

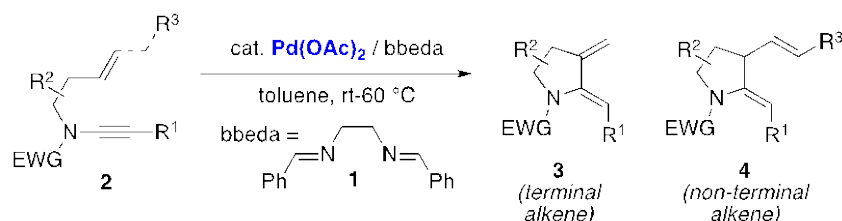
Isotope effects in mechanistic studies of transition metal-catalyzed cycloisomerization

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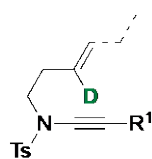
Transition metal-catalyzed cycloisomerization reactions offer an atom-economic route from simple acyclic unsaturated precursors to relatively complex (hetero)cyclic products. The product structure depends intimately on the reaction mechanism, which varies widely between different catalysts.¹

Among the most well-established cycloisomerization catalyst systems are the palladium(II) and palladium(0) catalyzed transformations of enynes into 1,3- and 1,4-dienes, developed extensively by the Trost group.² Our group has found that the combination of Pd(II)(OAc)₂ and the ligand *bis*-benzylidene ethylenediamine (bbeda, **1**, Scheme 1)³ is particularly efficient at promoting the cyclization of enynamides **2** to various azacycles **3** / **4**.⁴ However, the mechanism by which this cycloisomerization proceeds has remained a mystery for almost 30 years.

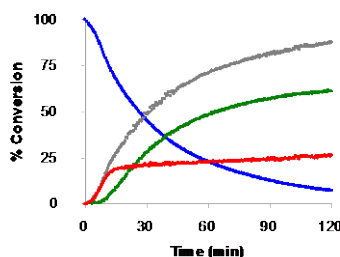


Using a combination of isotopically-labelled substrates and catalysts, deuterium-enriched reaction media, computational modelling, and kinetic profiling, we have gathered a significant body of evidence that offers a solution to this mechanistic puzzle, and in doing so reveals unusual mechanistic features of this classic cyclization process.⁵ The details of our investigations will be discussed in this talk.

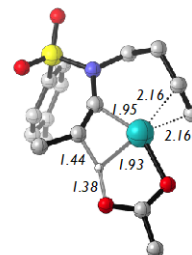
Labelled substrates



Kinetic profiling



Computational modelling



References

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5. Mekareeya A, Walker PR, Couce-Rios A, Campbell CD, Steven A, Paton RS, Anderson EA, *J. Am. Chem. Soc.* 2017, **139**, 10104-10114.