

Partial Melting and Structural Disorder in Models of Irradiated Amorphous $\text{Ge}_2\text{Sb}_2\text{Te}_5$

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Identifying the link between the energy landscape that arises in chemically complex amorphous solids and the evolution of the structural disorder in the host glassy matrix upon ion irradiation can provide a fundamental understanding needed for the development of functional materials with enhanced radiation tolerance. Herein, a comprehensive analysis of the structural transformations that occur when an amorphous phase-change memory material is exposed to radiation-damage events is presented. Different measures of atomic-structural (dis)order—geometric, topological, and chemical—are analyzed in models of irradiated glassy $\text{Ge}_2\text{Sb}_2\text{Te}_5$. The modeling results demonstrate that the degree of chemical disorder is increased within the postirradiation recovered amorphous network of $\text{Ge}_2\text{Sb}_2\text{Te}_5$, while local atomic environments with homopolar bonds are formed in the geometry of the irradiated structures, as a result of the primary knock-on atom simulations. The molecular dynamics trajectories of the high-energy nonequilibrium cascades indicate partial local melting of the glass structure. The simulated system can access liquid-like states in the energy landscape when irradiated, which lowers the energy barriers for local relaxation, leading to further atomic rearrangements and a structural recovery of the glass network. This observation provides an important insight into the radiation tolerance of these amorphous materials.

1. Introduction

Every glassy material has its own (thermal) history, which is encoded in the evolution of its properties, while the final amorphous state depends on the applied temperature profile protocol used to generate it.^[1] For a given composition, there is a mosaic of metastable amorphous structures that correspond to local energy minima, namely, “inherent structures,”^[2] while this ensemble of accessible states can vary when exposed to nonequilibrium conditions. Moreover, the degree of short-range structural ordering in amorphous materials also depends on their preparation protocol.^[3] Ion irradiation is a nonequilibrium process that involves the direct bombardment of a material with high-energy ions, leading to unique amorphous structures. This technique has been employed for a wide range of materials research and purposes, including dopant implantation in semiconductor device fabrication and material-processing routes toward enhanced functionalities, as well as for introducing alloy disorder

in chalcogenide materials.^[4] The ability to predict and characterize the structural transformations of glassy materials exposed to nonequilibrium conditions is crucial in developing semiconducting materials with tailored properties.


Phase-change random-access memory (PCRAM) devices, based on Ge–Sb–Te alloys, are part of some commercial nonvolatile solid-state drives, and they are also promising candidates for neuromorphic and in-memory computing applications,^[5–7] as well as for new storage-class memory devices.^[8] Such PCRAM devices encode stored digital binary data as metastable structural states of the same chalcogenide material.^[9] Their operational principle relies on the fast (of the order of nanoseconds) and reversible (up to millions of cycles) switching between a low-electrical-resistance (i.e., conductive) crystalline state (“1-bit”) and a high-electrical-resistance (i.e., resistive) glassy state (“0-bit”), induced via Joule heating by the application of appropriate voltage (or laser) pulses.^[10,11] As an innovative solution, chalcogenide phase-change materials have been proposed for spaceborne solid-state memory modules because of their nonvolatile, reconfigurable, fast-switching, and space-radiation-tolerant capabilities.^[12,13]

In our previous work, nonequilibrium irradiation cascades in amorphous $\text{Ge}_2\text{Sb}_2\text{Te}_5$ phase-change memory material have been investigated by means of first-principles molecular

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dynamics simulations.^[13] Self-ion irradiation was modeled by carrying out a thermal-spike simulation, in which the collision effects were caused by primary knock-on Te atoms with initial kinetic energies ranging from 25 to 200 eV.^[13] A detailed analysis of the kinetic profile of the nonequilibrium radiation-damage cascade showed the rapid time evolution of the dynamics of the cascade within the glass network, with the ballistic phase being very short.^[14] These simulations revealed the formation of large voids within the amorphous network during the irradiation cascade events.^[13] However, the final equilibrated simulated models showed a notable degree of healing and reversibility, as both the atomic and electronic structures recovered from the damage imposed during irradiation.^[13] Nevertheless, structural modifications of the chemical order did occur within the glass structure after exposure to irradiation.^[13,14] An increased proportion of edge-sharing tetrahedra was observed, which is in direct correlation with the increased population of (specific) homopolar bonds in the final equilibrated structure.^[13,14] Moreover, the simulations suggested, based on an analysis of the thermal transients from the molecular dynamics trajectories, that the lattice thermal conductivity of the irradiated amorphous models can be significantly reduced (by as much as 60%).^[15]

It has been reported, using classical molecular dynamics simulations, that the topography of the enthalpy landscape plays a significant role in the irradiation-induced amorphization of α -quartz.^[16] It appeared that, upon the irradiation process, the simulated structure explores “forbidden” states of the landscape, which typically are not accessible via a thermal quench of molten silica, resulting in a different character of disordering compared to that obtained upon the normal vitrification process.^[16,17] In addition, it was shown that the damage imparted during irradiation of the crystalline structure saturates when the system accesses a local region of the energy landscape that resembles the atomic configuration of a liquid structure, thereby facilitating relaxation and, hence, preventing the accumulation of further damage.^[16]

Comparisons between glassy models before and after irradiation are not always straightforward since every glassy structure is out of equilibrium, and hence their properties depend on their thermal history. Moreover, the intricate energy landscape of an amorphous structure can have profound effects on the properties of the material and its response to irradiation. An atom-density-based similarity metric was utilized to examine the degree of structural (dis)similarity between the irradiated $\text{Ge}_2\text{Sb}_2\text{Te}_5$ configurations studied here and by visualization in a 2D map. The evolution of the potential energy landscape of the simulated system during ion irradiation is presented in a configurational map to illustrate how its nature can facilitate intrinsically the reorganization of the atomic structure and the subsequent recovery of the glassy network. Our previous structural analysis on irradiated glassy $\text{Ge}_2\text{Sb}_2\text{Te}_5$ models is extended here by employing geometric, topological, and chemical estimations of disorder to gain further insight into the structural evolution and the overall response of the amorphous material to the irradiation events. In this way, the impact of the radiation-damage cascade on the local atomic structure of the glass is explored, while implications related to the functional properties of the material are discussed.

2. Partial Melting of the Amorphous Structure

In this study, the final atomic configurations, at the end of the molecular dynamics trajectories, obtained previously from 50, 100, and 200 eV Te-atom irradiation simulations,^[13] as well as the initial pristine glassy $\text{Ge}_2\text{Sb}_2\text{Te}_5$ structure, were further optimized with density-functional theory (DFT) simulations and by performing geometry optimization calculations. The similarity relationships between the optimized amorphous models, as well as the structural changes upon the different irradiation events, can be established, at a glance, through the construction of a configurational map.^[18] This can be achieved by using an informative representation for the configuration of each model structure, together with a suitable dimensionality-reduction technique.^[19] A smooth-overlap-of-atomic-positions (SOAP) descriptor was employed to represent each simulated structure,^[20] globally averaged over every local atomic environment within each different configuration, while multidimensional scaling (MDS) has been adopted as the embedding technique.^[21,22]

Each SOAP descriptor is a vector (\mathbf{q}) representing an atomic environment and existing in a high-dimensional space. A measure of the pairwise similarity between two atomic environments is given by the dot product of the corresponding SOAP vectors.^[18,20] SOAP descriptors are normalized such that the dot product of any two of them lies between 0, corresponding to no similarity whatsoever, and 1, corresponding to indistinguishable local environments. SOAP descriptors of an atomic environment can be aggregated into a global description of a configuration by appropriate averaging over the power-spectrum components of all the different atomic environments in the configuration.^[20]

Based on this information, for two simulated geometries α and β , a configurational distance metric, $d_{\alpha\beta}$, can be constructed as follows:

$$d_{\alpha\beta} = \sqrt{2 - 2\langle \mathbf{q}_\alpha, \mathbf{q}_\beta \rangle} \quad (1)$$

where $\langle \mathbf{q}_\alpha, \mathbf{q}_\beta \rangle$ is the dot product of the corresponding SOAP descriptors of the configurations α and β . A more detailed explanation on the SOAP descriptor and the approach employed here can be found in the Experimental Section.

