110	0.82	0.8	3	16	0.0065	12
A-07	700	23	2.8	170	1	0.65	1.1	9.4	0.016	6
A-09	530	31	1.8	170	0.89	1.3	1.6	33	0.016	13
A-10	370	47	0.94	76	0.66	0.58	1.7	15	0.0079	19
A-11	570	24	1.3	83	0.98	0.58	1.7	11	0.0071	11
A-12	690	61	5	110	0.86	0.66	0.9	13	0.01	11
B-01	900	32	0.53	260	1.1	1.1	2.9	16	0.03	12
B-02	760	24	0.96	110	0.45	0.58	0.74	13	0.0095	5.2
B-03	350	17	1.7	83	0.61	0.81	0.59	11	0.0096	5.7
B-04	690	80	2.3	140	1.5	1	1.4	22	0.014	10