MDS is a dimensionality-reduction technique that approximately preserves distances and can thus be used to visualize the SOAP data in a 2D space, where similar configurations appear closer together.^[23] For all model structures, the MDS approach constructs embedding vectors (\mathbf{e}), which have a Euclidean distance that is as close as possible to the high-dimensional (SOAP-based) distance metric defined in Equation (1). This is achieved by minimizing the stress function (σ), using the SMACOF algorithm,^[22] calculated as

$$\sigma(\mathbf{q}_\alpha, \mathbf{q}_\beta) = \sqrt{\frac{\sum_{\alpha \neq \beta} (d_{\alpha\beta} - \|\mathbf{e}_\alpha - \mathbf{e}_\beta\|)^2}{\sum_{\alpha \neq \beta} d_{\alpha\beta}^2}} \quad (2)$$

where $d_{\alpha\beta}$ is the SOAP distance between the α and β configurations in the high-dimensional space, and $\|\mathbf{e}_\alpha - \mathbf{e}_\beta\|$ is the

distance of the corresponding embedding vectors in the reduced (in our case 2D) representation. The final value of the stress reflects the quality of the representation for all the produced data points; the stress approaches zero when the original distances are faithfully maintained.^[23,24] We note that the MDS implementation available in the scikit-learn library^[25] was employed here. A further description on the working principle of the MDS technique is given in the Experimental Section.

The MDS map that is constructed with configurations from the DFT geometry optimization calculations of three irradiated (50, 100, 200 eV), as well as the initial (melt-quenched-generated, 0 eV) $\text{Ge}_2\text{Sb}_2\text{Te}_5$ glassy model structures, is shown in **Figure 1**. In addition, data points that are sampled during the optimization process of the respective DFT calculations performed here have been added to the map to highlight the local search of the geometry optimizer while moving “downhill” on the potential energy surface into the local minimum. The rocksalt-like crystalline structure of $\text{Ge}_2\text{Sb}_2\text{Te}_5$ ^[26,27] and configurations that resemble the liquid phase of the material under study^[28] have been used as reference points in the 2D map. We note that for the data set of the simulated structures analyzed in this study, a stress value of 0.06 was obtained, suggesting that the embedding faithfully preserves the high-dimensional distances between the different configurations.

In the 2D space of **Figure 1**, data points that are close to each other on the map indicate close similarity between the model structures, from the point of view of the configurational descriptor.^[18] It can be observed that the configurations associated with the final irradiated glassy $\text{Ge}_2\text{Sb}_2\text{Te}_5$ models, obtained from the nonequilibrium radiation-damage simulations, move further away from the point corresponding to the pristine amorphous model, obtained from a melt-quench simulation. This highlights that the glass has experienced structural modifications due to the irradiation events that are captured by the embedding.

The distances between the configurations obtained after the fast thermal quench (i.e., Newtonian cooling) of each of the

radiation-damage events considered here, and the inherent structures that are obtained from their subsequent geometry optimization calculations, can be roughly correlated with the local energy barrier heights in the energy landscape picture.^[29,30] This is similar to the more commonly used metric of the mean-squared displacements between inherent and thermalized structures.^[31] Sample configurations, at different iteration steps, from the geometry optimization process to the respective inherent structures move away from the crystal and liquid regions, as they approach the local minima, and they gather into a distinct cluster area of the configurational map in **Figure 1**.

During irradiation, the short timescale within which the radiation-damage cascade interacts with the host solid matrix implies that the structural transformations in the amorphous network occur far from thermodynamic equilibrium. In order to probe the evolution of the energy landscape during the atomic-damage events, raw data from the DFT molecular dynamics trajectories of the different simulations previously performed^[13,14] were utilized to construct 2D maps, by using again a combination of SOAP descriptors and MDS for the embedding, for the 50, 100, and 200 eV primary knock-on atom events. In addition, each data point was plotted as a function of the calculated potential energy from the electronic-structure simulations, and the overall results are shown in **Figure 2**. Similarly to what was presented in the previous map, crystalline, liquid (1200 K), and glassy (300 K) configurations of the $\text{Ge}_2\text{Sb}_2\text{Te}_5$ material^[28] have been used as three distinct reference points to facilitate comparisons.

It can be observed that, during the irradiation simulations, the system visits configurations between the liquid and amorphous states, while with increasing initial primary knock-on atom energy, it approaches the reference liquid configuration. For the case of the 200 eV radiation-damage simulation, the trajectory of the modeled system was able to cover the entire distance between the liquid and amorphous regions, highlighting clearly that, during this irradiation event, liquid-like configurations are being accessed by the simulated structure. From the high-energy

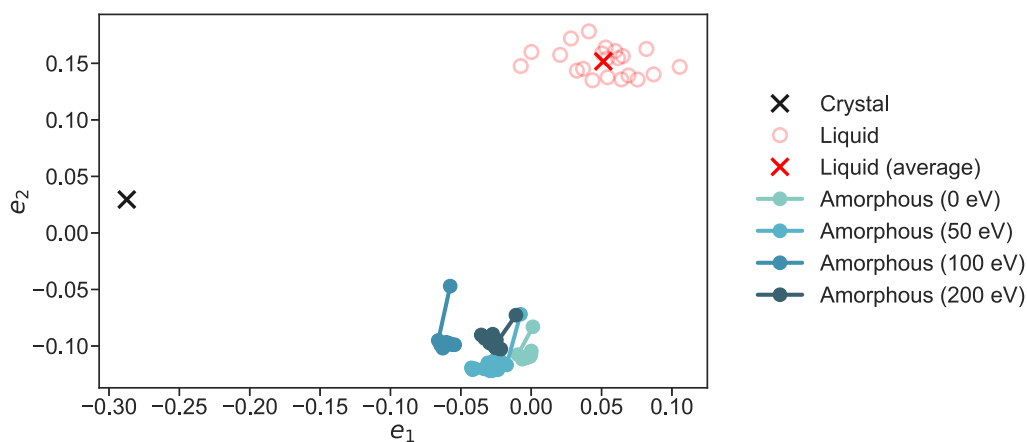


Figure 1. A 2D SOAP-based similarity map for the geometry optimization processes of configurations obtained from different irradiation simulations, corresponding to 0, 50, 100, and 200 eV primary knock-on Te atom energies, shown as different shades of blue. The map includes liquid configurations (at 1200 K) and a model crystalline structure, shown in red and black, respectively. The drawn lines are a guide to the eye connecting amorphous configurations from the end of each nonequilibrium molecular dynamics simulation to the basin of attraction of the nearest local minimum (i.e., inherent structure), as obtained through geometry optimization of the simulated structures. The x-axis and y-axis correspond to the two MDS coordinates (e_1 and e_2 , respectively) that reflect the embedding of the SOAP data and they are dimensionless.

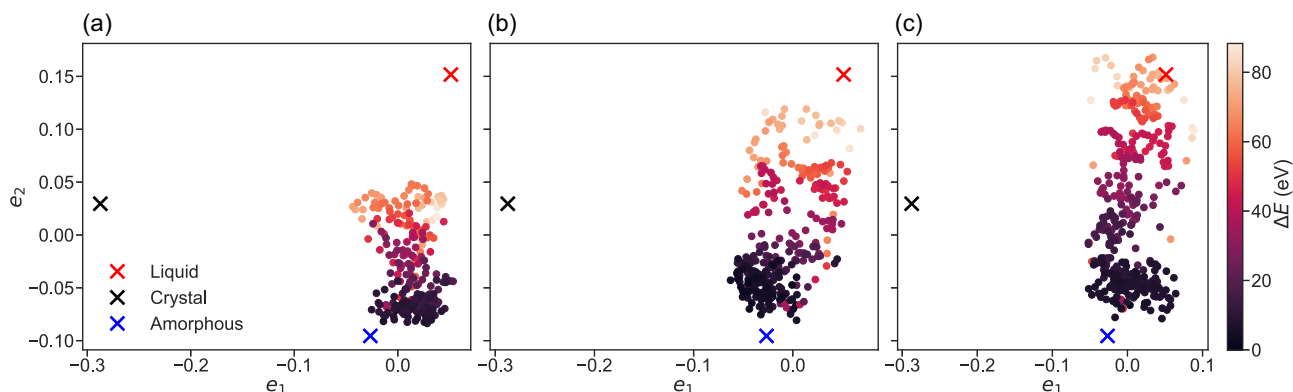


Figure 2. 2D SOAP-based similarity maps visualizing the entire molecular dynamics trajectories for: a) 50 eV; b) 100 eV; and c) 200 eV primary knock-on Te atom energies. The data are plotted as a function of the calculated potential energy of the modeled system, at each point, relative to the lowest energy configuration in each trajectory (ΔE). Reference configurations for the liquid (1200 K), crystalline, and amorphous (inherent structure, quenched from 300 K) regions are shown as red, black, and blue crosses, respectively. The x-axis and y-axis, at each panel, correspond to the two MDS coordinates (e_1 and e_2 , respectively) that reflect the embedding of the SOAP data and they are dimensionless.

liquid states, structural relaxation can occur quickly during the remaining thermal quench of the irradiated structure. Such fast relaxation and structural reorganization can assist the recovery of the glassy network of the $\text{Ge}_2\text{Sb}_2\text{Te}_5$ simulated structure,^[13] which is also facilitated by the existence of many different bonding arrangements, with various atomic-coordination environments, within this chalcogenide material.^[32,33]

3. Structural Disorder Evolution

Amorphous $\text{Ge}_2\text{Sb}_2\text{Te}_5$ is a material that exhibits a wide range of different degrees of structural ordering within its glassy network. In order to investigate the effects of the radiation-damage cascade on the atomic structure of the $\text{Ge}_2\text{Sb}_2\text{Te}_5$ models, geometric, topological, and chemical measures of disorder have been employed here for the different primary knock-on atom events.

For the geometric disorder, the two-body excess entropy was used as an indicator,^[34] as given by the formula:

$$S^2 = -2\pi\rho k_B \int \{g(r) \ln[g(r)] - [g(r) - 1]\} r^2 dr \quad (3)$$

where $g(r)$ is the pair-distribution function, ρ is the density of the simulated system, and k_B is the Boltzmann constant. **Figure 3** shows the variation of the geometric disorder of the simulated amorphous structure, at each irradiation event, as a function of the initial kinetic energy of the primary knock-on atom, which was quantified relative to the two-body excess entropy of the pristine glassy model (0 eV) as follows:

$$\Delta S^2(E_{PKA}) = \frac{S^2(E_{PKA}) - S^2(0)}{S^2(0)} \quad (4)$$

It can be observed that the geometric-disorder indicator oscillates across the different nonequilibrium irradiation simulations. The final equilibrated structures after the 25 and 100 eV primary knock-on atom events are more disordered than the initial $\text{Ge}_2\text{Sb}_2\text{Te}_5$ glassy model, whereas the 200 eV cascade-damage

event results in an amorphous structure with enhanced structural order compared to the pristine model. We note that the $g(r)$, that is required in the calculation of the respective S^2 in Equation (3), was averaged over 10 ps of a DFT molecular dynamics simulation at 300 K, at the end of each irradiation simulation previously performed, in order to consider the dynamical behavior of the system in the estimation of the geometric-disorder indicator.

The shortest-path ring statistics were calculated using the Franzblau algorithm,^[35] to evaluate the topological disorder in the glassy network of the irradiated amorphous $\text{Ge}_2\text{Sb}_2\text{Te}_5$ models. As a measure of disorder, we use the Shannon entropy of the relevant distribution of rings,^[36] defined as

$$H = - \sum_n P_n \log_2(P_n) \quad (5)$$

where P_n is the probability of forming a ring of size n , based on the calculated frequency of each shortest-path ring that is present inside the simulated structure. Considering that, for all the

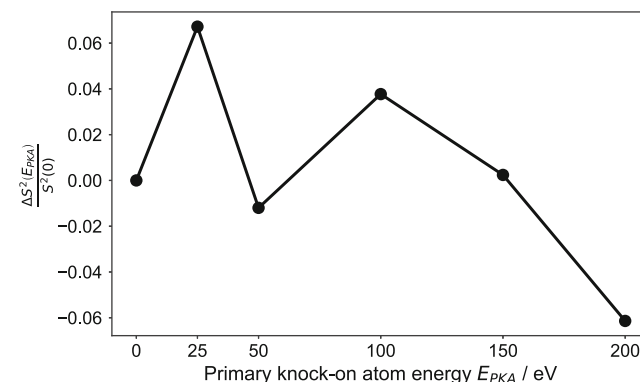


Figure 3. A measure of the geometric disorder of the simulated amorphous structure. Two-body excess entropy for the different primary knock-on Te atom events, relative to the calculated value for the pristine glass structure.

amorphous models, the ring distributions are defined on the same discrete probability space, the Kullback–Leibler divergence (or discrimination information),^[37] relative to the pristine glass structure before irradiation, is calculated as

$$D_{KL}(E_{PKA}||0) = \sum_n P_n(E_{PKA}) \log \left[\frac{P_n(E_{PKA})}{P_n(0)} \right] \quad (6)$$

The results of the analysis are shown in **Figure 4** as a function of the different primary knock-on atom energies. The topological disorder of the final equilibrated amorphous structures after the different irradiation simulations is increased compared to the pristine glassy model. The magnitude of this measure reaches a maximum for the 100 eV irradiation event, while it monotonically decreases for the 150 and 200 eV cascade-damage simulations, highlighting that the degree of change of the topological disorder in the glassy structures is smaller for these initial kinetic energies. A possible explanation for this behavior is due to the fact that, during the 200 eV primary knock-on atom event, the simulated structure is able to explore liquid configurations, in contrast to the 100 eV cascade event (see Figure 2), that can help the system to escape any otherwise forbidden states in the potential energy landscape.

The ratio between the calculated two-body excess entropy of the different irradiated models and the corresponding value of the pristine amorphous model ranges between -0.07 and 0.07 . The Kullback–Liebler divergence of their ring-size distributions covers a similar range of values, calculated to be between 0 and 0.14, again with respect to the glass structure before the primary knock-on atom events. Hence, these two measures of disorder show a similar span of structural modifications for the irradiated simulated structures compared to the initial glass structure, indicating a possible correlation between the topology of the amorphous network and the atomic geometry of the models.

In order to capture the time evolution of the chemical order of the amorphous structure during the 200 eV primary knock-on atom simulation, the binary order parameter of Cargill and Spaepen has been calculated,^[38] from the perspective of the quasibinary (A/B) classification, which has been previously used in other investigations of phase-change memory materials.^[39,40] This classification groups Ge and Sb atoms together as type

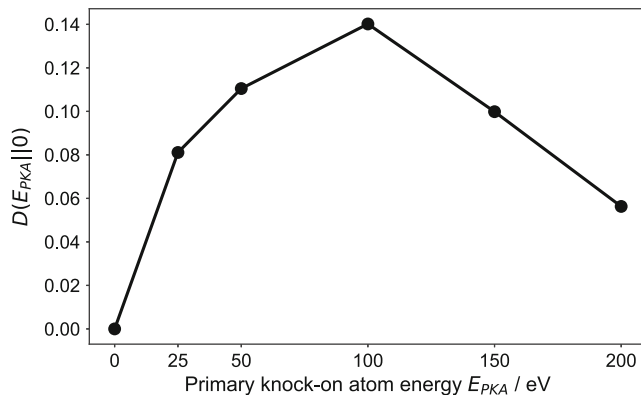


Figure 4. A measure of the topological disorder of the simulated amorphous structure. The Kullback–Leibler divergence of the shortest-path ring distributions for the different primary knock-on Te atom events, relative to the ring-size distribution of the pristine glass structure.

A and leaves Te as type B, highlighting the prevalent pattern of A – B alternation observed in phase-change materials.^[41,42] In such a case, the quasibinary chemical-order parameter of a given structure is defined as

$$\eta_{AB}^0 = \frac{N_{AB}N}{X_B N_A N_B} - 1 \quad (7)$$

where the parameter η^{\max} is given by

$$\eta_{AB}^{\max} = \begin{cases} \frac{X_B N_B}{X_A N_A}, & \text{if } X_B N_B \leq X_A N_A \\ \frac{X_A N_A}{X_B N_B}, & \text{if } X_B N_B > X_A N_A \end{cases} \quad (8)$$

In the above equations, X_A and X_B are the atomic fractions of each (generic) atom type. The quantities N_A and N_B are the average coordination numbers of A and B atoms, while N_{AB} and N_{BA} are the average number of B atoms around A, and the average number of A atoms around B, respectively. Finally, N is the total average coordination number of the simulated structure.

In addition, a related order parameter, denoted as a_x (where $x = \text{Ge, Sb, or Te}$), can be obtained to evaluate the difference in quasibinary chemical order within an amorphous structure from the perspective of each atomic species. This species-resolved parameter has been used in previous simulation studies to describe chemical order in glassy $\text{Ge}_2\text{Sb}_2\text{Te}_5$ models,^[40,43] and is given by

$$a_x = \begin{cases} 1 - \frac{N_{xB}}{x_A x_B (N_A + N_B)}, & \text{if } x = \text{Ge, Sb} \\ 1 - \frac{N_x}{x_A x_B (N_A + N_B)}, & \\ 1 - \frac{N_{xA}}{x_A x_B (N_A + N_B)}, & \text{if } x = \text{Te} \\ 1 - \frac{N_x}{x_A x_B (N_A + N_B)}, & \end{cases} \quad (9)$$

We note that the two approaches share a common interpretation, namely that a value of 1 signifies a fully (chemically) ordered structure, such as the ideal cubic-crystal phase, a value of 0 denotes a random arrangement, and a value of -1 points toward a tendency for clustering, or phase separation into A-rich and B-rich regions.^[40]

The results of the analysis for the 200 eV primary knock-on atom event, along the entire molecular dynamics trajectory, are shown in **Figure 5**. The initial $\text{Ge}_2\text{Sb}_2\text{Te}_5$ structure, generated with a melt-quench simulation, has a chemical-order parameter of 0.7. It can be observed that, on the timescale of 1–2 ps after the beginning of the simulation, the radiation-damage cascade causes a significant reduction in the chemical order of the simulated structure (by as much as 50%), highlighting the impact of the nonequilibrium cascade event on the amorphous network. The drop in the chemical-order parameter to its minimum value is indicative of the destruction and formation of bonds inside the glass model. Thereafter, and during the thermal quench of the cascade event, the chemical order of the simulated structure partially recovers toward its initial value, resulting in a semiordered configuration. The degree of disorder in the final equilibrated amorphous structure, at the end of the molecular dynamics simulation, is

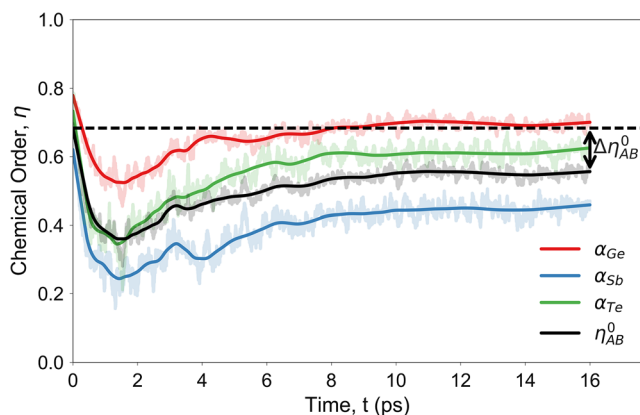


Figure 5. Time evolution of the chemical order within the glass network of the amorphous $\text{Ge}_2\text{Sb}_2\text{Te}_5$ model for the 200 eV primary knock-on Te atom simulation. The calculated Cargill–Spaepen order parameter is shown in black, while traces corresponding to the chemical-order parameters of the different atomic species are shown in red for Ge, in blue for Sb, and in green for Te atoms. The raw data for the order parameters are shown with a slight transparency, while the solid lines represent locally smoothed traces. The horizontal dashed line shows the chemical-order parameter of the pristine (melt-quenched) amorphous structure. The black double-headed arrow indicates the difference in chemical order of the glass model, $\Delta\eta_{AB}^0$, before and after the irradiation simulation.

increased for the whole model and per atomic species, compared to the initial glass model, before the irradiation event.

The overall picture of the temporal evolution of the chemical order inside the $\text{Ge}_2\text{Sb}_2\text{Te}_5$ model indicates the partial local melting of the glass structure due to the radiation-damage cascade, as was demonstrated in the energy landscape of the 200 eV primary knock-on atom simulation, shown in Figure 2c. Moreover, it verifies the recovery of the short- and medium-range order of the amorphous network in the final equilibrated structure at 300 K, while suggesting an enhanced amount of disorder after the irradiation event.

The amount of disorder that is generated in the irradiated amorphous structure can be estimated from the difference in chemical-order parameters, $\Delta\eta_{AB}^0$, between the initial, melt-quenched, glassy structure and the final equilibrated configuration, at the end of the molecular dynamics simulation, relative to the chemical order of the pristine model, denoted as $\eta(0)$. The degree of disorder is plotted against the different initial primary knock-on atom energies, ranging from 0 to 200 eV, and the results are shown in Figure 6. We note that the calculated values of this disorder measure have been averaged over the last 2 ps of each molecular dynamics trajectory at 300 K. It can be seen that the degree of disorder in the irradiated amorphous structure increases as the energy of the primary knock-on atom increases. An analysis of the species-resolved order parameters shows that the Sb atoms experience the largest increase in disorder, indicating specific structural rearrangements associated with the local atomic environments of Sb atoms that occurred within the glass network due to the impact of the cascade-damage event.

In order to further investigate the induced chemical disorder in the amorphous $\text{Ge}_2\text{Sb}_2\text{Te}_5$ models after exposure to the primary knock-on atom events, the coordination number of each

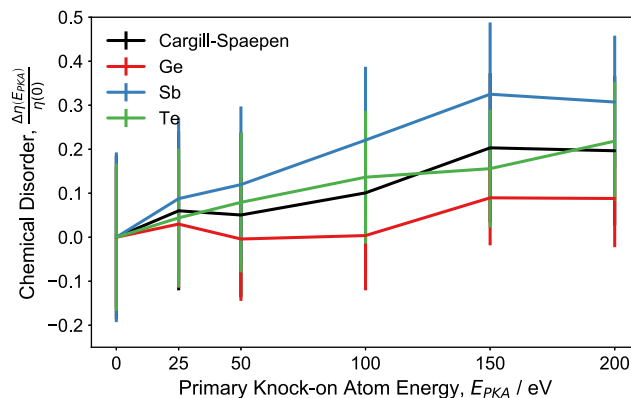


Figure 6. Estimation of chemical disorder in the amorphous network of the $\text{Ge}_2\text{Sb}_2\text{Te}_5$ model versus the kinetic energy of the primary knock-on atom for the different irradiation simulations. The Cargill–Spaepen order parameter is shown in black, while traces corresponding to the chemical-order parameters of different atom types are shown in red for Ge, in blue for Sb, and in green for Te. The calculated disorder indicators are averaged during a molecular dynamics run with the (NVE) microcanonical ensemble for the final irradiated configurations, for each case, while error bars reflect the standard deviation of the chemical-order parameters during the sampling.

atom i in the simulated structure has been calculated to construct simple order parameters, based on the different local atomic environments inside the glassy network, given by the following terms:

$$a_i = \begin{cases} \frac{N_i^B - N_i^A}{N_i}, & i = \text{Ge, Sb} \\ \frac{N_i^A - N_i^B}{N_i}, & i = \text{Te} \end{cases} \quad (10)$$

where N_i^B and N_i^A correspond to the number of B (Te) and A (Ge, Sb) neighbors of the respective atom i . In this analysis, bond formation between two nearby atoms is deemed to occur if the interatomic distance between them is shorter than or equal to the cut-off given to describe the maximum distance for nearest-neighbor atoms between the relative species within their first coordination shell (i.e., local atomic coordination).^[44] Here, a uniform geometric bond cut-off distance of 3.2 Å was applied for all pairs of atomic species, similar to previous modeling studies of amorphous $\text{Ge}_2\text{Sb}_2\text{Te}_5$.^[28,32,39,43,45–47]

The modifications in the local order within the amorphous network are evaluated by calculating the local-order parameters, for all the atoms, in the initial, melt-quenched, glassy structure, and in the final irradiated models after the 100 and 200 eV primary knock-on atom simulations, and the results are shown in Figure 7. When the surrounding atoms in the vicinity of a given atom i are in an ordered arrangement, the local-order parameter, a_i , is close to 1. In contrast, when the local environments of the atoms in the neighborhood of an atom i follow a random arrangement, then a_i is closer to 0, while if these atoms are locally clustered, the value of a_i is close to -1 . It can be seen that the ordered $A - B$ alternation that is apparent in the glass model before irradiation, visible in yellow, is disrupted by random and disordered

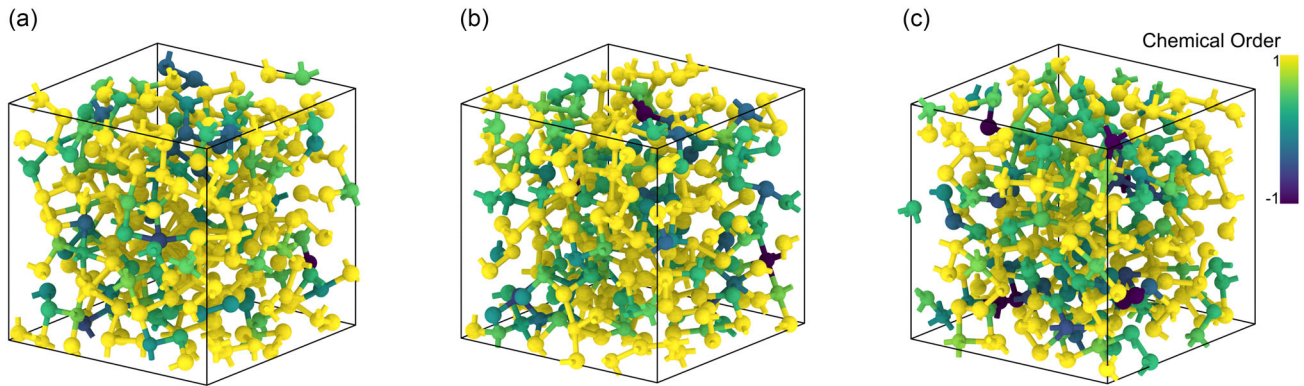


Figure 7. Local quasibinary chemical order, a_i , ranging from 1 (yellow, fully ordered) to -1 (indigo, clustered), with values close to 0 denoting a random arrangement, in: a) the pristine amorphous $\text{Ge}_2\text{Sb}_2\text{Te}_5$ model; b) the optimized irradiated glassy structure after the 100 eV primary knock-on atom simulation; and c) the optimized irradiated glassy structure after the 200 eV primary knock-on atom simulation.

atoms in the geometry of the irradiated model structures, that are shown in green, blue, and indigo. These configurations are typically correlated to local atomic environments with homopolar, $A - A$ and $B - B$, bonds. This observation is in accordance with the increased proportion of Te-Te, Ge-Sb, and Sb-Sb homopolar bonds that was found in the simulated $\text{Ge}_2\text{Sb}_2\text{Te}_5$ structure after the 200 eV primary knock-on atom event.^[14]

4. Vibrational Properties

In order to gain an insight into the short-time dynamics of the irradiated models, information extracted from the velocity autocorrelation function, at low temperature, can be used to investigate the vibrational properties of the simulated glass structures and to compare them to experimental measurements.^[48,49] The normalized velocity autocorrelation can be calculated by averaging over the total number of atoms, N , in the modeled system,^[50] defined as

$$C_{vv}(t) = \frac{1}{N} \sum_{i=1}^N \frac{\langle \mathbf{v}_i(t) \cdot \mathbf{v}_i(0) \rangle}{\langle \mathbf{v}_i(0) \cdot \mathbf{v}_i(0) \rangle} \quad (11)$$

where $\mathbf{v}_i(0)$ is the velocity of an atom i at time $t = 0$, and $\mathbf{v}_i(t)$ is the velocity of the same atom at some later time, t . It is noted that the angle brackets in the equation above correspond to a time average of the relevant quantities.

The Fourier transform of the velocity-autocorrelation results in the power spectrum of the glass system, which, given the fact that $C_{vv}(t)$ is an even function, can be calculated as

$$C(\omega) = \frac{2}{3Nk_B T} \int_0^\infty C_{vv}(t) \cos(\omega t) dt \quad (12)$$

where ω is the frequency, k_B is the Boltzmann constant, and T is the temperature of the molecular dynamics simulation. In practice, only a finite trajectory can be utilized (of certain time-length, t_{max}) and, usually, a damping or filtering function is applied to smooth out any potential artefacts due to the finite-time truncation.

Molecular dynamics trajectories at 300 K, sampled under the canonical (NVT) thermodynamical ensemble, were used for the estimation of the vibrational power spectrum of the amorphous $\text{Ge}_2\text{Sb}_2\text{Te}_5$ models after the radiation-damage simulations, as well as for the initial glassy model before the primary knock-on atom events, and the results are shown in **Figure 8**. Due to the relatively short molecular dynamics runs used for the calculations of the vibrational density of states, the patterns related to the observations for the peaks are treated from a qualitative perspective. It can be seen that the vibrational power spectrum of the glass models is centered at ≈ 1.5 THz (at 0 K the power spectrum obtained from lattice dynamics is centered at ≈ 3 THz as in ref. [45]), which is indicative of the anharmonicity of the simulated structures.^[15,51] In addition, it can be seen that the intensity of the vibrational density of states around the frequency regions of 1.5, 2.0, and 2.75 THz is enhanced as the primary knock-on atom energy increases, indicating a subtle irradiation-induced effect on the vibrational properties. This is in qualitative agreement (not in a similar frequency range) with the experimentally reported observations related to the Raman peak in the

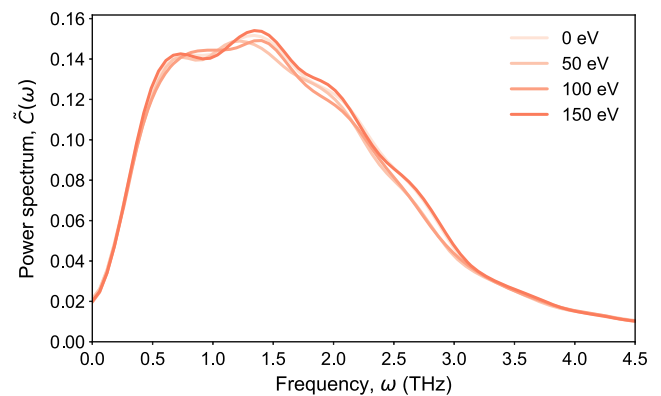


Figure 8. Vibrational density of states calculated from the Fourier transform of the velocity autocorrelation function for amorphous $\text{Ge}_2\text{Sb}_2\text{Te}_5$ models obtained from different irradiation simulations, corresponding to 0, 50, 100, and 150 eV primary knock-on atom energies, shown as different shades of orange.

high-frequency part of the spectrum in $\text{Ge}_2\text{Sb}_2\text{Te}_5$ films irradiated with Sb^+ ions.^[52]

It is worth mentioning that, in contrast to the lattice dynamics, the approach that was followed here for the estimation of the vibrational density of states of the amorphous models does not provide a direct access to the normal modes, in order to evaluate their degree of localization with respect to the different primary knock-on atom simulations; however, its advantage is related to the fact that the calculated power spectrum can reflect the full anharmonicity of the potential energy surface of the simulated structures. Nevertheless, the observed subtle effects in the vibrational density of states of the irradiated models, shown in Figure 8, may vary significantly when an analysis of longer trajectories is utilized for the calculations.

5. Conclusion

In this study, irradiation of amorphous $\text{Ge}_2\text{Sb}_2\text{Te}_5$ phase-change memory material reveals the impact of exploring liquid-like states in the energy landscape of the glassy structure during radiation-damage simulations, expanding our underlying knowledge associated with the response of this material to such nonequilibrium processes. Moreover, our previous structural analysis of the primary knock-on atom events is extended here to include various pieces of information related to the time evolution and the final degree of disorder in the irradiated $\text{Ge}_2\text{Sb}_2\text{Te}_5$ glass models.

The simulated system is able to access liquid-like states in the irradiation simulation, starting with the 200 eV primary knock-on atom energy, as revealed from a configurational map constructed from suitable structural descriptors. Consequently, from these high-energy states, the irradiated configuration can relax, due to lower energy barriers for structural reorganization. This provides a thermodynamic insight into the radiation tolerance of the amorphous material, which, combined with the pronounced flexibility (softness) of the chemical bonding and the available free volume within the glassy $\text{Ge}_2\text{Sb}_2\text{Te}_5$ structure, accounts for the essential attributes leading to the fast recovery of this material after irradiation.

From the perspective of the structural disorder, the most important effect was found to be an increase in the chemical disorder within the amorphous network of the $\text{Ge}_2\text{Sb}_2\text{Te}_5$ models. This is in agreement with previous work that showed a systematic increase in the amount of homopolar bonds and the proportion of edge-sharing tetrahedra in the simulated structure with respect to the primary knock-on atom energy.^[13,14] The general increase in the degree of chemical and topological disorder that was measured after the irradiation events can be correlated with the decrease in lattice thermal conductivity previously reported for the irradiated glass models, as such disordered atomic environments may scatter the heat-carrying acoustic modes.^[15] In addition, it is worth mentioning that similar structural motifs have also been associated with fragility and anharmonicity in models of other chalcogenide glasses.^[53]

Regarding the short-time dynamics of the simulated structures, the qualitative picture that emerges is that the irradiation-induced chemical disorder in the amorphous models leads to an increased vibrational density of states in the high-frequency region of the power spectrum. Alterations in this frequency

region in the Raman spectra of glassy $\text{Ge}_2\text{Sb}_2\text{Te}_5$ have been attributed to the induced formation of homopolar bonds and to an increased proportion of GeTe_4 edge-sharing tetrahedra in the irradiated amorphous structures.^[52] Also, we note that, in the pristine glass structure, spatial localization of the high-frequency normal modes (at 0 K) has been correlated with tetrahedral configurations centered on Ge atoms.^[45]

In follow-up studies, machine-learned molecular dynamics simulations using one of the available interatomic potentials developed for $\text{Ge}_2\text{Sb}_2\text{Te}_5$,^[54–56] should be utilized to perform large-scale radiation-damage simulations with primary knock-on atom energies in the range of keV (closer to the experimental values). Furthermore, this will allow the systematic investigation of finite-size effects and equilibration time-scales, which could significantly influence calculations of the vibrational density of states. Ultimately, this can provide a stronger connection between atomic disorder and the reduction in thermal conductivity in the irradiated amorphous $\text{Ge}_2\text{Sb}_2\text{Te}_5$ structures. Moreover, it remains to be seen whether the calculated lower thermal conductivity of the irradiated amorphous structures might be a result of the localization of vibrational modes^[57] at specific defective structural environments in glassy $\text{Ge}_2\text{Sb}_2\text{Te}_5$.

In conclusion, the current study is another step forward in an effort to build a comprehensive theoretical perspective on the effects of ion irradiation of phase-change memory materials. This may facilitate their design for radiation-hard integration in environments where exposure to radiation is a factor, such as in space exploration, where electronic devices, including but not limited to, microcontrollers and solid-state data recorders play an important role.

6. Experimental Section

Computer Models: The simulated systems are 315-atom models of the glassy $\text{Ge}_2\text{Sb}_2\text{Te}_5$ phase-change memory material. The irradiated structures are the final thermalized configurations obtained from nonequilibrium radiation-damage simulations with Te primary knock-on atom (PKA) energies of 25, 50, 100, 150, and 200 eV. Details related to the computational setup, the radiation-damage cascades, and the ab initio molecular dynamics simulations with the stochastic boundary-conditions approach can be found in our previous work.^[13,14]

Electronic-Structure Calculations: DFT, as implemented in the CP2K code,^[58,59] was used to optimize the geometries of the irradiated glassy $\text{Ge}_2\text{Sb}_2\text{Te}_5$ structures, using the Perdew–Burke–Ernzerhof (PBE) exchange–correlation functionals.^[60] The CP2K code employs a mixed Gaussian basis set with an auxiliary plane-wave basis set to represent the electrons in the modeled system.^[61] A double- ζ valence-polarized Gaussian basis set was used for all atomic species (Ge, Sb, and Te),^[62] in conjunction with the Goedecker–Teter–Hutter pseudopotential.^[63] The plane-wave energy cutoff was set to 5440 eV (400 Ry). The limited-memory Broyden–Fletcher–Goldfarb–Shanno algorithm^[64] was applied in the geometry optimizations of the amorphous structures to minimize the total energy of the modeled systems with respect to the atomic coordinates. The convergence criterion for the forces on individual atoms of the current configuration in an iteration step was set to 10^{-3} eV \AA^{-1} ($\approx 2.0 \times 10^{-5}$ Ha Bohr⁻¹). Periodic-boundary conditions were enforced in all calculations.

Molecular Dynamics Simulations: Born–Oppenheimer ab initio molecular dynamics simulations were carried out using the CP2K code.^[58,59] The electronic structure was treated through the Kohn–Sham formulation of DFT using the generalized-gradient approximation with the PBE exchange–correlation functionals.^[60] The canonical ensemble (constant

number of particles, volume, and temperature or NVT) was applied and the generalized Langevin equation, or colored-noise, thermostat^[65] was chosen to control the temperature fluctuations. At 300 K molecular dynamics trajectories of 10 ps were generated with a timestep of 1 fs for the pristine and each irradiated glassy models. The velocity autocorrelation functions were calculated from the 10000 configurations collected during these molecular dynamics runs.

SOAP: The SOAP descriptor is a mathematical approach employed to smoothly and efficiently encode the local density of atomic environments, in a form readily amenable for machine-learning algorithms, for problems related to classification or regression.^[20] To describe the environment of an atom i , an atomic density $\rho_i(\mathbf{r})$, is constructed by placing Gaussian functions, of broadness σ , on the atomic positions, up to a cutoff distance, r_{cut} . In the construction of the SOAP descriptor, the local atomic density is expanded radially with an orthonormal basis $R_n(r)$, and angularly into spherical harmonics Y_{lm} , as follows:

$$\rho_i(\mathbf{r}) = \sum_{n=0}^{n_{\text{max}}} \sum_{l=0}^{l_{\text{max}}} \sum_{m=-l}^l c_{nlm}^i R_n(r) Y_{lm}(\theta, \phi) \quad (13)$$

The obtained representation is invariant under permutation of equivalent atoms, and under translations and rotations of the neighbor environment.^[20] The coefficients (c_{nlm}^i) of the expansion of the atomic-neighbor density are used to form the power spectrum, given by

$$p_{nm'l}^j = \sqrt{\frac{8\pi^2}{2l+1}} \sum_m (c_{nlm}^j)^* c_{n'l'm}^j \quad (14)$$

The elements of a finite truncation of the power spectrum (up to a given n_{max} and l_{max}) are utilized as components of the many-body atomic-descriptor vector, \mathbf{q}_i , employed in this study. We note that this vector is typically normalized to have unit length.

Then, a bounded measure of the similarity between the atomic environments of an atom i and of an atom j is provided by the SOAP kernel, as a simple dot product of the corresponding SOAP vectors:

$$k^{\text{SOAP}}(i,j) = \sum_{nm'l} p_{nm'l}^i p_{nm'l}^j = \mathbf{q}_i \cdot \mathbf{q}_j \quad (15)$$

Metric MDS: MDS is a technique used to represent high-dimensional data in a lower-dimensional space to aid visualization. The goal of MDS is to map a finite set of data points onto a reduced space, in a way that the relative distances are maintained as faithfully as possible in the new (low-dimensional) space.^[22] This process involves embedding the points such that their new coordinates reflect their original distances. Consequently, the embedding vectors (\mathbf{e}) in the context of MDS correspond to these new coordinates of the data points after dimensionality reduction. MDS follows an optimization process to achieve the optimal embedding, by minimizing a stress (objective) function, which measures the discrepancy between the original (high-dimensional) and the reduced-space (embedded) distances.^[22] The Scaling by MAjorizing a COmplicated Function algorithm was used for this purpose, employing a majorization technique that guarantees a monotonic convergence of the stress function. The algorithm starts with a random distribution of points and iteratively adjusts their positions until the stress function is minimized. The resulting embedding vectors can be visualized, in this study, as a 2D map, where distances reflect the true relationships between data points. This allows for an intuitive understanding of similarities and dissimilarities since similar configurations remain close together, whereas dissimilar structures stay far apart. Overall, MDS captures the intrinsic structure of the data by preserving significant variations in the reduced space.^[22]

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Conflict of Interest

The authors declare no conflict of interest.

Author Contributions

Felix C. Mocanu: conceptualization (equal); data curation (lead); formal analysis (lead); investigation (equal); methodology (lead); visualization (lead); writing—review and editing (equal). **Stephen R. Elliott:** investigation (supporting); supervision (equal); writing—review and editing (equal). **Konstantinos Konstantinou:** conceptualization (equal); formal analysis (supporting); funding acquisition (lead); investigation (equal); methodology (supporting); supervision (equal); writing—original draft (lead).

Data Availability Statement

The data that support the findings of this study are available from the corresponding author upon reasonable request.

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amorphous materials, chemical disorder, Ge–Sb–Te, ion irradiation, liquid states, phase-change memory

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- [1] E. D. Zanotto, J. C. Mauro, *J. Non-Cryst. Solids* **2017**, 471, 490.
- [2] P. K. Gupta, W. Kob, *J. Non-Cryst. Solids X* **2019**, 3, 100031.
- [3] K. Vollmayr, W. Kob, K. Binder, *Phys. Rev. B* **1996**, 54, 15808.
- [4] S. M. S. Privitera, E. Rimini, *Mater. Sci. Semicond. Process.* **2021**, 135, 106087.
- [5] C. D. Wright, P. Hosseini, J. A. Vazquez Diosdado, *Adv. Funct. Mater.* **2013**, 23, 2248.
- [6] T. Tuma, A. Pantazi, M. Le Gallo, A. Sebastian, E. Eleftheriou, *Nat. Nanotech.* **2016**, 11, 693.
- [7] I. Boybat, M. Le Gallo, S. R. Nandakumar, T. Moraitis, T. Parnell, T. Tuma, B. Rajendran, Y. Leblebici, A. Sebastian, E. Eleftheriou, *Nat. Commun.* **2018**, 9, 2514.
- [8] R. F. Freitas, W. W. Wilcke, *IBM J. Res. Dev.* **2008**, 52, 439.
- [9] S. R. Elliott, *Int. J. Appl. Glass Sci.* **2015**, 6, 15.
- [10] R. Bez, A. Pirovano, *Mater. Sci. Semicond. Process.* **2004**, 7, 349.
- [11] S. Raoux, W. Wetnic, D. Ielmini, *Chem. Rev.* **2010**, 110, 240.
- [12] S. R. Ovshinsky, E. J. Evans, D. L. Nelson, H. Fritzsche, *IEEE Trans. Nucl. Sci.* **1968**, 15, 311.
- [13] K. Konstantinou, T. H. Lee, F. C. Mocanu, S. R. Elliott, *Proc. Natl. Acad. Sci. USA* **2018**, 115, 5353.
- [14] K. Konstantinou, F. C. Mocanu, T. H. Lee, S. R. Elliott, *J. Phys.: Condens. Matter* **2018**, 30, 455401.
- [15] F. C. Mocanu, K. Konstantinou, S. R. Elliott, *Appl. Phys. Lett.* **2020**, 116, 031902.
- [16] N. M. A. Krishnan, B. Wang, Y. Yu, Y. Le Pape, G. Sant, M. Bauchy, *Phys. Rev. X* **2017**, 7, 031019.

- [17] N. M. A. Krishnan, B. Wang, Y. Le Pape, G. Sant, M. Bauchy, *J. Chem. Phys.* **2017**, *146*, 204502.
- [18] S. De, A. P. Bartók, G. Csányi, M. Ceriotti, *Phys. Chem. Chem. Phys.* **2016**, *18*, 13754.
- [19] B. Cheng, R.-R. Griffiths, S. Wengert, C. Kunkel, T. Stenczel, B. Zhu, V. L. Deringer, N. Bernstein, J. T. Margraf, K. Reuter, G. Csányi, *Acc. Chem. Res.* **2020**, *53*, 1981.
- [20] A. P. Bartók, R. Kondor, G. Csányi, *Phys. Rev. B* **2013**, *87*, 219902.
- [21] J. B. Kruskal, *Psychometrika* **1964**, *29*, 1.
- [22] I. Borg, P. J. F. Groenen, *Modern Multidimensional Scaling: Theory and Applications*, *Springer Series in Statistics*, 2 ed., Springer-Verlag, New York **2005**.
- [23] K. Konstantinou, F. C. Mocanu, J. Akola, *Phys. Rev. B* **2022**, *106*, 184103.
- [24] T. C. Nicholas, A. L. Goodwin, V. L. Deringer, *Chem. Sci.* **2020**, *11*, 12580.
- [25] F. Pedregosa, G. Varoquaux, A. Gramfort, V. Michel, B. Thirion, O. Grisel, M. Blondel, P. Prettenhofer, R. Weiss, V. Dubourg, J. Vanderplas, A. Passos, D. Cournapeau, M. Brucher, M. Perrot, E. Duchesnay, *J. Mach. Learn. Res.* **2011**, *12*, 2825.
- [26] N. Yamada, T. Matsunaga, *J. Appl. Phys.* **2000**, *88*, 7020.
- [27] T. Matsunaga, N. Yamada, *Phys. Rev. B* **2004**, *69*, 104111.
- [28] K. Konstantinou, F. C. Mocanu, T. H. Lee, S. R. Elliott, *Nat. Commun.* **2019**, *10*, 3065.
- [29] F. H. Stillinger, *Science* **1995**, *267*, 1935.
- [30] F. Sciortino, *J. Stat. Mech. Theor. Exp.* **2005**, *2005*, P05015.
- [31] S. Sastry, P. G. Debenedetti, F. H. Stillinger, *Nature* **1998**, *393*, 554.
- [32] T. H. Lee, S. R. Elliott, *Adv. Mater.* **2017**, *29*, 1700814.
- [33] R. O. Jones, *J. Phys.: Condens. Matter* **2018**, *30*, 153001.
- [34] A. Baranyai, D. J. Evans, *Phys. Rev. A* **1989**, *40*, 3817.
- [35] D. S. Franzblau, *Phys. Rev. B* **1991**, *44*, 4925.
- [36] C. E. Shannon, *Bell Syst. Techn. J.* **1948**, *27*, 379.
- [37] S. Kullback, R. A. Leibler, *Ann. Math. Stat.* **1951**, *22*, 79.
- [38] G. S. Cargill, F. Spaepen, *J. Non-Cryst. Solids* **1981**, *43*, 91.
- [39] J. Akola, J. Larrucea, R. O. Jones, *Phys. Rev. B* **2011**, *83*, 094113.
- [40] F. C. Mocanu, K. Konstantinou, S. R. Elliott, *J. Phys. D: Appl. Phys.* **2020**, *53*, 244002.
- [41] J. Akola, R. O. Jones, *Phys. Rev. B* **2007**, *76*, 235201.
- [42] J. Kalikka, J. Akola, R. O. Jones, *Phys. Rev. B* **2016**, *94*, 134105.
- [43] J. Akola, R. O. Jones, *J. Phys.: Condens. Matter* **2008**, *20*, 465103.
- [44] K. Konstantinou, S. R. Elliott, J. Akola, *J. Mater. Chem. C* **2022**, *10*, 6744.
- [45] S. Caravati, M. Bernasconi, T. D. Kühne, M. Krack, M. Parrinello, *J. Phys.: Condens. Matter* **2009**, *21*, 255501.
- [46] K. Konstantinou, F. C. Mocanu, J. Akola, S. R. Elliott, *Acta Mater.* **2022**, *223*, 117465.
- [47] K. Konstantinou, S. R. Elliott, *Phys. Status Solidi RRL* **2023**, *17*, 2200496.
- [48] J. Du, Y. Xiang, *J. Non-Cryst. Solids* **2012**, *358*, 1059.
- [49] K. Konstantinou, P. V. Sushko, D. M. Duffy, *J. Non-Cryst. Solids* **2015**, *422*, 57.
- [50] A. Pedone, *J. Phys. Chem. C* **2009**, *113*, 20773.
- [51] P. Zalden, K. S. Siegert, S. Rols, H. E. Fischer, F. Schlich, T. Hu, M. Wuttig, *Chem. Mater.* **2014**, *26*, 2307.
- [52] R. De Bastiani, A. M. Piro, M. G. Grimaldi, E. Rimini, G. A. Baratta, G. Strazzulla, *Appl. Phys. Lett.* **2008**, *92*, 241925.
- [53] M. Wilson, P. S. Salmon, *Phys. Rev. Lett.* **2009**, *103*, 157801.
- [54] F. C. Mocanu, K. Konstantinou, T. H. Lee, N. Bernstein, V. L. Deringer, G. Csányi, S. R. Elliott, *J. Phys. Chem. B* **2018**, *122*, 8998.
- [55] Y. Zhou, W. Zhang, E. Ma, V. L. Deringer, *Nat. Electron.* **2023**, *6*, 746.
- [56] O. Abou El Kheir, L. Bonati, M. Parrinello, M. Bernasconi, *npj Comput. Mater.* **2024**, *10*, 33.
- [57] P. B. Allen, J. L. Feldman, *Phys. Rev. B* **1993**, *48*, 12581.
- [58] J. VandeVondele, M. Krack, F. Mohamed, M. Parrinello, T. Chassaing, J. Hutter, *Comp. Phys. Comm.* **2005**, *167*, 103.
- [59] T. D. Kühne, M. Iannuzzi, M. Del Ben, V. V. Rybkin, P. Seewald, F. Stein, T. Laino, R. Z. Khaliullin, O. Schütt, F. Schiffmann, D. Golze, J. Wilhelm, S. Chulkov, M. H. Bani-Hashemian, V. Weber, U. Borštnik, M. Taillefumier, A. S. Jakobovits, A. Lazzaro, H. Pabst, T. Müller, R. Schade, M. Guidon, S. Andermatt, N. Holmberg, G. K. Schenter, A. Hehn, A. Bussy, F. Belleflamme, G. Tabacchi, et al., *J. Chem. Phys.* **2020**, *152*, 194103.
- [60] J. P. Perdew, K. Burke, M. Ernzerhof, *Phys. Rev. Lett.* **1996**, *77*, 3865.
- [61] G. Lippert, J. Hutter, M. Parrinello, *Mol. Phys.* **1997**, *92*, 477.
- [62] J. VandeVondele, J. Hutter, *J. Chem. Phys.* **2007**, *127*, 114105.
- [63] S. Goedecker, M. Teter, J. Hutter, *Phys. Rev. B* **1996**, *54*, 1703.
- [64] R. H. Byrd, P. Lu, J. Nocedal, C. Zhu, *SIAM J. Sci. Comput.* **1995**, *16*, 1190.
- [65] M. Ceriotti, G. Bussi, M. Parrinello, *Phys. Rev. Lett.* **2009**, *102*, 020601.